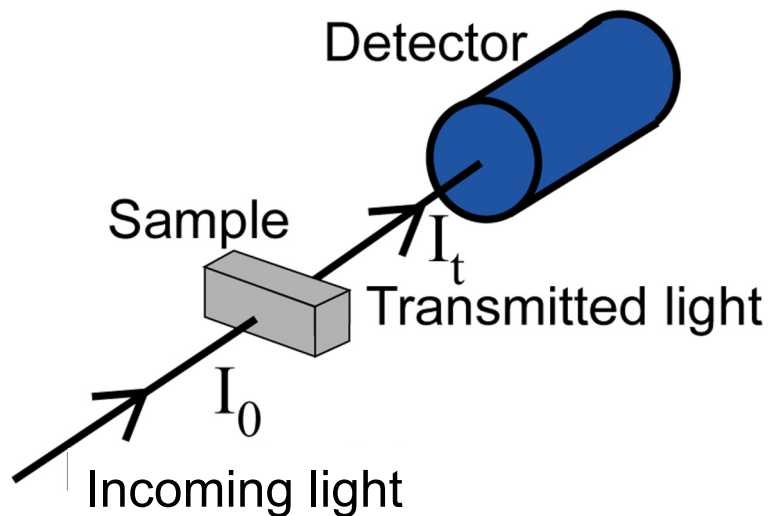


# ESC1: Circular Dichroism: best practice and data analysis

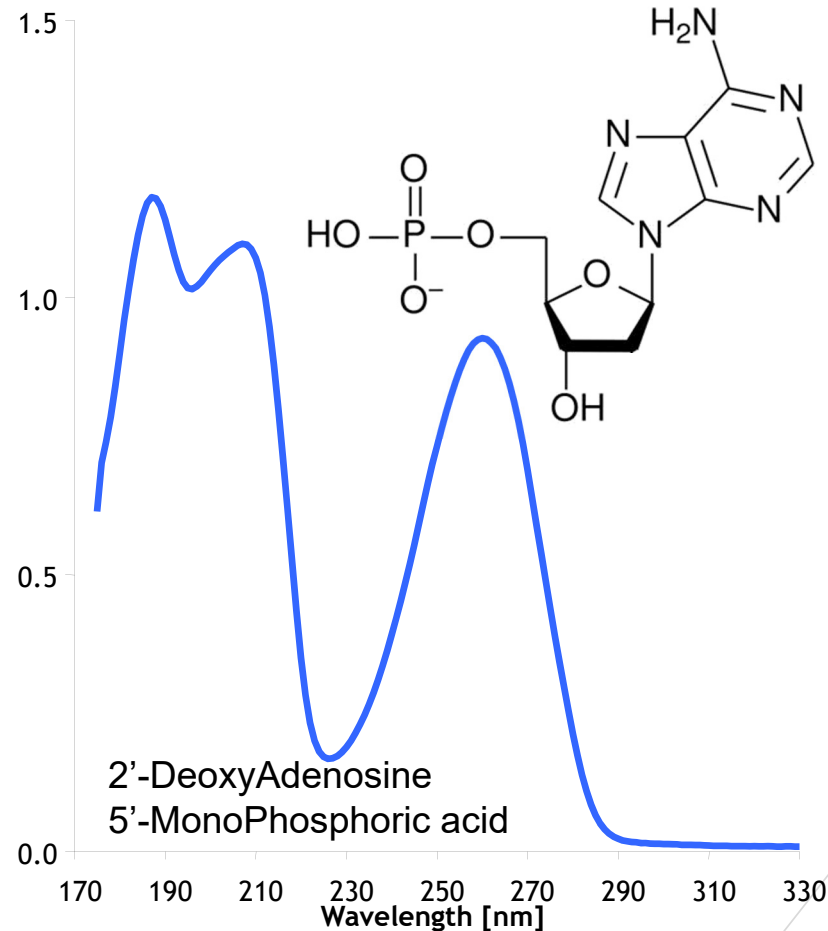
**Lecture 1:** Introduction to CD spectroscopy: Principles and information content

# Absorption Spectroscopy

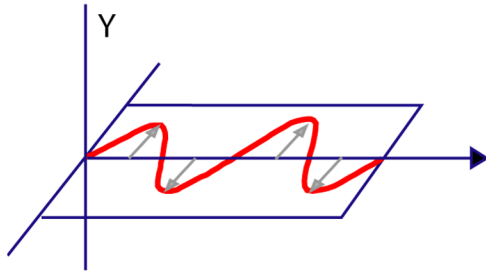


Absorbance:

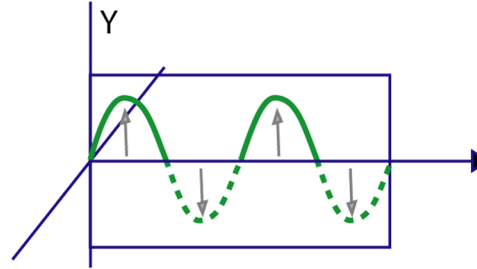
$$A = \log(I_0/I_t)$$



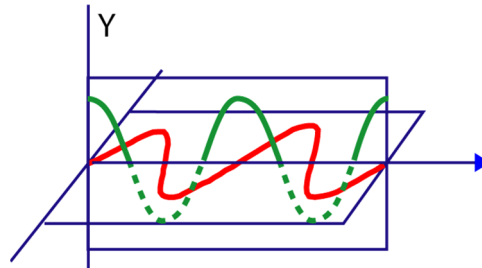
# Polarized light



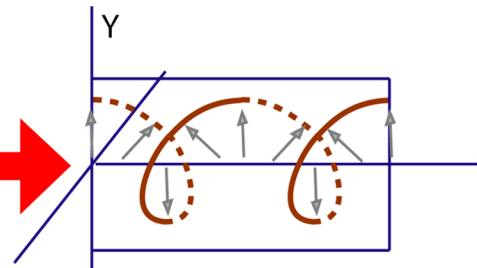
Horizontally pol.



Vertically pol.

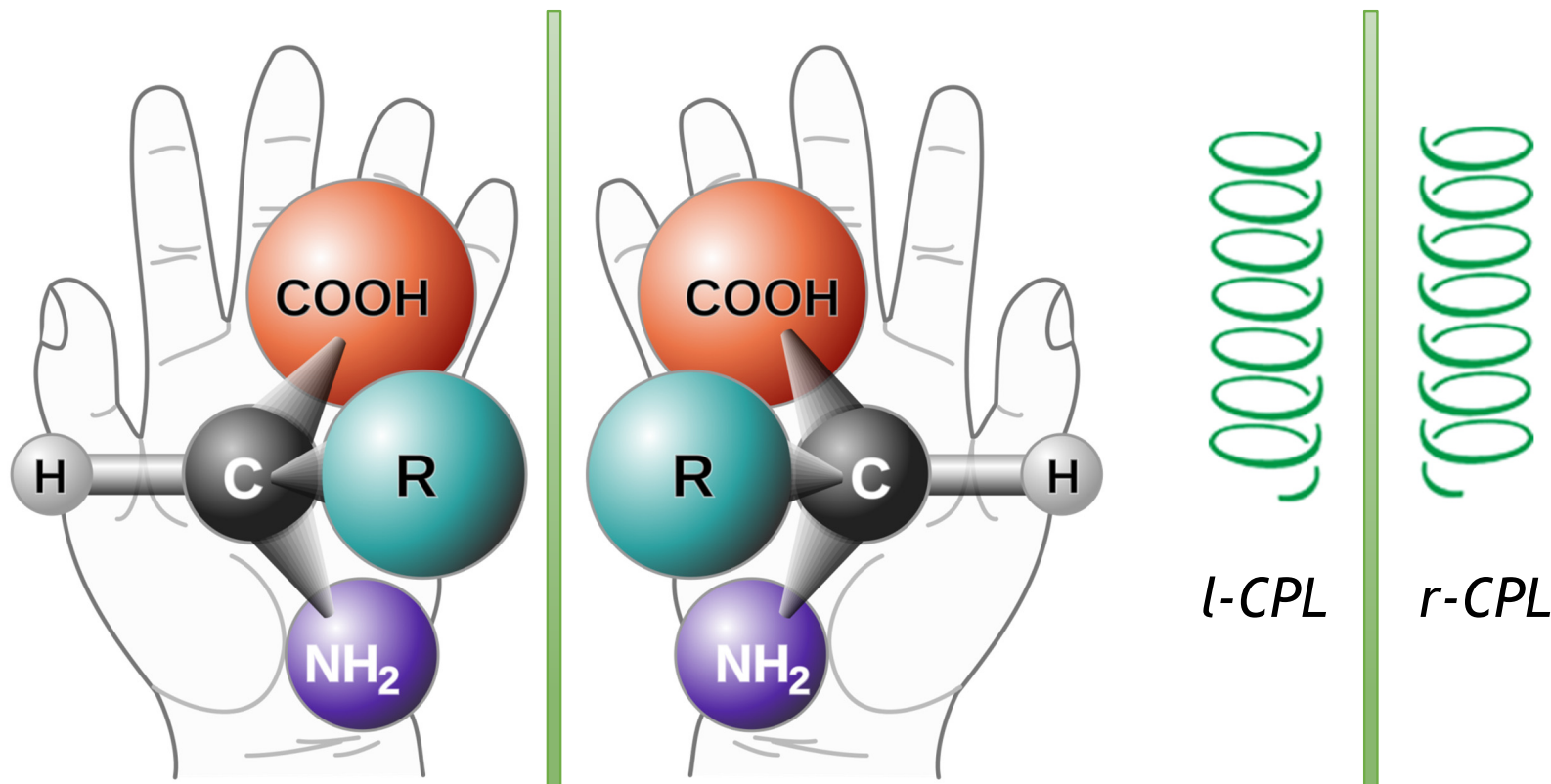


Sum of two  
plane pol.



Circularly pol.

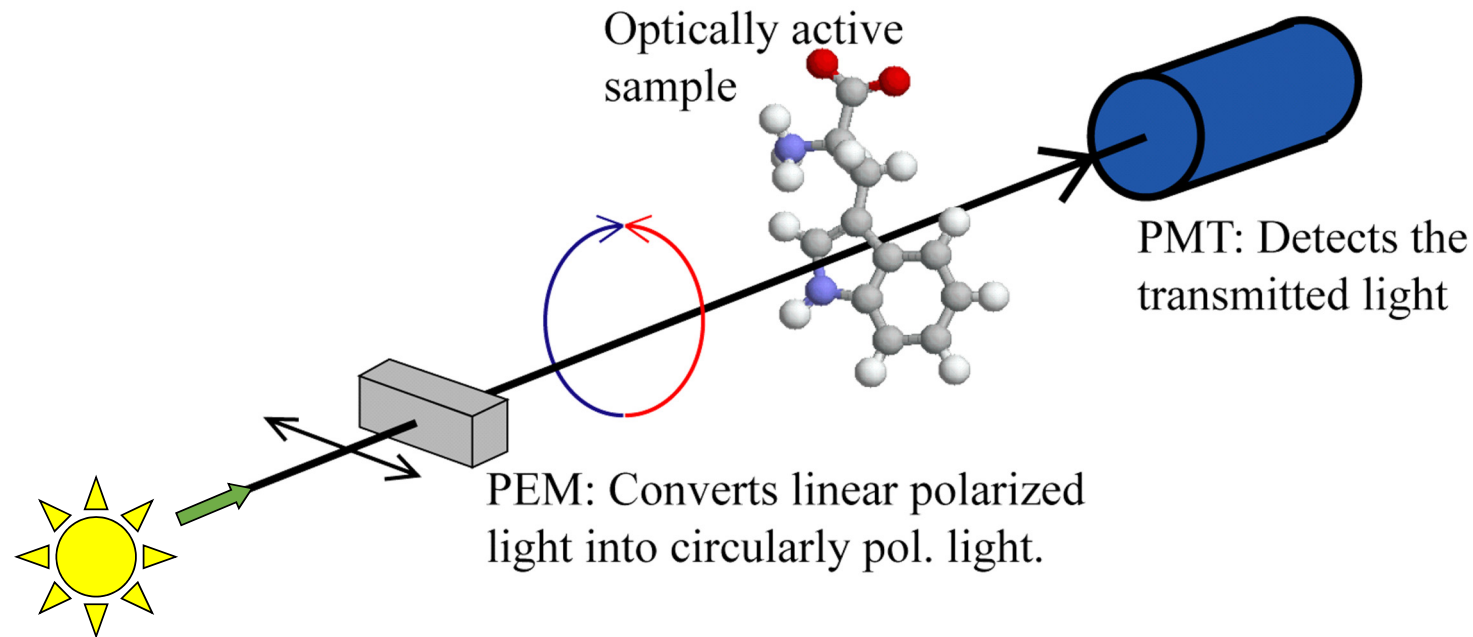
# Chirality



Mirror images



# Circular Dichroism

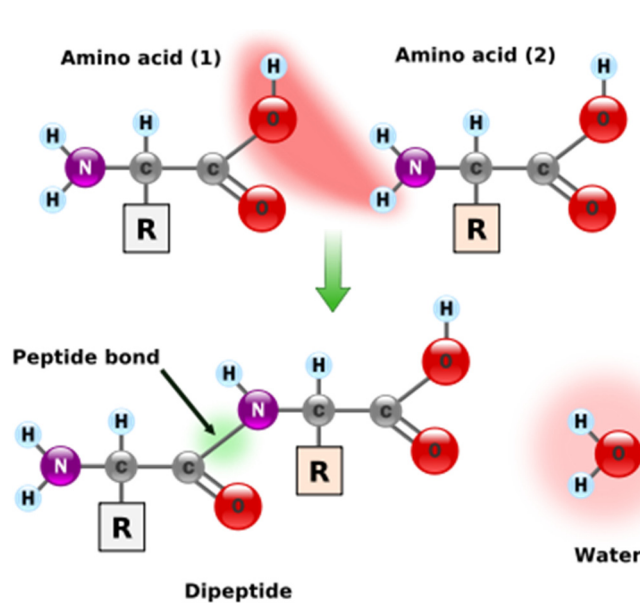


$$\text{The CD signal: } CD = A_L - A_R$$

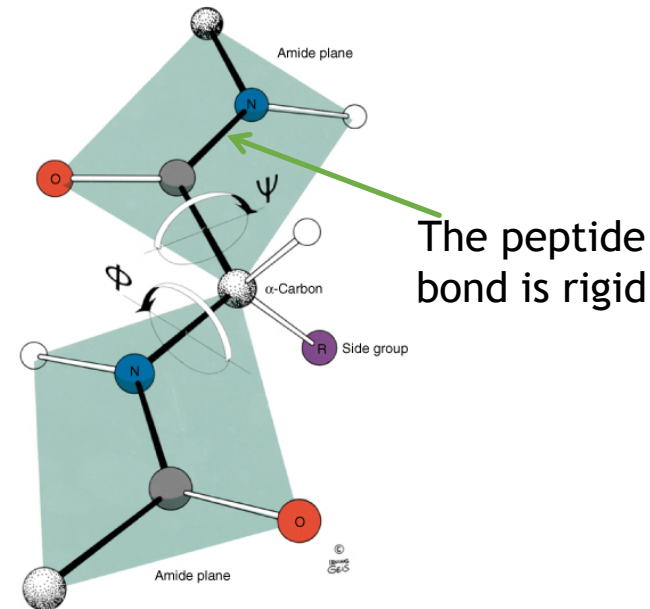
# Protein structure

## The peptide bond

Amino acids form polymers with a peptide bond between them



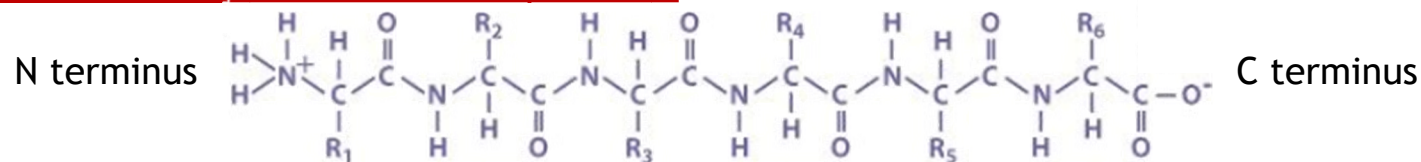
The peptide bond is formed between the carboxyl and the amine groups



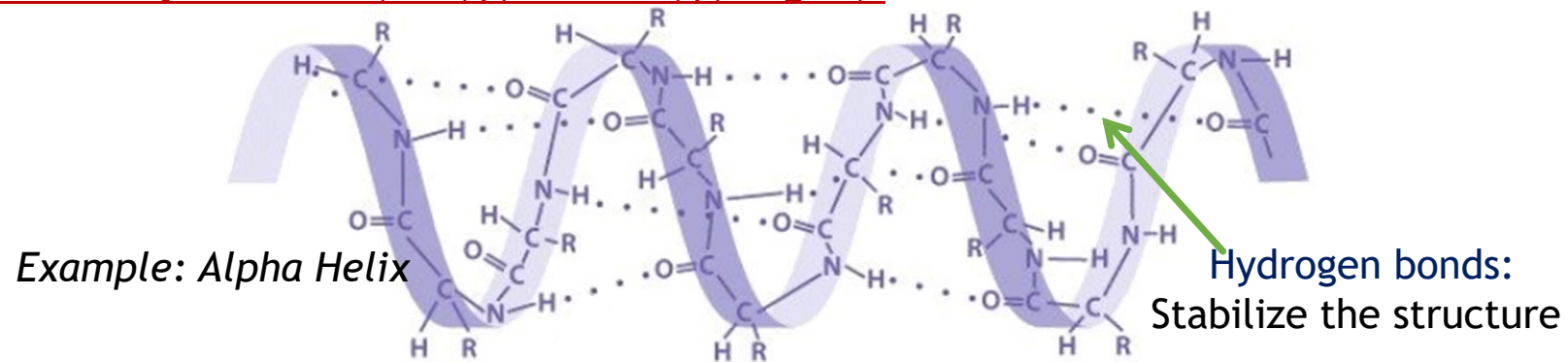
Phi ( $\varphi$ ) and Psi ( $\psi$ ) angles define the secondary structure

# Protein structure: Levels of structures

## Primary structure (amino acid sequence):



## Secondary structure (Phi ( $\varphi$ ) and Psi ( $\psi$ ) angles):



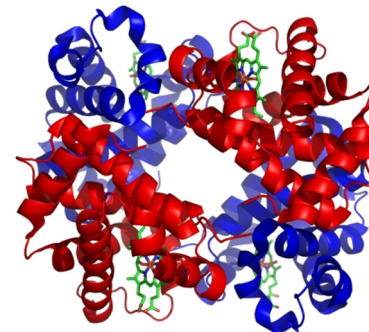
## Tertiary structure:

Relative organization  
of secondary  
structures



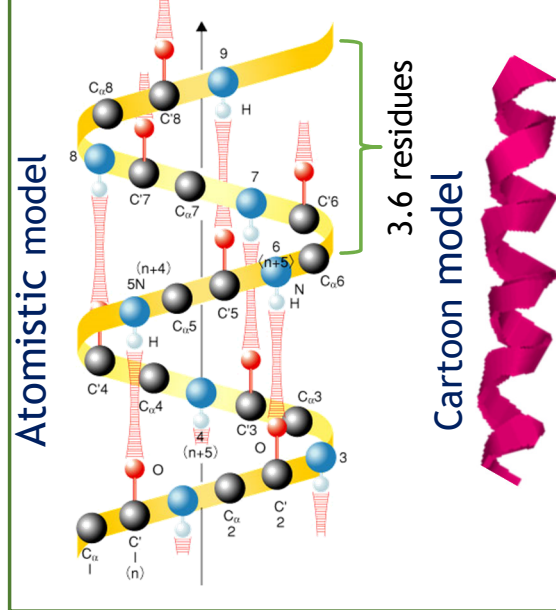
## Quaternary structure:

Relative organization  
of tertiary structures



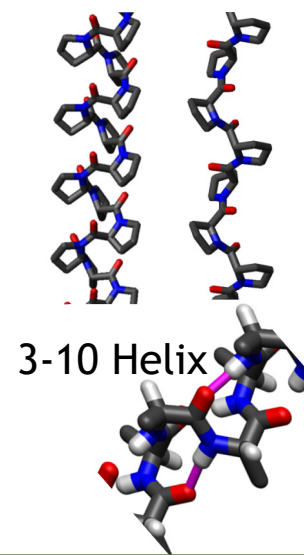
# Protein structure: Secondary structures

## Alpha Helix

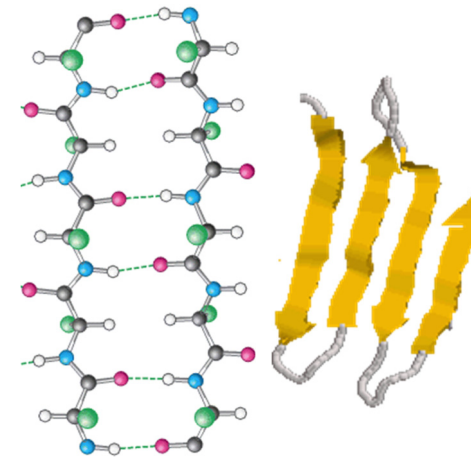


## Other Helices

PP-I PP-II

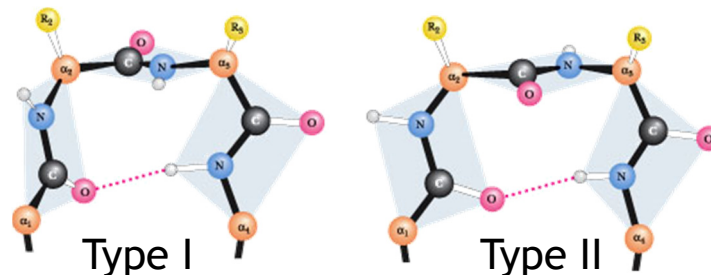


## Beta Sheet



## Beta turns

Atomistic model



Cartoon model



# Protein structure: Secondary structures

Circular dichroism can give secondary structure information  
from a large range of proteins/peptides

Very simple



Glucagon (Hormone)  
 $M_w \sim 3485$  Daltons  
29 amino acids



Pepsin (Enzyme)  
 $M_w \sim 35.000$  Daltons  
 $\sim 320$  amino acids

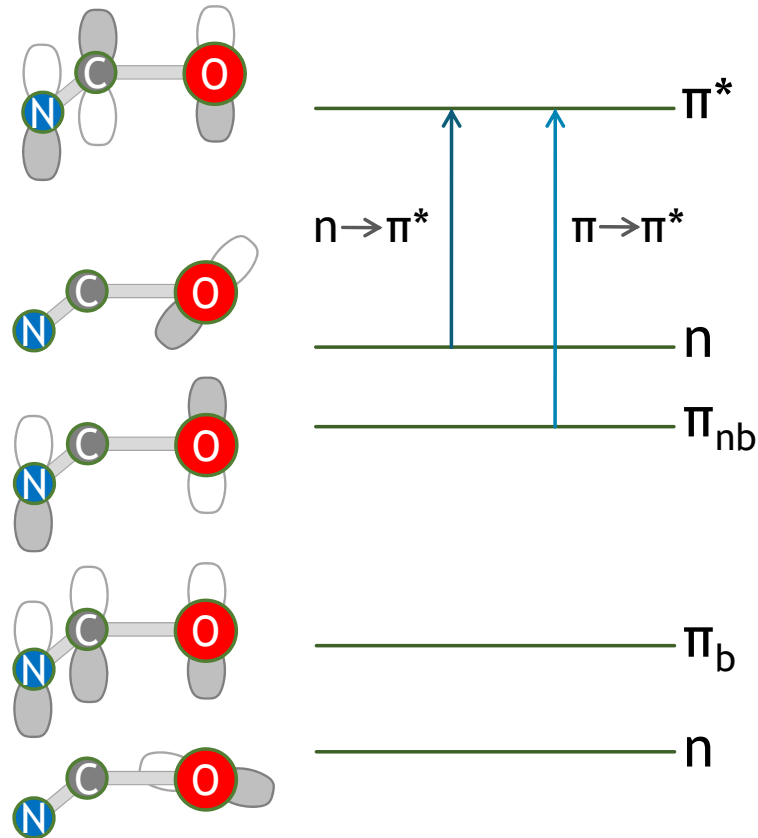
Very complicated



Antibody  
 $M_w \sim 150.000$  Daltons  
 $\sim 1.400$  amino acids

# Protein *electronic* structure

The backbone of the polypeptide has important chromophores



The  $n \rightarrow \pi^*$  and the  $\pi \rightarrow \pi^*$  have transitions in the *far UV spectral range*

$n \rightarrow \pi^* \sim 220 \text{ nm}$

$\pi \rightarrow \pi^* \sim 190 \text{ nm}$

These transitions are sensitive to the  $\varphi$  and  $\psi$  angles

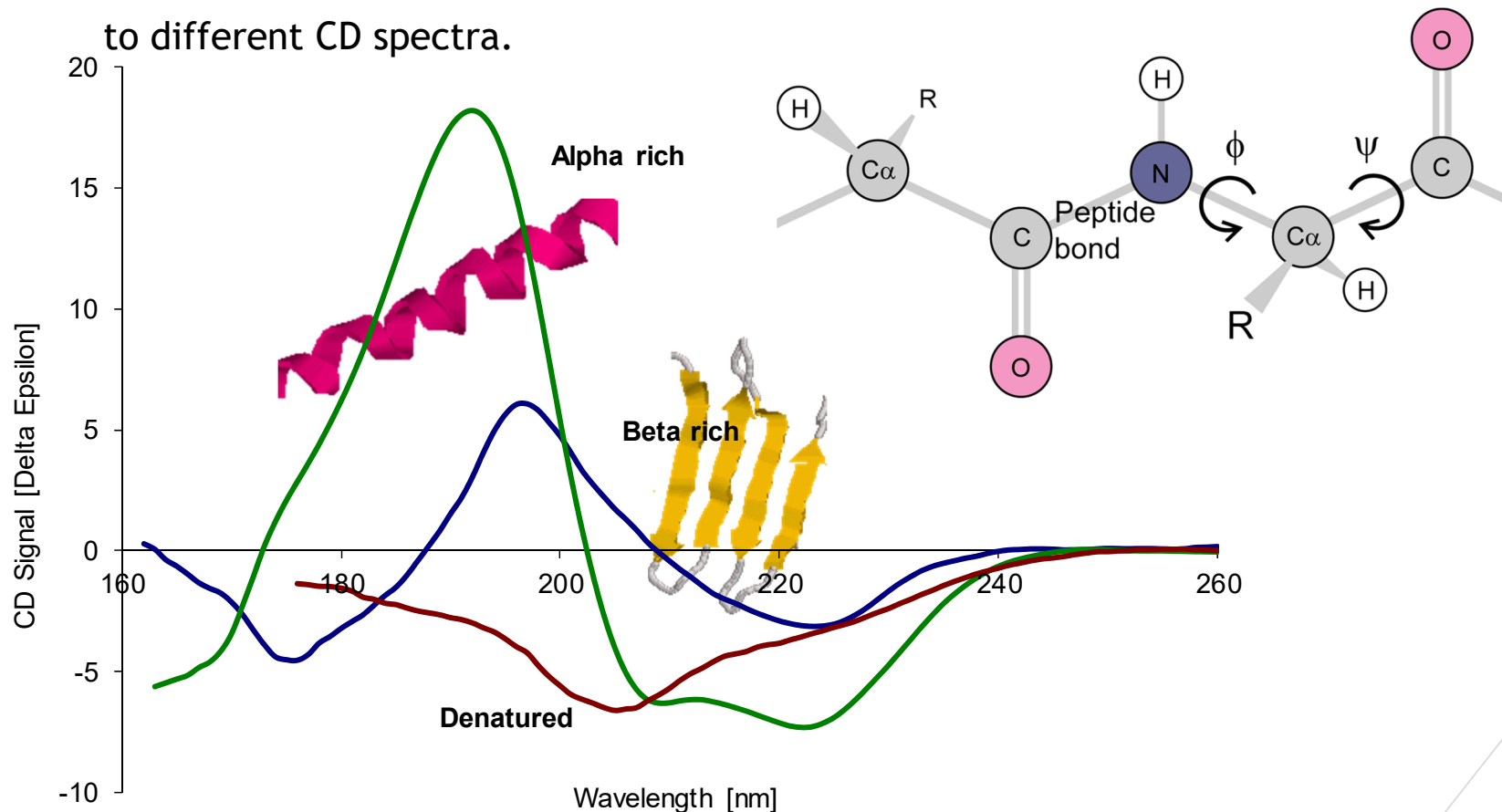
i.e. sensitive to the *Secondary Structure*

Obviously important for Circular Dichroism

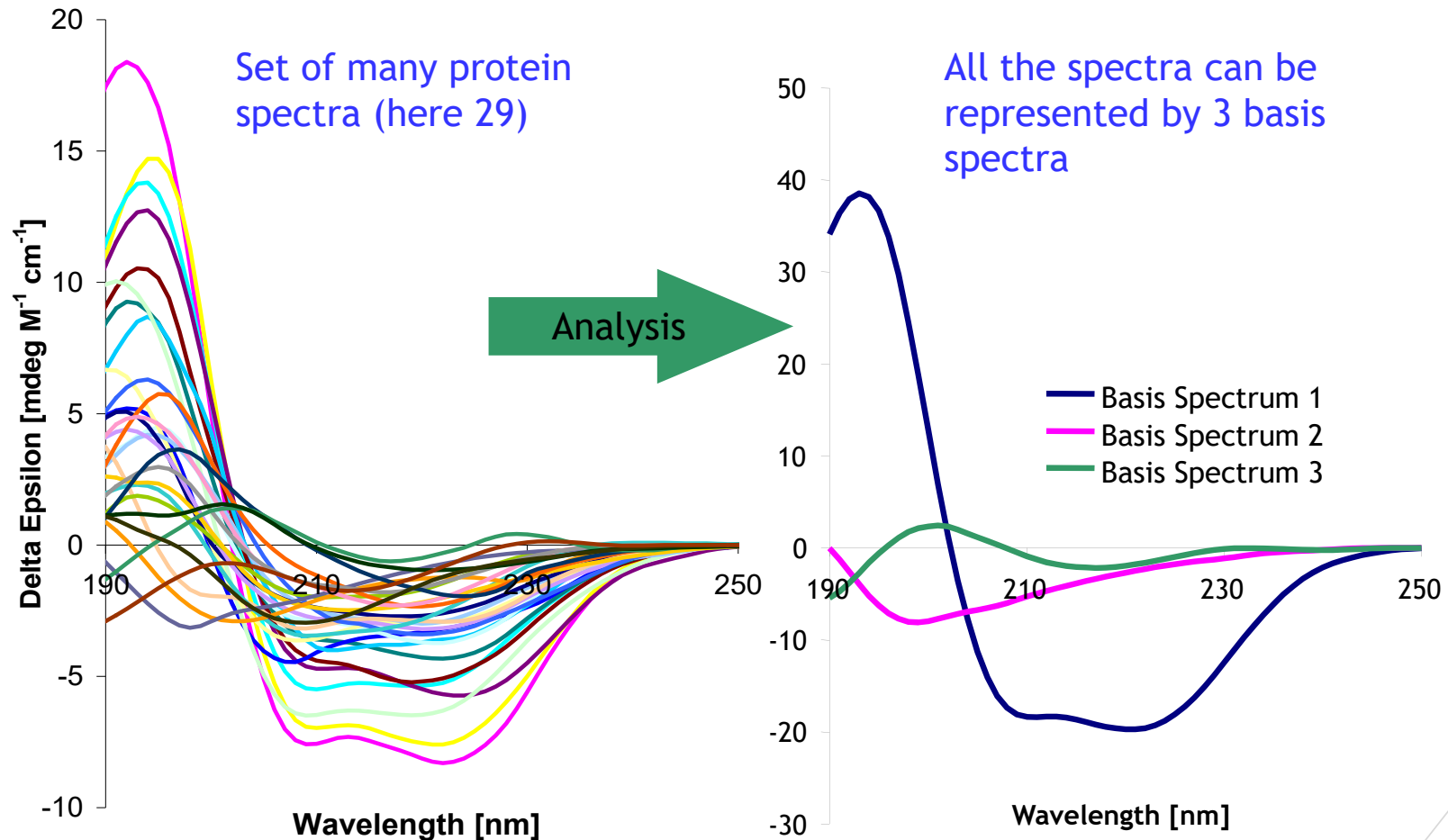


# Information in a protein CD spectrum

Different secondary structure in proteins gives rise to different CD spectra.

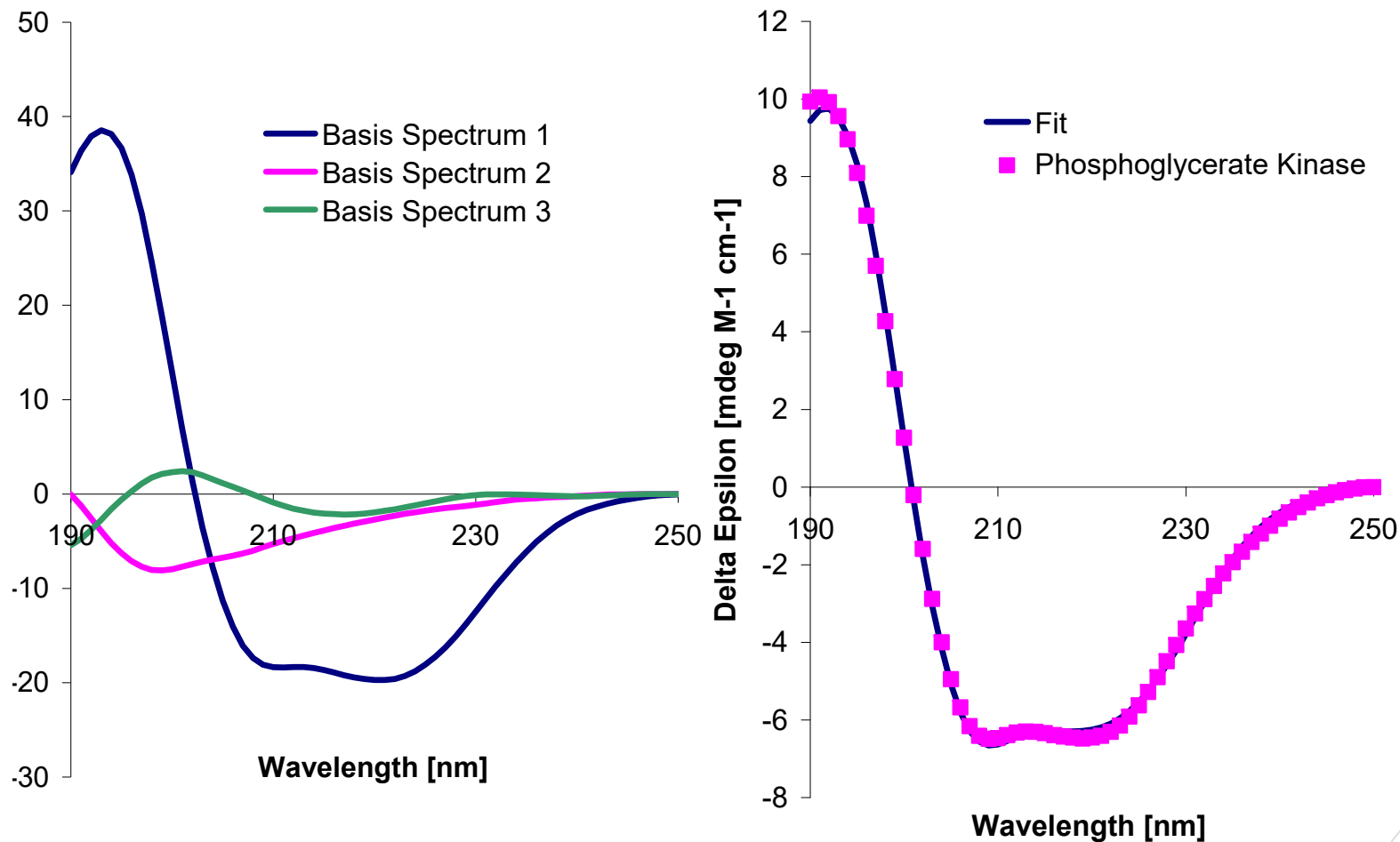


# Information in a protein CD spectrum

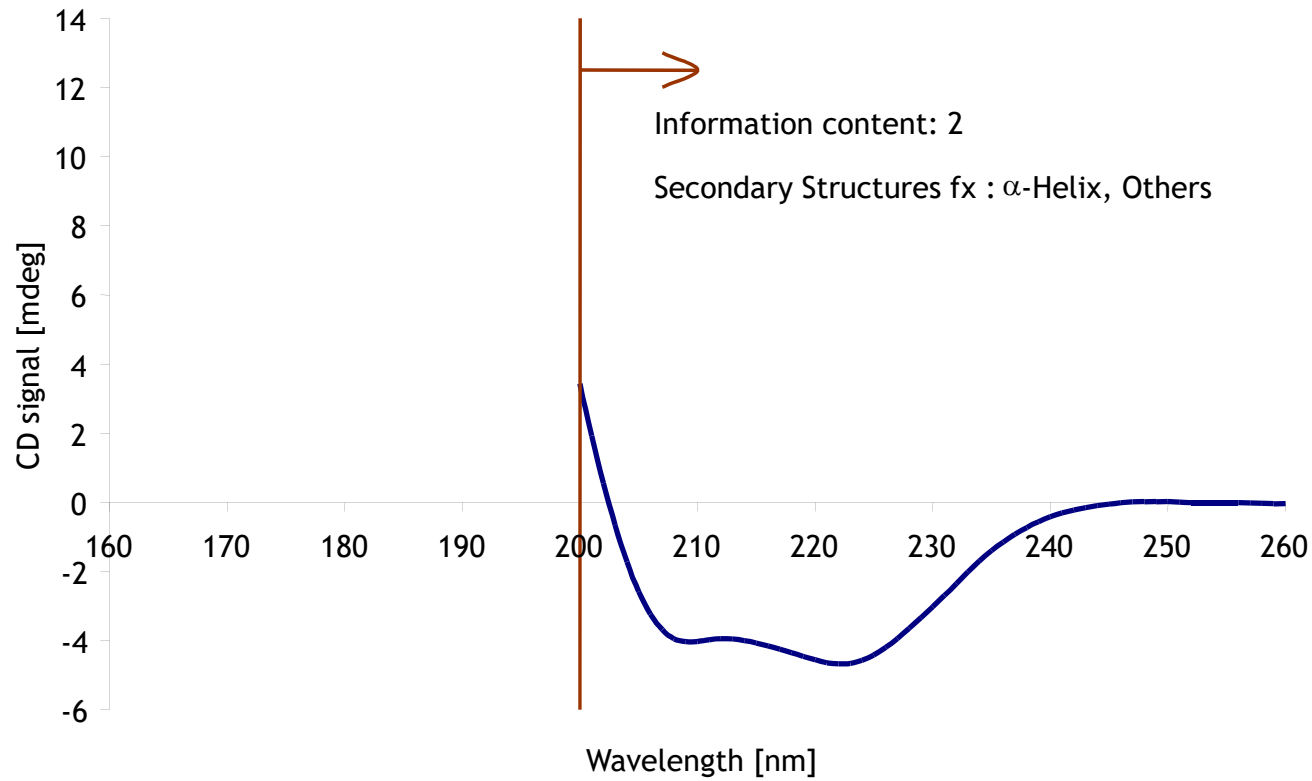




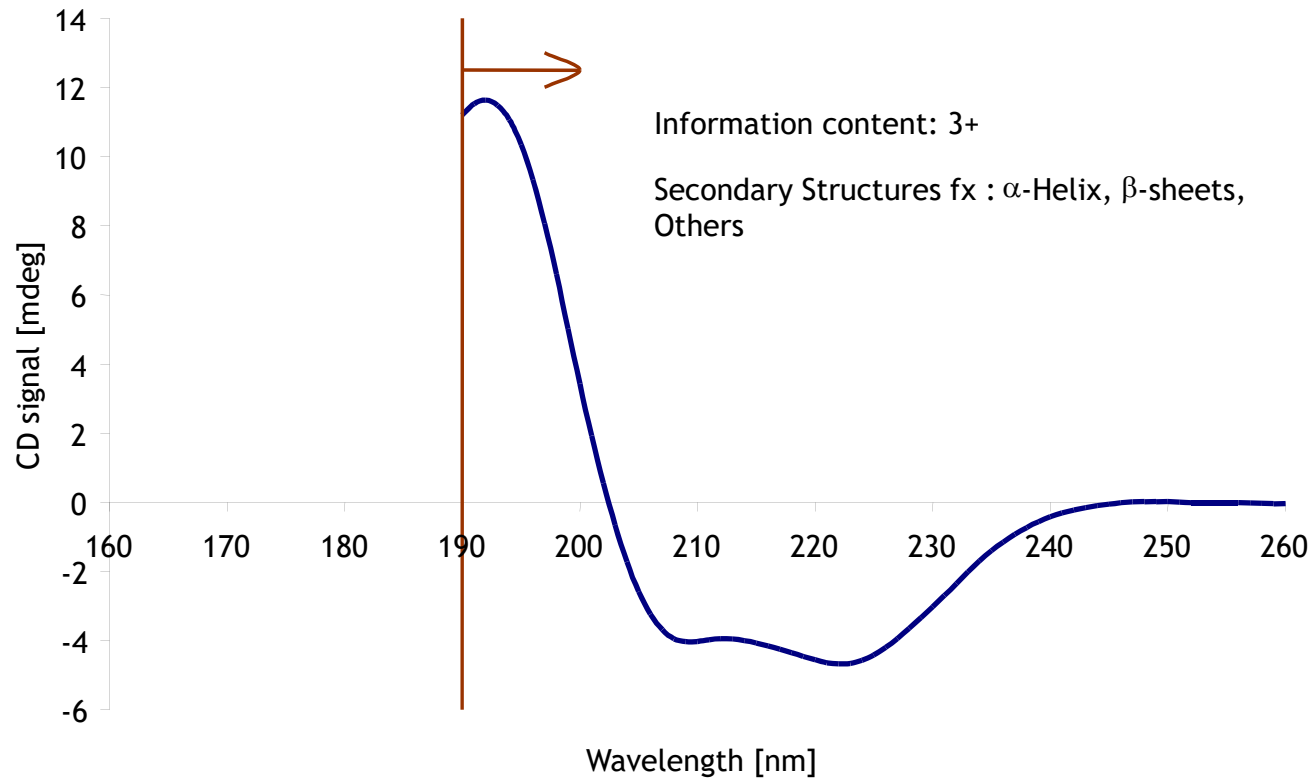
# All spectra may be reconstructed from the three basis spectra



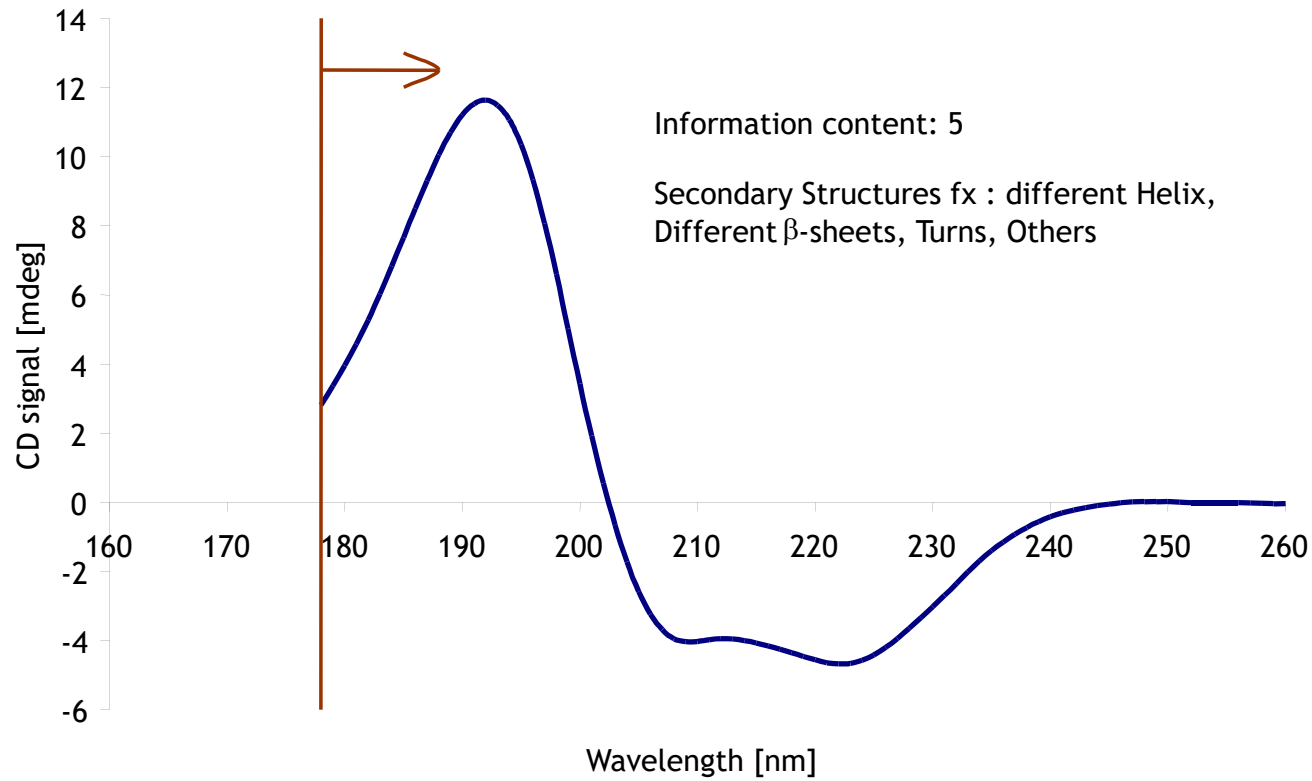
# Information in a protein CD spectrum



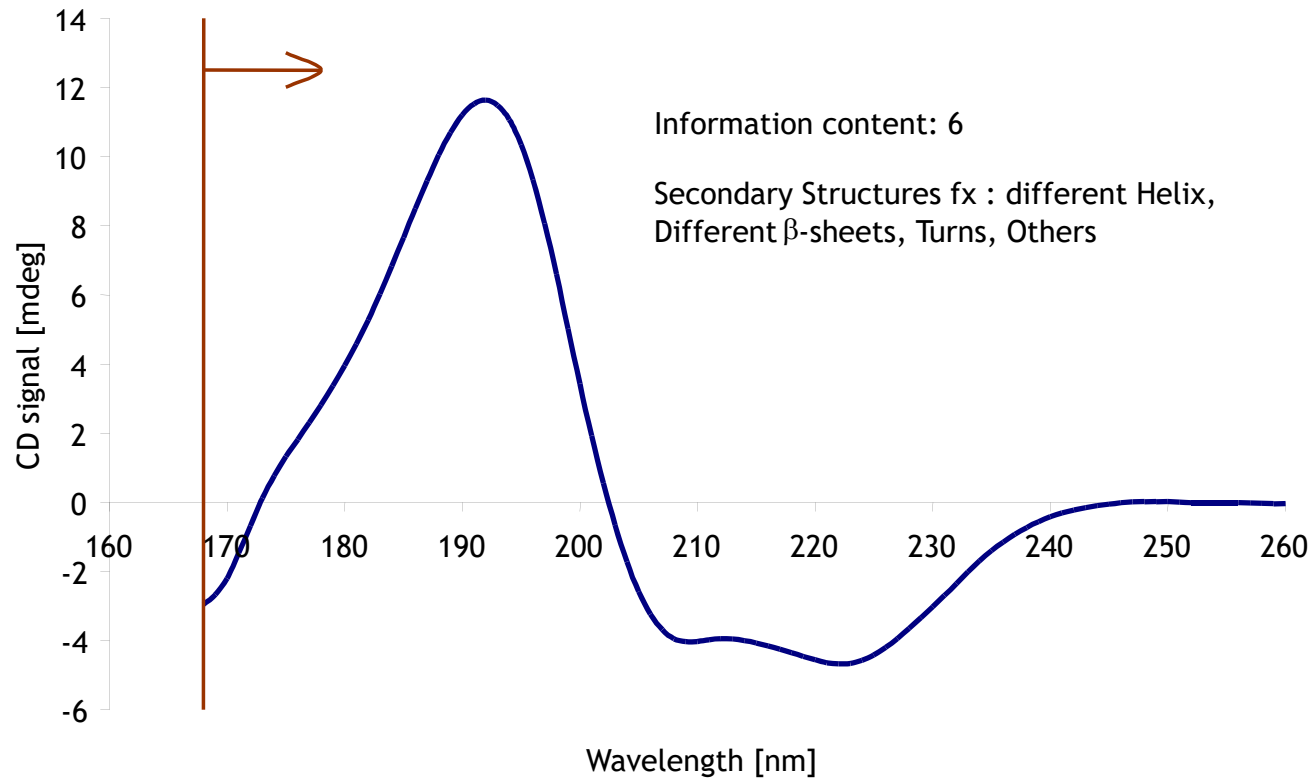
# Information in a protein CD spectrum



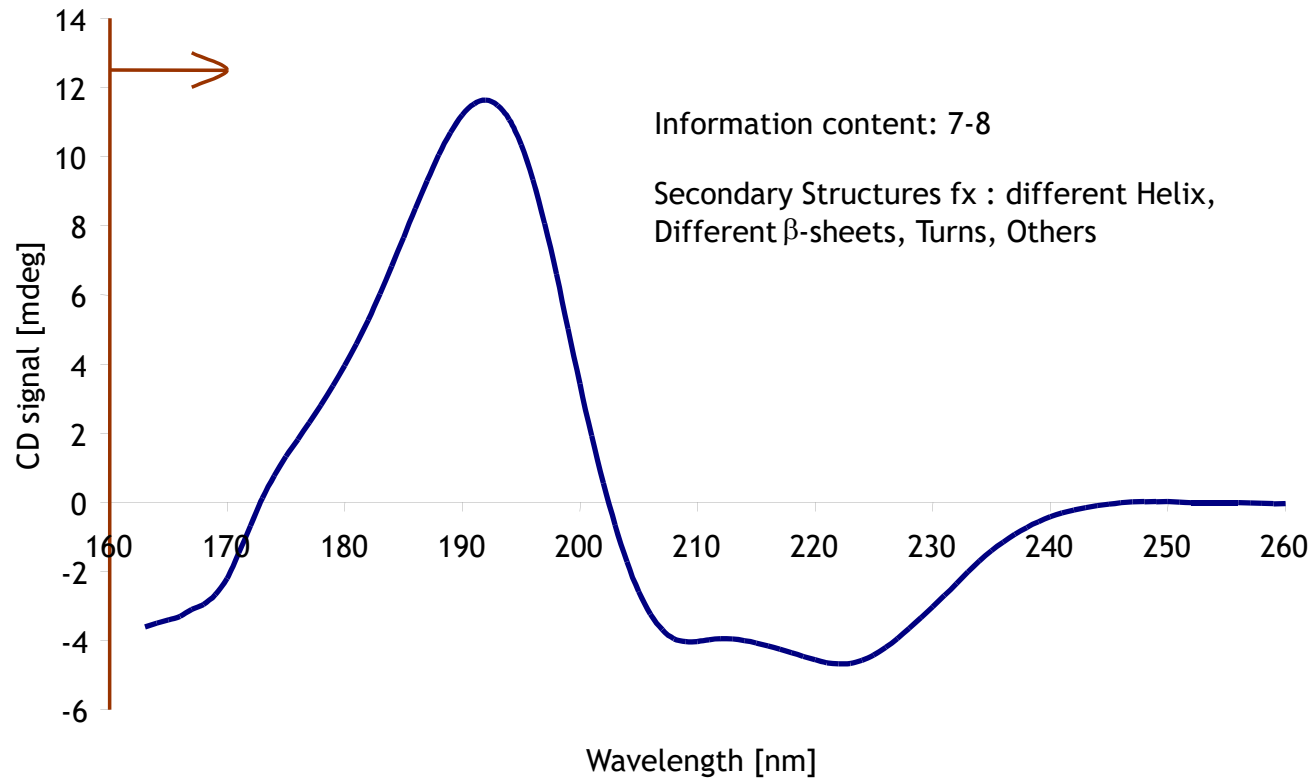
# Information in a protein CD spectrum



# Information in a protein CD spectrum



# Information in a protein CD spectrum



# CD units

## CD units:

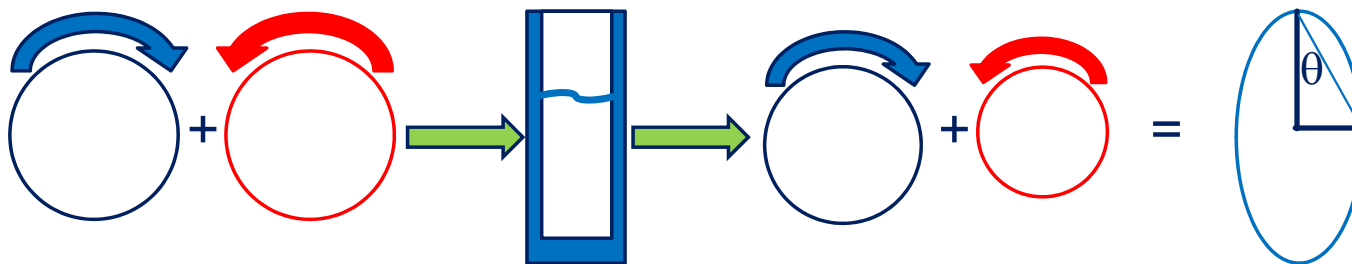
$$A = \log (I_0/I_t)$$

Typically  $\sim 0.1 - 1.5$

$$CD = \Delta A = A_L - A_R$$

Typically  $\sim 10^{-4} - 10^{-3}$

Historically CD measured as an angle (deg. or mdeg.)



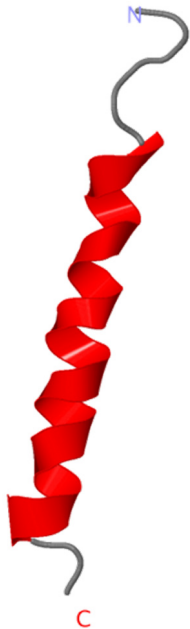
$$\theta[\text{mdeg}] = 32980 \Delta A$$

$\theta$  is typically 1-100 mdeg

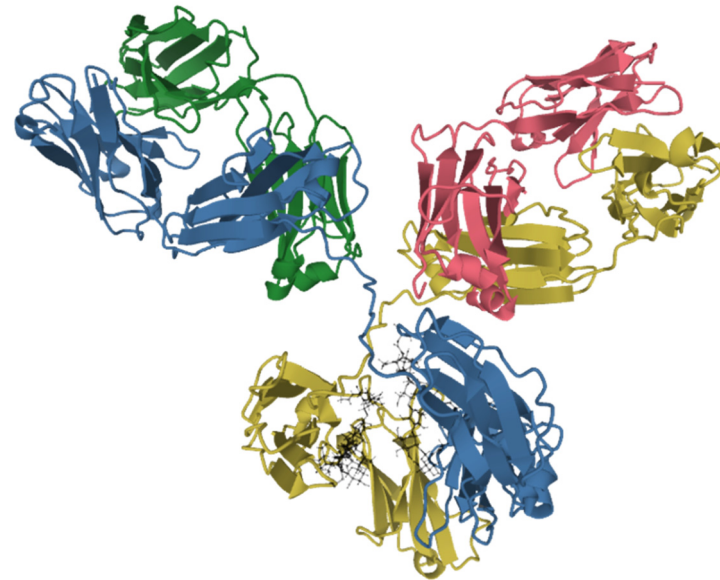
# CD spectrum: compare signals

CD units:  $CD = \Delta A = A_L - A_R$

Compare CD between proteins:



Short peptide  
Vs  
Large protein





# CD spectrum: compare signals

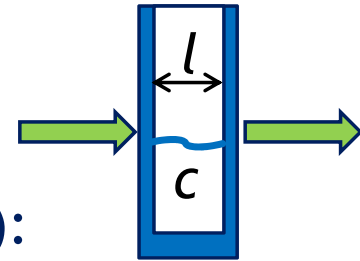
CD units:  $CD = \Delta A = A_L - A_R$

Compare CD between proteins:

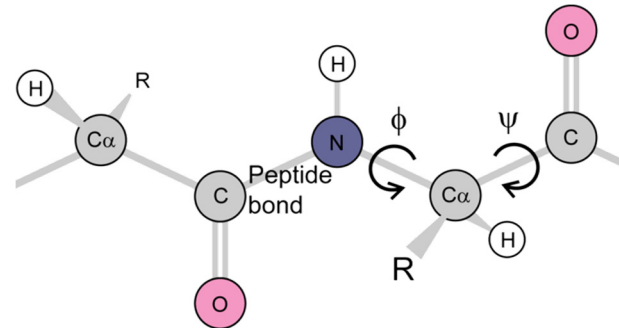
Correct for pathlength ( $l$ ) and concentration ( $c$ ):

$$\Delta \varepsilon = \Delta A / (l \cdot c)$$

Unit  $M^{-1} \text{ cm}^{-1}$



CD signal  
originates from  
each amino acid



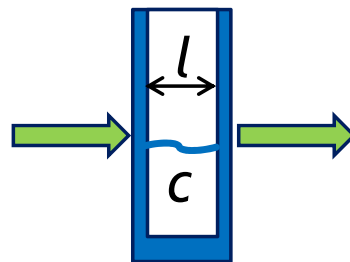
➡ Concentration is per residue/amino acid

# CD units

## CD units:

$$\Delta\epsilon = \Delta A / (l \cdot c)$$

Unit  $M^{-1} \text{ cm}^{-1}$



CD signal originates from each amino acid

➡ Concentration is per residue/amino acid

MRW: Mean Residual Weight (~110 Da)

Average weight of the amino acids in the protein

$c$  measured in mg/mL (g/L) ➡  $c/\text{MRW}$  is 'mol residues per L'

$$\Delta\epsilon [M^{-1} \text{ cm}^{-1}] = \text{CD}[\text{mdeg}] \cdot \text{MRW}[\text{Da}] / (32980 \cdot l[\text{cm}] \cdot C[\text{g/L}])$$

# How to get the concentration?

There are many ways used to get the concentration of proteins

- Quantitative Amino Acid Analysis
- Bradford Assay: Coomassie blue G-250 assay reagent, 595 nm
- BCA Protein Assay Kits: reduction of  $\text{Cu}^{2+}$  to  $\text{Cu}^{1+}$  by protein in an alkaline medium, 562 nm
- Protein absorbance 280 nm:
  - Concentration (mg/ml) = Absorbance at 280 nm divided by path length (cm)



*This just makes me sad*



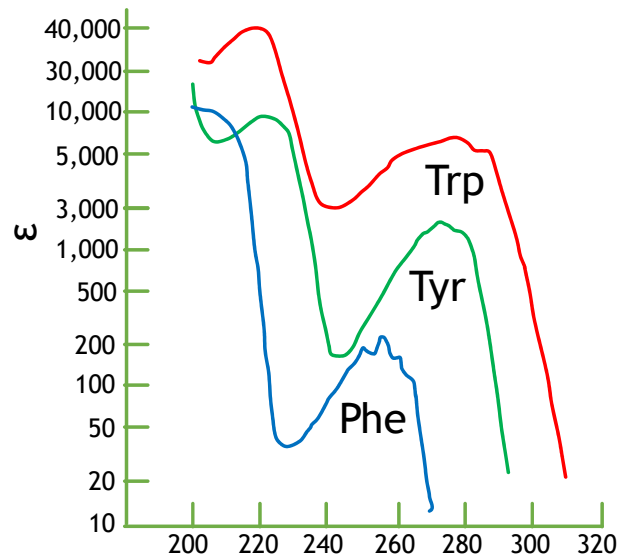
- **Better** use calculated  $\epsilon_{280}$  from primary sequence

$$c = A / (l \cdot \epsilon_{280}), \text{ Unit for } \epsilon_{280} \text{ M}^{-1} \text{ cm}^{-1}$$

# How to get the concentration?

Protein absorbance 280 nm:  $c = A / (l \cdot \epsilon_{280})$ , Unit for  $\epsilon_{280}$   $\text{M}^{-1} \text{cm}^{-1}$

The 280 nm protein absorbance in protein originates mostly from Tryptophan and Tyrosine



Known primary sequence

$$\epsilon_{280} = 5690 \cdot n_{\text{Trp}} + 1280 \cdot n_{\text{Tyr}}$$

*Absorbance at 280 nm is much lower than in the far-UV range (below 240 nm)*

- Often necessary to use different pathlengths ( $l$ ) for 280 nm and CD measurements



# How to get the concentration?

Protein absorbance 205 nm:  $c = A / (l \cdot \epsilon_{205})$ , Unit for  $\epsilon_{280}$   $M^{-1} \text{ cm}^{-1}$

*This is our method of choice for CD measurements*

N.J. Anthis and G.M. Clore. *Protein Science* **22** (2013) 851-858

Side chain/feature	$\epsilon_{205}$ ( $M^{-1} \cdot \text{cm}^{-1}$ )
Tryptophan	20,400
Phenylalanine	8600
Tyrosine	6080
Histidine	5200
Methionine	1830
Arginine	1350
Cysteine	690
Asparagine <sup>a</sup>	400
Glutamine <sup>a</sup>	400
Cystine <sup>b</sup>	2200
Backbone peptide bond <sup>c</sup>	$2780 \pm 168$

<sup>b</sup>If the protein has a disulfide bond, add 820 to its  $\epsilon_{205}$

Calculate the  $\epsilon_{205}$

$$\epsilon_{205} = \sum_{aa} (n_{aa} \cdot \epsilon_{aa}) + n_{S-S} \cdot 820 + (\text{Seq. length} - 1) \cdot \epsilon_{backbone}$$

Sum over amino acids in table

Disulfide bonds

$n_{aa}$  : number of type *aa* amino acid

Seq. length : number of all amino acids

***The absorbance is measured simultaneously with CD on the same exact sample !!!***



# How to get the concentration?

Protein absorbance 205 nm:  $c = A / (l \cdot \epsilon_{205})$ , Unit for  $\epsilon_{280}$   $M^{-1} \text{ cm}^{-1}$

Primary sequence of Myoglobin:

MGLSDGEWQQVLNVWGKVEADIAGHGQEVLRFTGHPETLEKFDK  
FKHLKTEAEMKASEDLKKHGTVVLTALGGILKKKGHHEALKPLAQSH  
ATKHKIPIKYLEFISDAIIHVLHSHKHPGDFGADAQGAMTKALELFRNDIA  
AKYKELGFQG

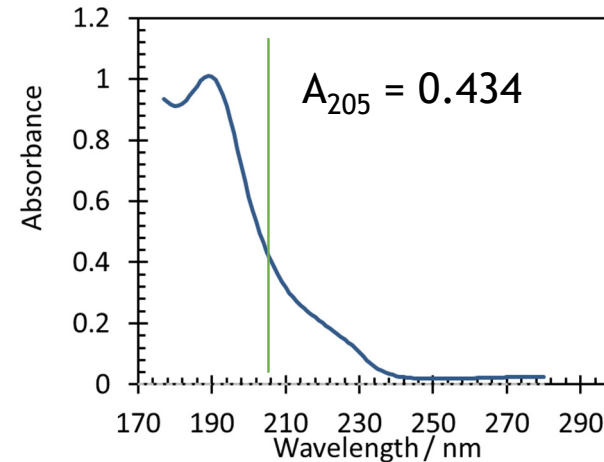
$$\epsilon_{205} = \sum_{aa} (n_{aa} \cdot \epsilon_{aa}) + n_{S-S} \cdot 820 + (\text{Seq. length} - 1) \cdot \epsilon_{backbone}$$

For the entire protein:

$$\epsilon_{205} = 607\,090 \text{ M}^{-1} \text{ cm}^{-1}$$

Per residue:

$$\epsilon_{205} = 3942 \text{ M}^{-1} \text{ cm}^{-1}$$



$$c \cdot l = A / \epsilon_{205} = 0.434 / 3942 = 0.11 \text{ mM} \cdot \text{cm}$$

## Amino Acids

Code	$n_{aa}$	$\epsilon_i$
F	7	8600
M	3	1830
C	0	690
Y	2	6080
W	2	20400
Q	6	400
N	2	400
H	11	5200
R	2	1350
Disulfide bonds	0	820
Sequence length	154	
Seq. length - 1	153	2780



# How to get the concentration?

What if I want the concentration in mg/ml ?

Primary sequence of Myoglobin:

```
MGLSDGEWQQVLNVWGKVEADIAGHGQEVLRFTGHPETLEKFDKFKHLKTEAEKASEDLKKHGTVVLTALGGILKKKGHHEAEKPLAQSHATKHKIPIKYLEFISDAIIHVLHSHKHPGDFGADAQGA  
MTKALELFRNDIAAKYKELGFQG
```

$$MW_{protein} = \sum_{aa} (n_{aa} \cdot MW_{aa}) - (Seq. length - 1) \cdot MW_{water}$$

Amino Acids					
Code	n <sub>aa</sub>	MW <sub>aa</sub>			
I	9	113.2	Y	2	163.2
L	17	113.2	W	2	186.2
V	7	99.1	Q	6	128.1
F	7	147.2	N	2	114.1
M	3	131.2	H	11	137.1
C	0	103.1	E	13	129.1
A	15	71.1	D	8	115.1
G	15	57.1	K	19	128.2
P	4	97.1	R	2	156.2
T	7	101.1	Sequence length	154	
S	5	87.1	Water	-153	18.015

Molecular weight

MW = 17082.6 Da

Mean Residual Weight

MRW = 110.93 Da

Using:

$c \cdot l = 0.11 \text{ mM} \cdot \text{cm}$ ,  $l = 0.01 \text{ cm}$

We get:

$c[\text{mg/ml}]$

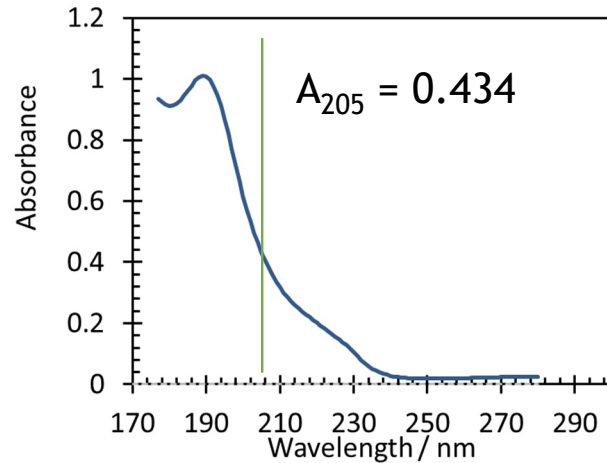
$= 110.93 \text{ g/mol} \cdot 0.11 \text{ mM} \cdot \text{cm} / 0.01 \text{ cm}$

$= \underline{1.2 \text{ mg/ml}}$



# What concentration to use?

Protein absorbance 205 nm and below:



$$c \cdot l = A / \epsilon_{205} = 0.434 / 3942 = 0.11 \text{ mM} \cdot \text{cm}$$

$$\begin{aligned} c[\text{mg/ml}] &= 110.93 \text{ g/mol} * 0.11 \text{ mM} \cdot \text{cm} / 0.01 \text{ cm} \\ &= \underline{1.2 \text{ mg/ml}} \end{aligned}$$

*Measured in a 0.01 cm = 0.1 mm pathlength cell*

Notice that for this protein the absorbance peaks at ~1 near 190 nm

➤ This is a good absorbance to optimize signal to noise, and thus a good concentration

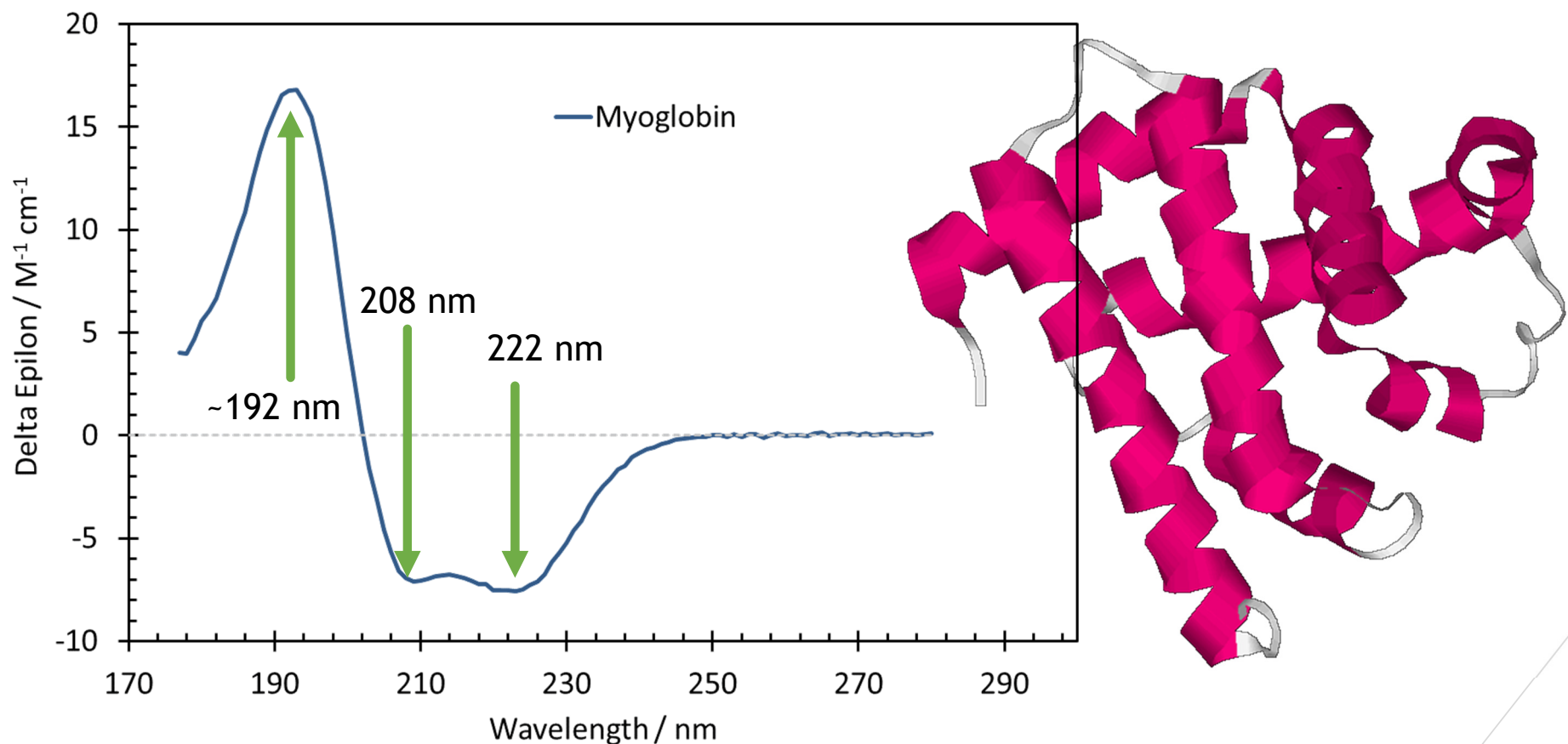
Roughly: 1 mg/ml protein in a 0.1 mm cell

- Lower for highly alpha helical proteins
- Higher for highly beta sheet or unordered proteins
- Scale concentration with pathlength of the cell you are using



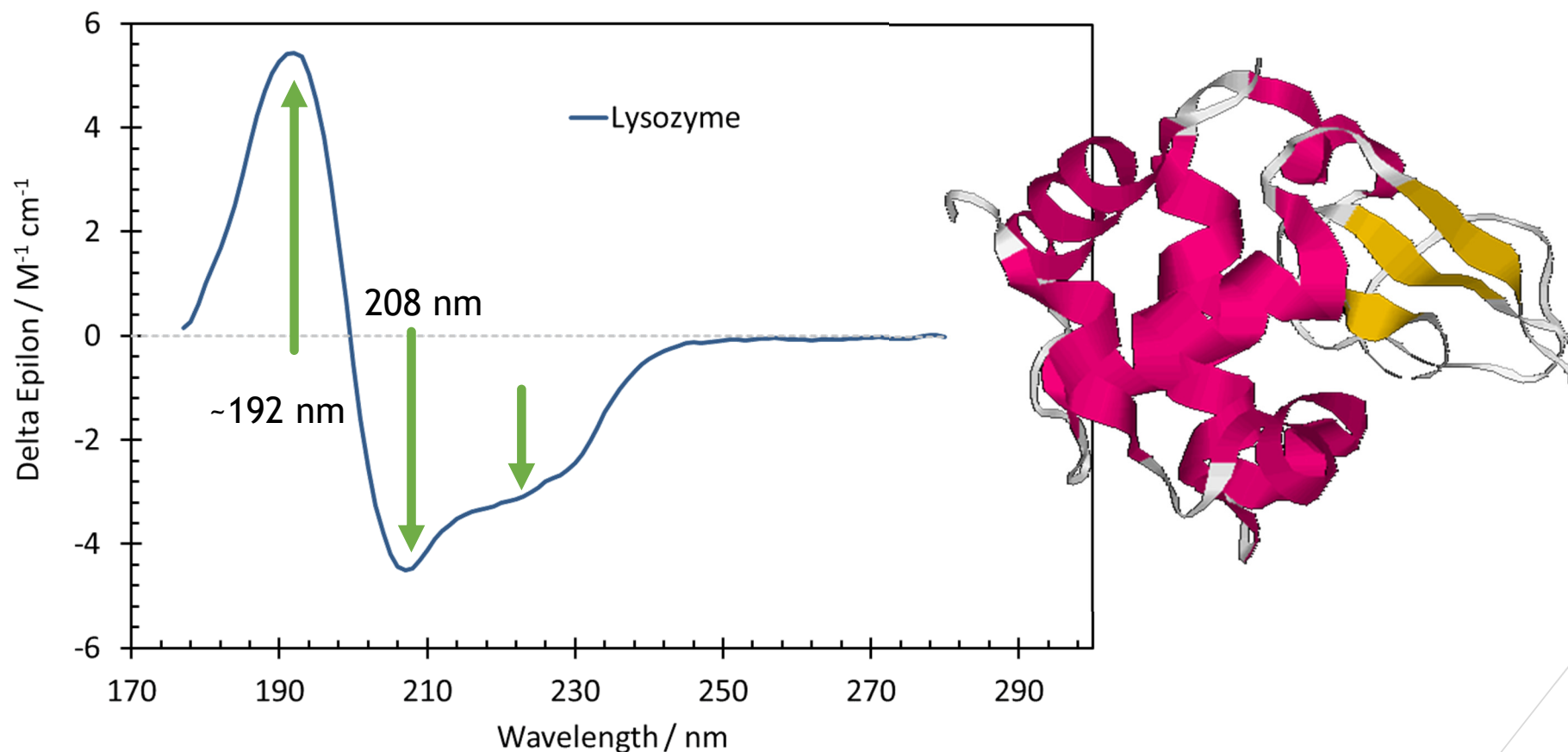
# Getting acquainted with protein CD spectra

**Myoglobin** *A typical Alpha Helical protein*



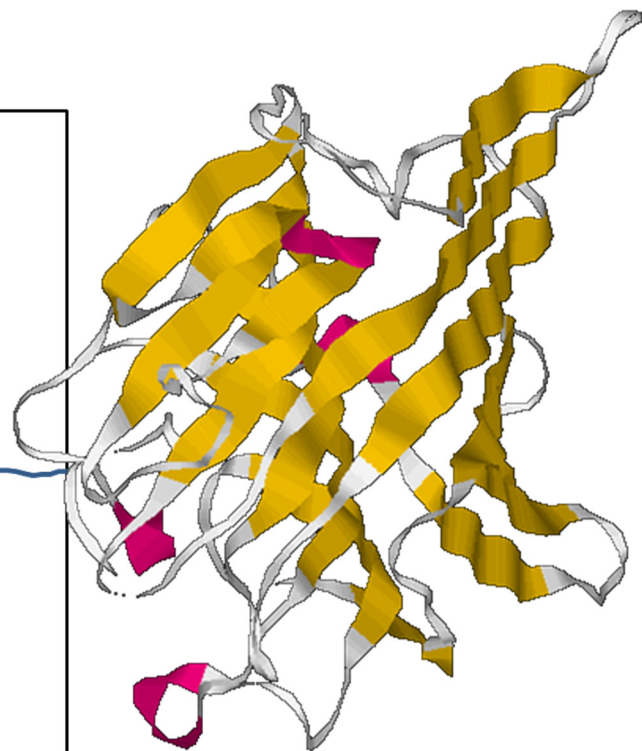
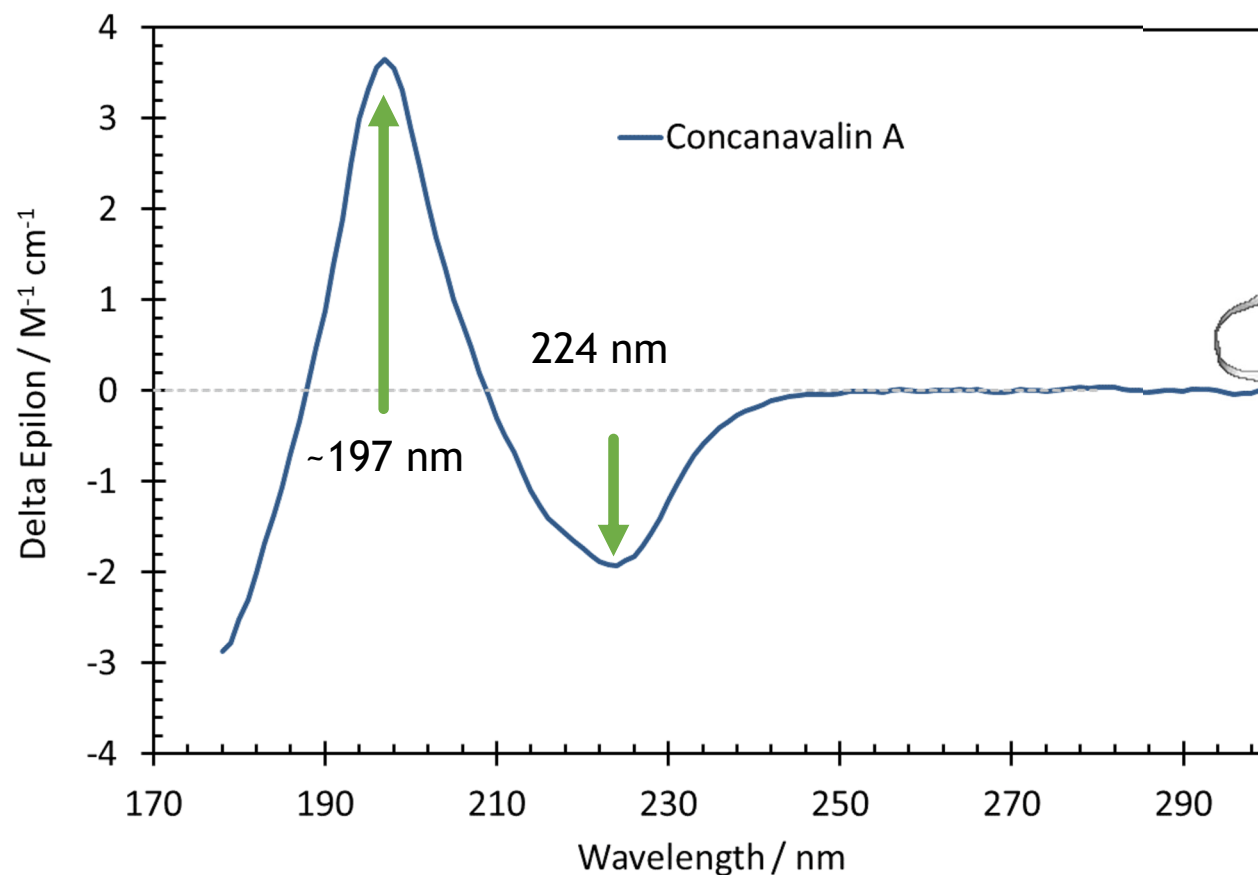
# Getting acquainted with protein CD spectra

**Lysozyme** *A Alpha Helical and Beta sheet mix protein*



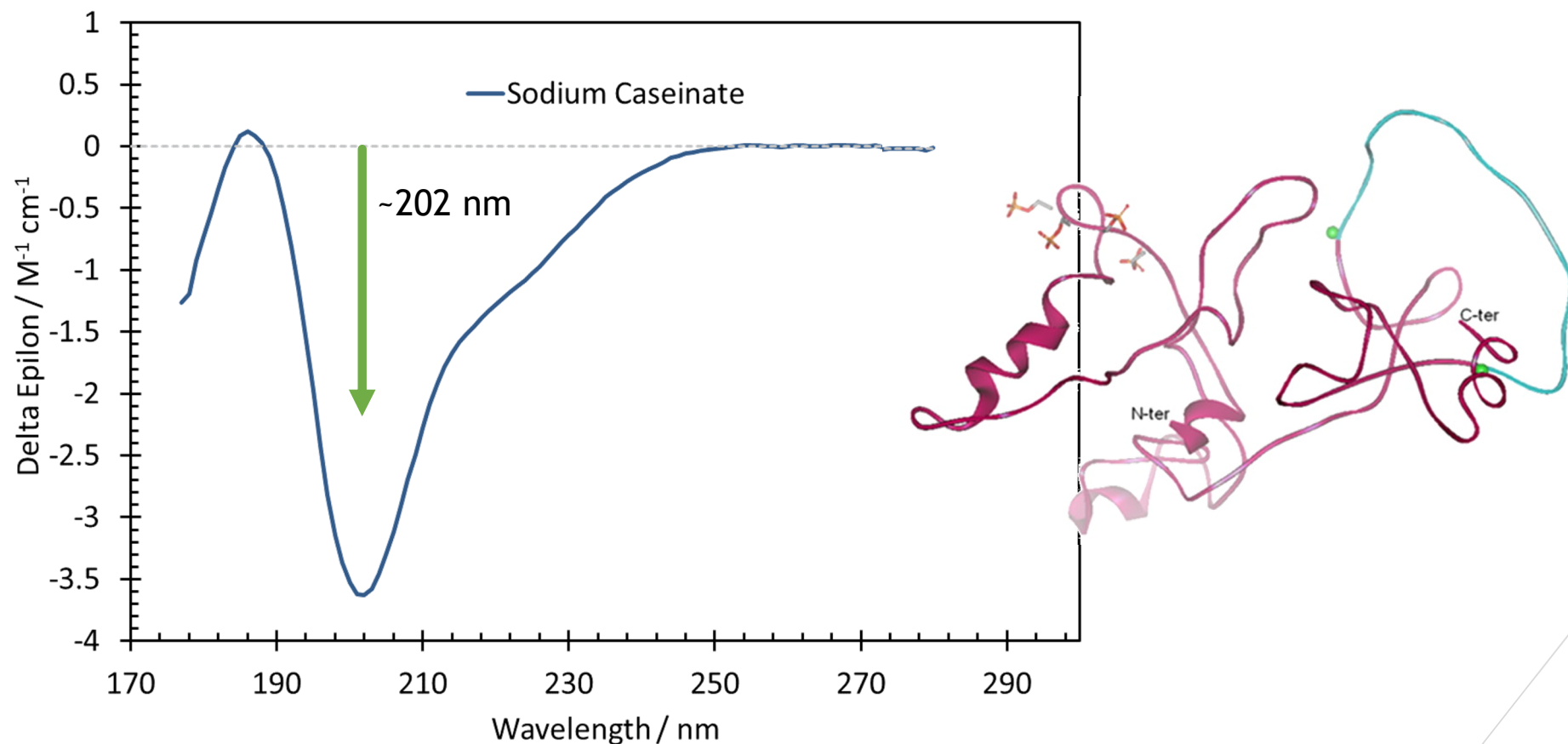
# Getting acquainted with protein CD spectra

## Concanavalin A *A Beta sheet protein*



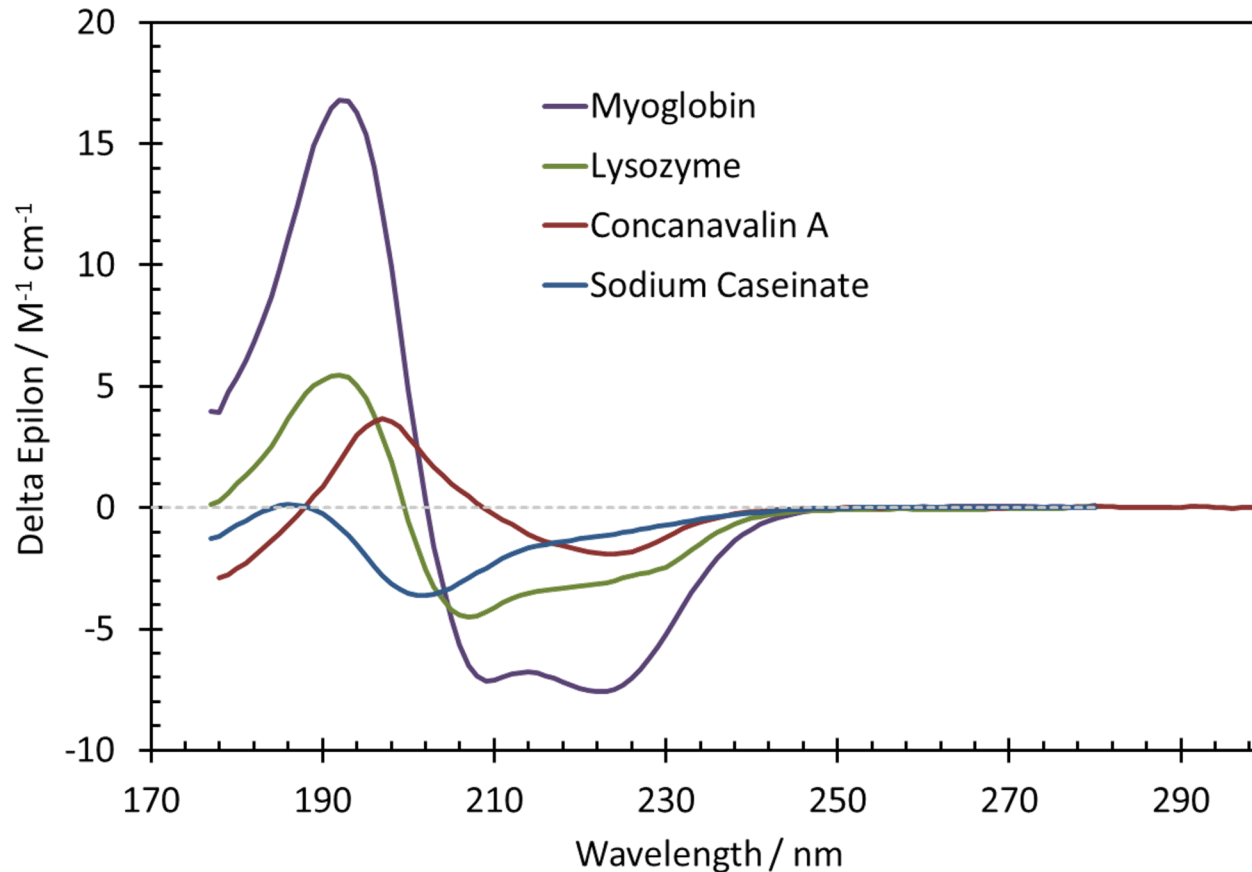
# Getting acquainted with protein CD spectra

## Sodium Caseinate *Intrinsically disordered protein (IDP)*



# Getting acquainted with protein CD spectra

## All spectra



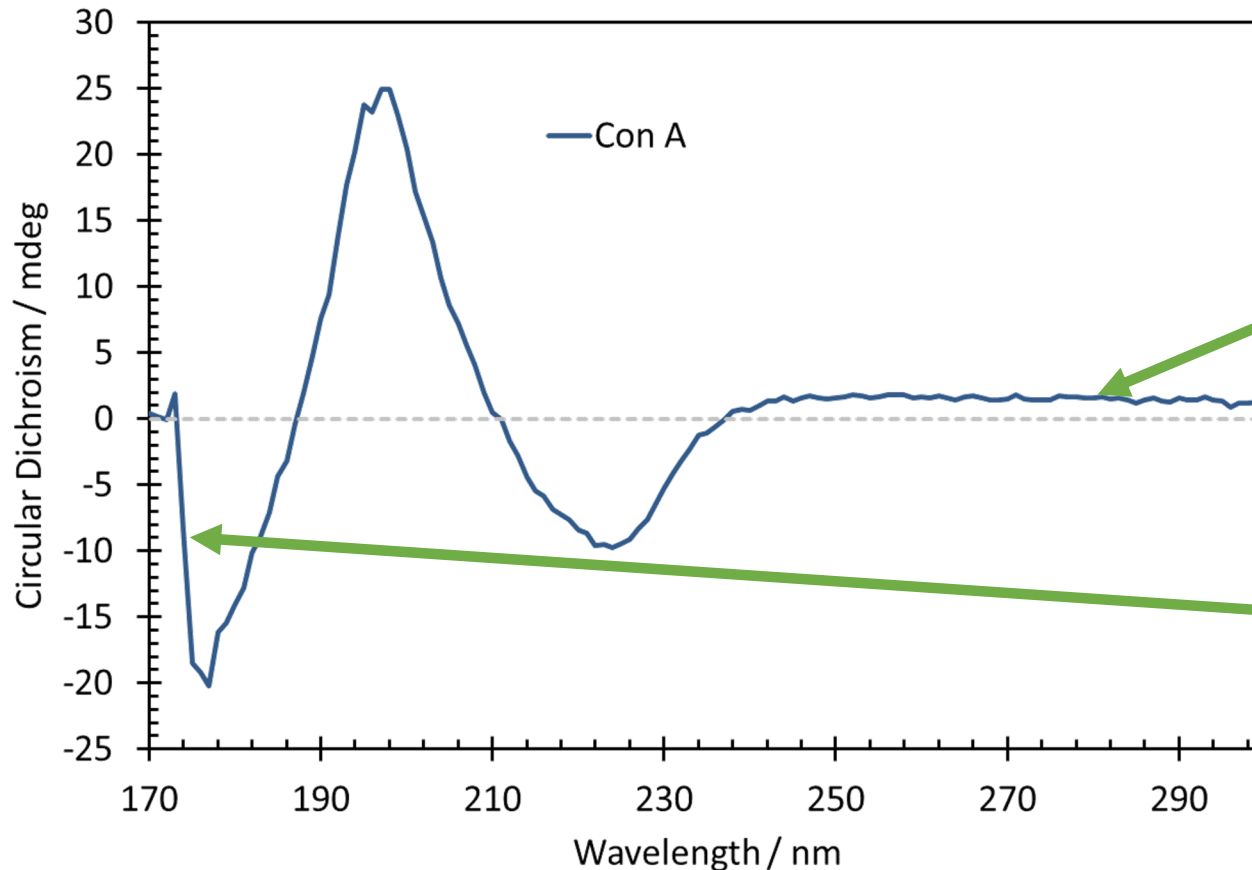
Note the difference in

- Signal magnitude
- Peak position
- Zero signal crossings

*All this is why we can differentiate between different secondary structure components*

# Actually acquiring data

## Sample scan



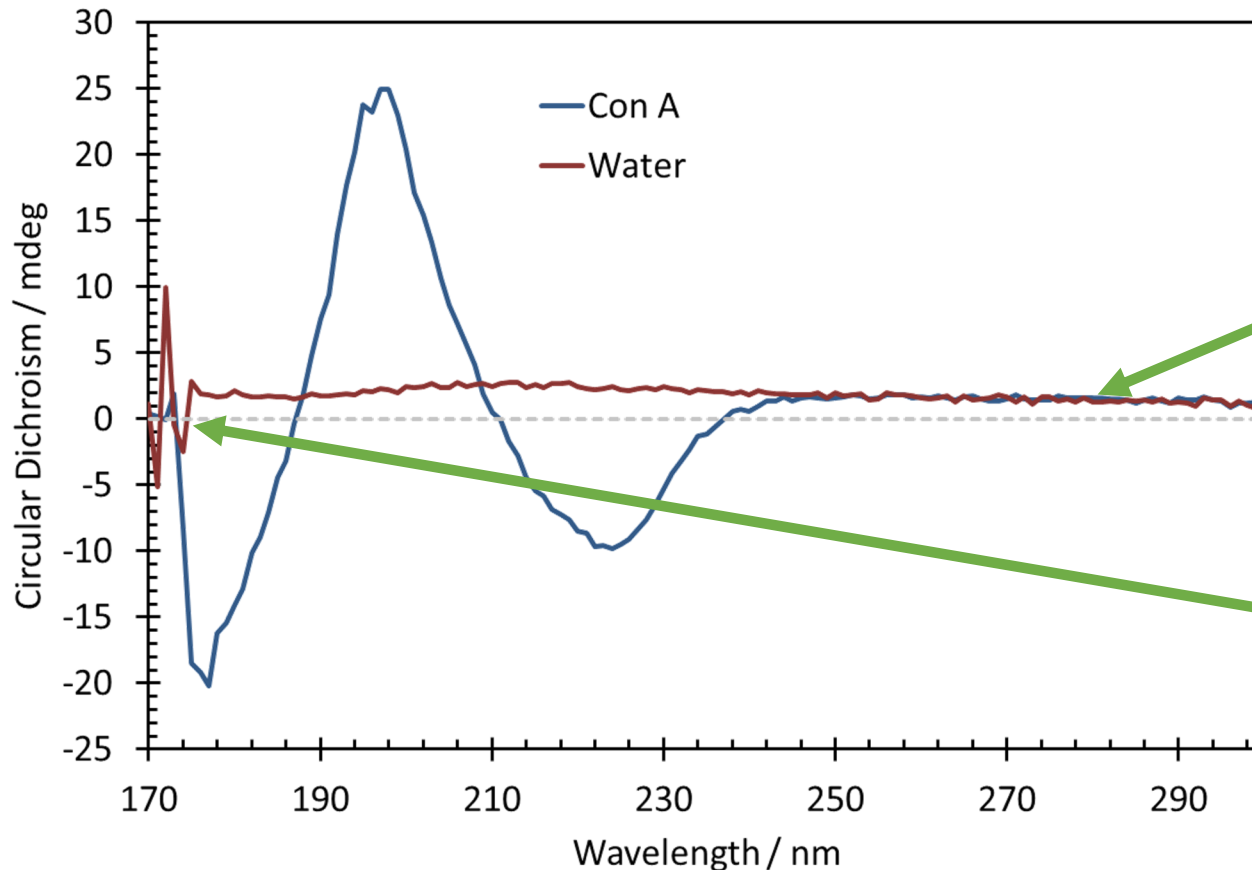
Acquiring a single scan of your sample

Non Zero signal at long wavelength !

Abrupt change at low wavelength, tending to zero !

# Actually acquiring data

## Sample and buffer scan



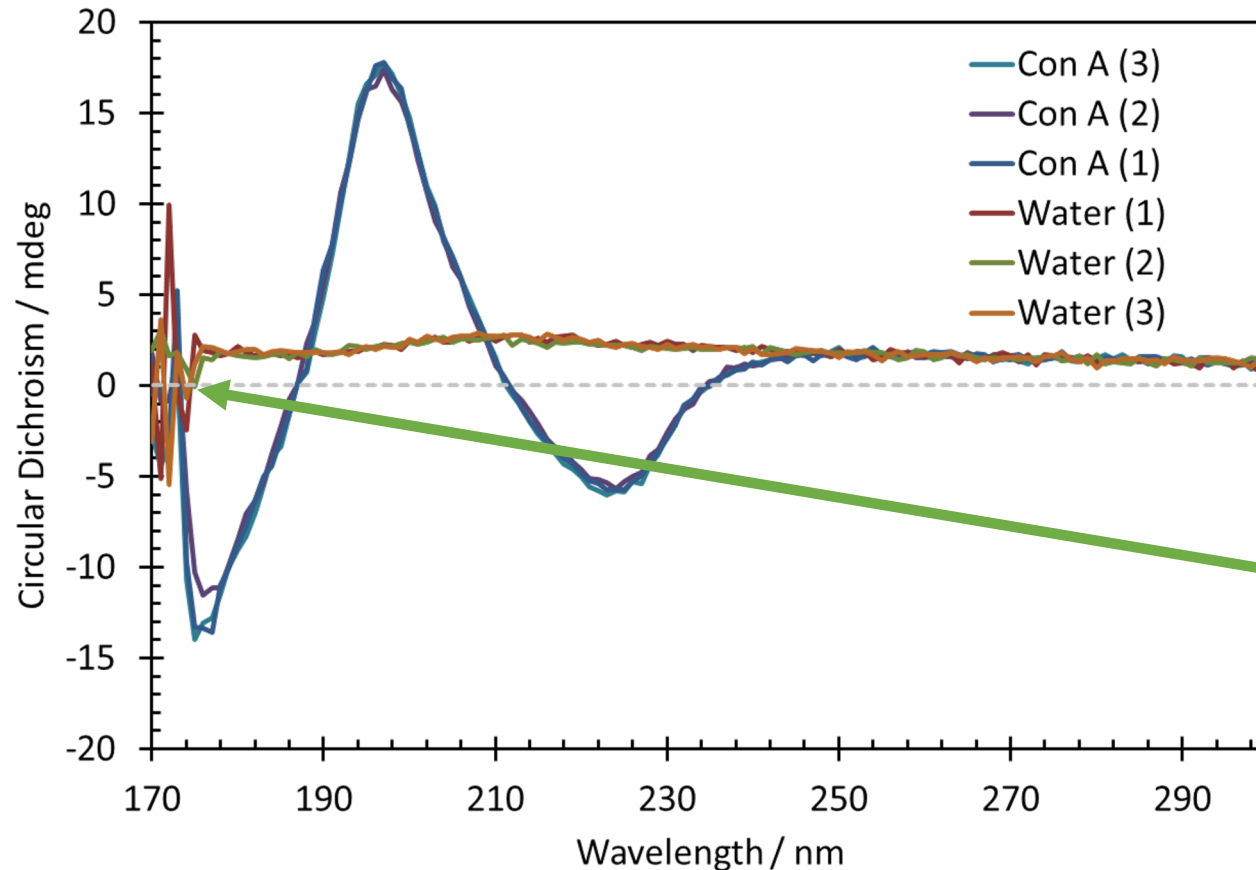
Acquiring a single scan of your sample and buffer

Buffer matches at long wavelength

Abrupt change at low wavelength, tending to zero and noisy !

# Actually acquiring data

## Multiple sample and buffer scans



Acquiring multiple scans of your sample and buffer

All spectra match within the noise

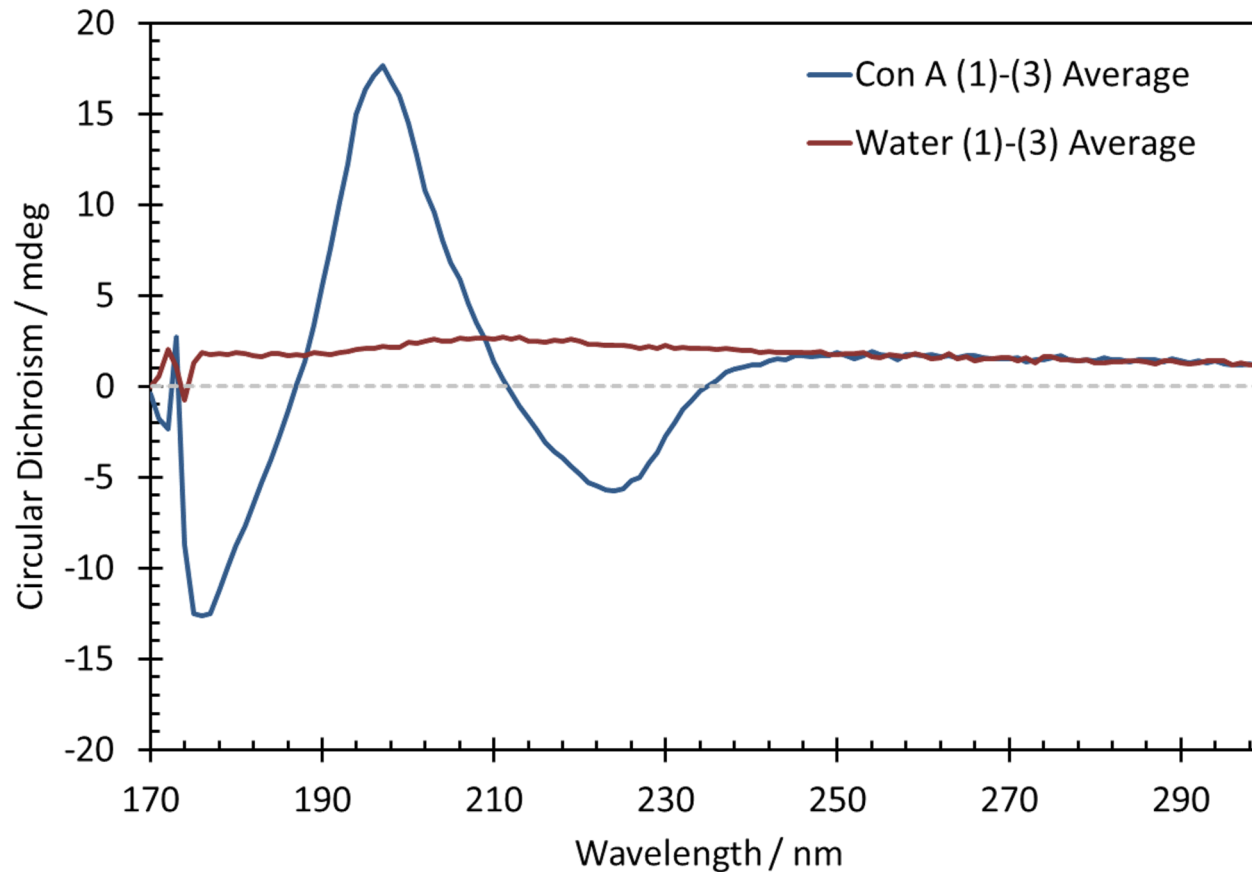
Why the noise at low wavelength ?

*Buffer and sample absorption high*



# Actually acquiring data

## Multiple sample and buffer scans - Average scans

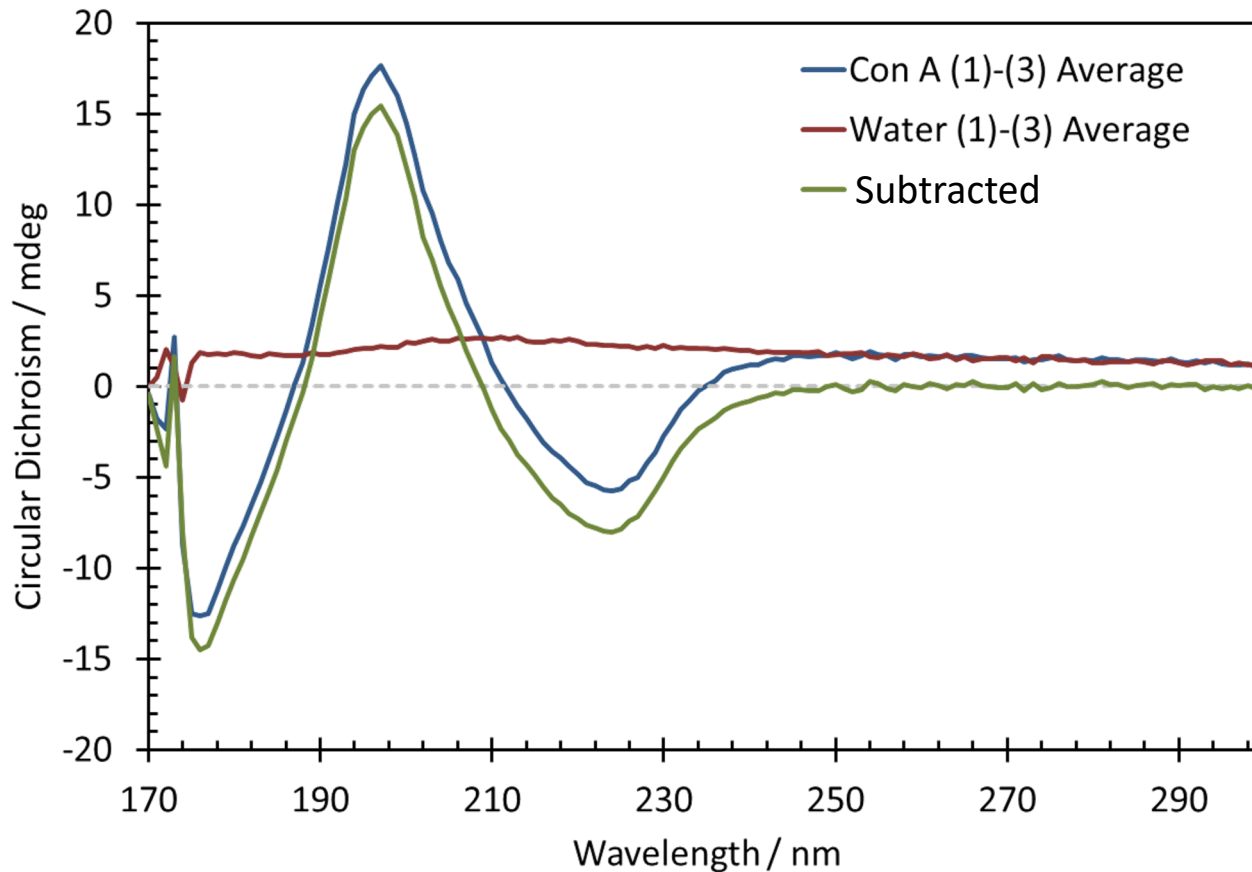


Acquiring multiple scans of your sample and buffer

**Average sample and buffer scans**

# Actually acquiring data

Multiple sample and buffer scans - Average scans - Subtract

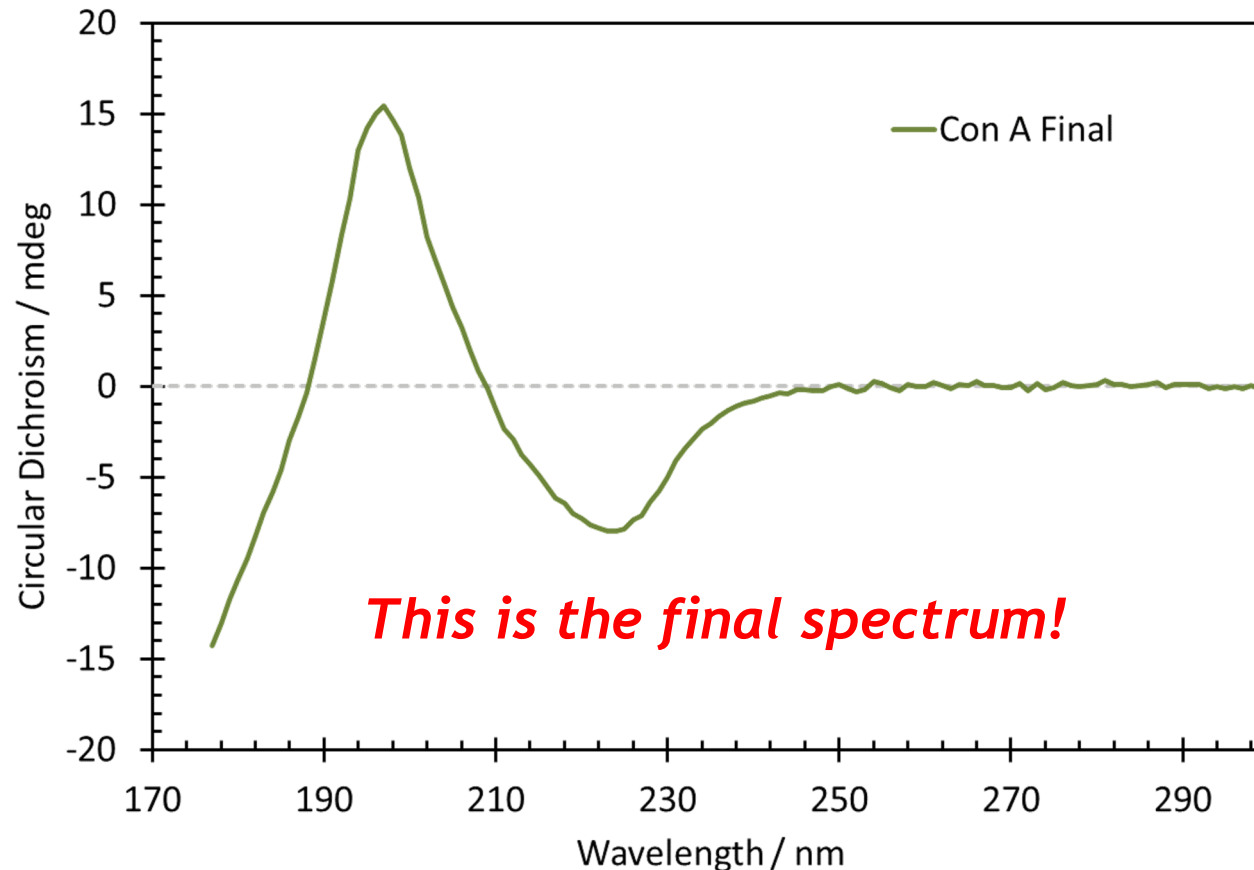


Acquiring multiple scans of your sample and buffer

**Subtract the average buffer scan from the average sample scan**

# Actually acquiring data

Multiple sample and buffer scans - Average scans - Subtract



Acquiring multiple scans of your sample and buffer

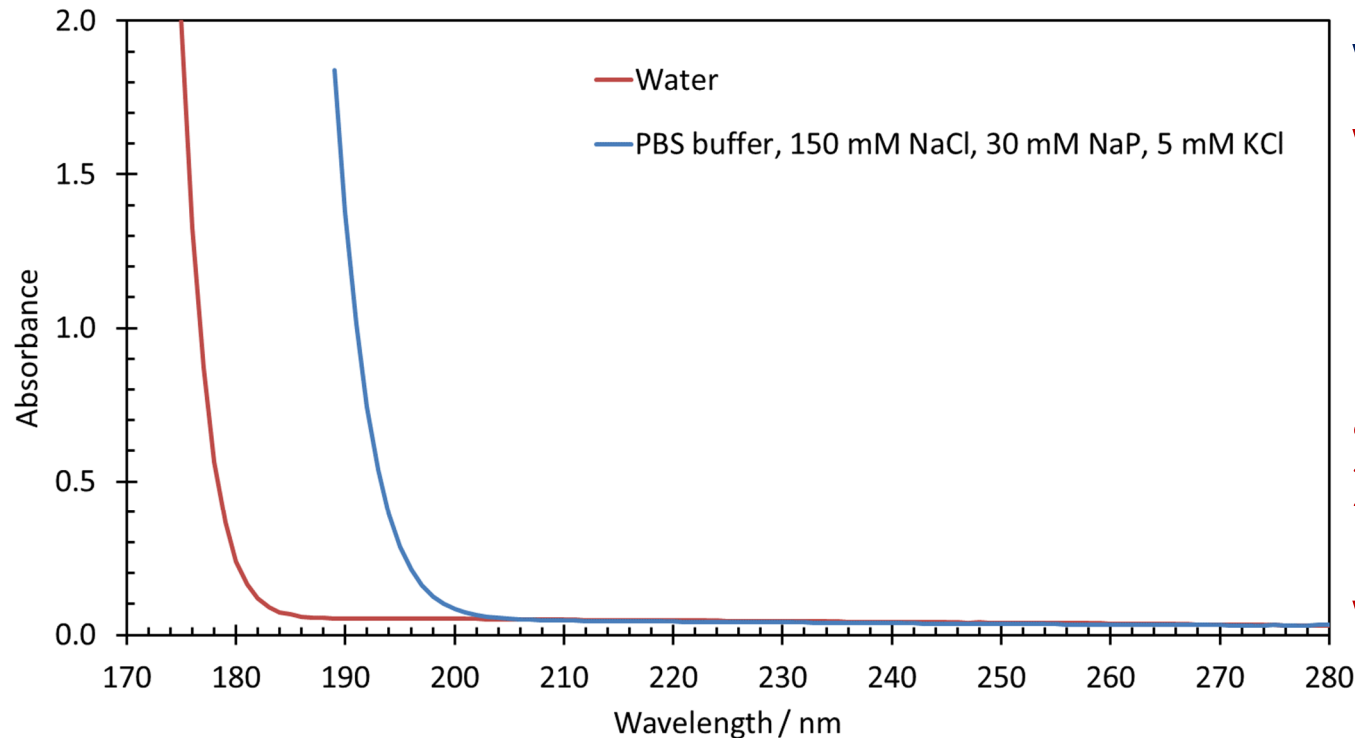
**Remove the untrustworthy low wavelength data points**

... and you are done



# Sample buffers

What is the best choice for buffer for *far-UV* CD



Water is the reference

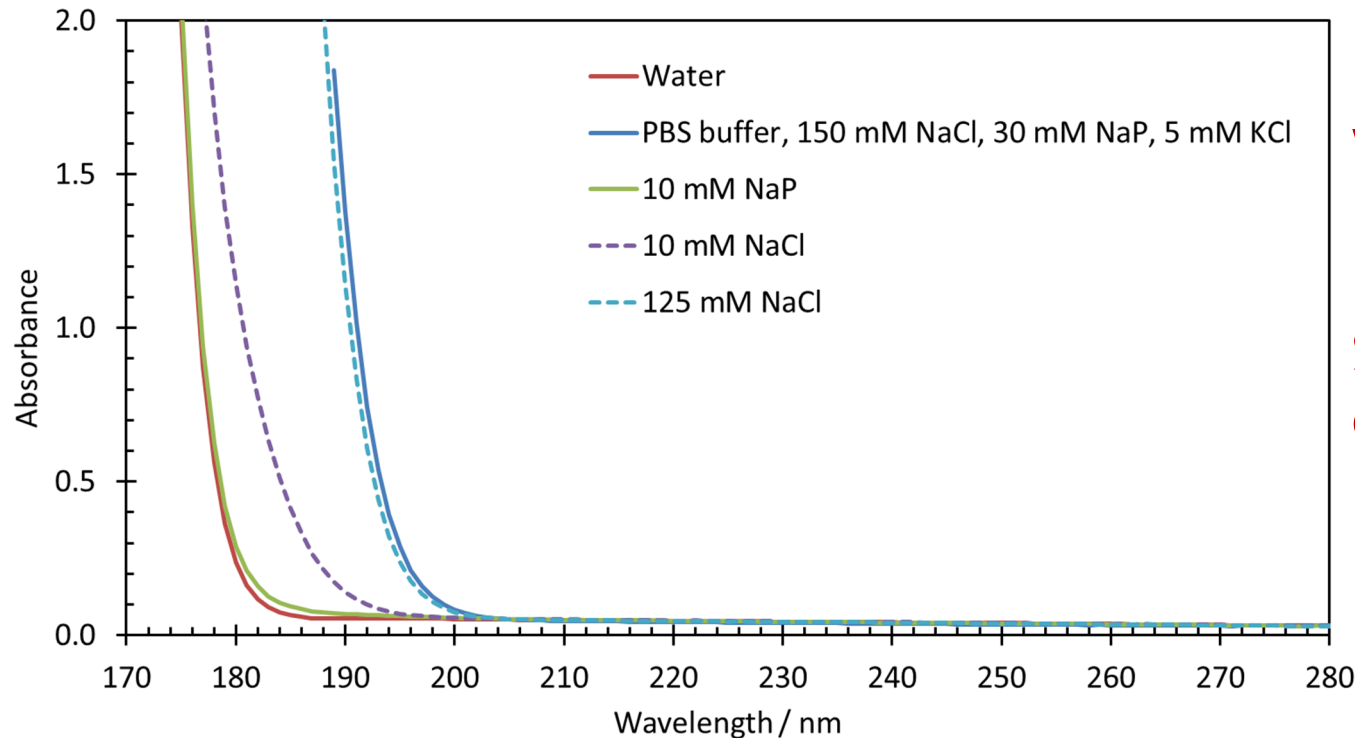
We like to use  
phosphate buffers

But PBS has a high  
absorbance below  
200 nm

Why...?

# Sample buffers

What is the best choice for buffer for *far-UV* CD



We like to use  
phosphate buffers

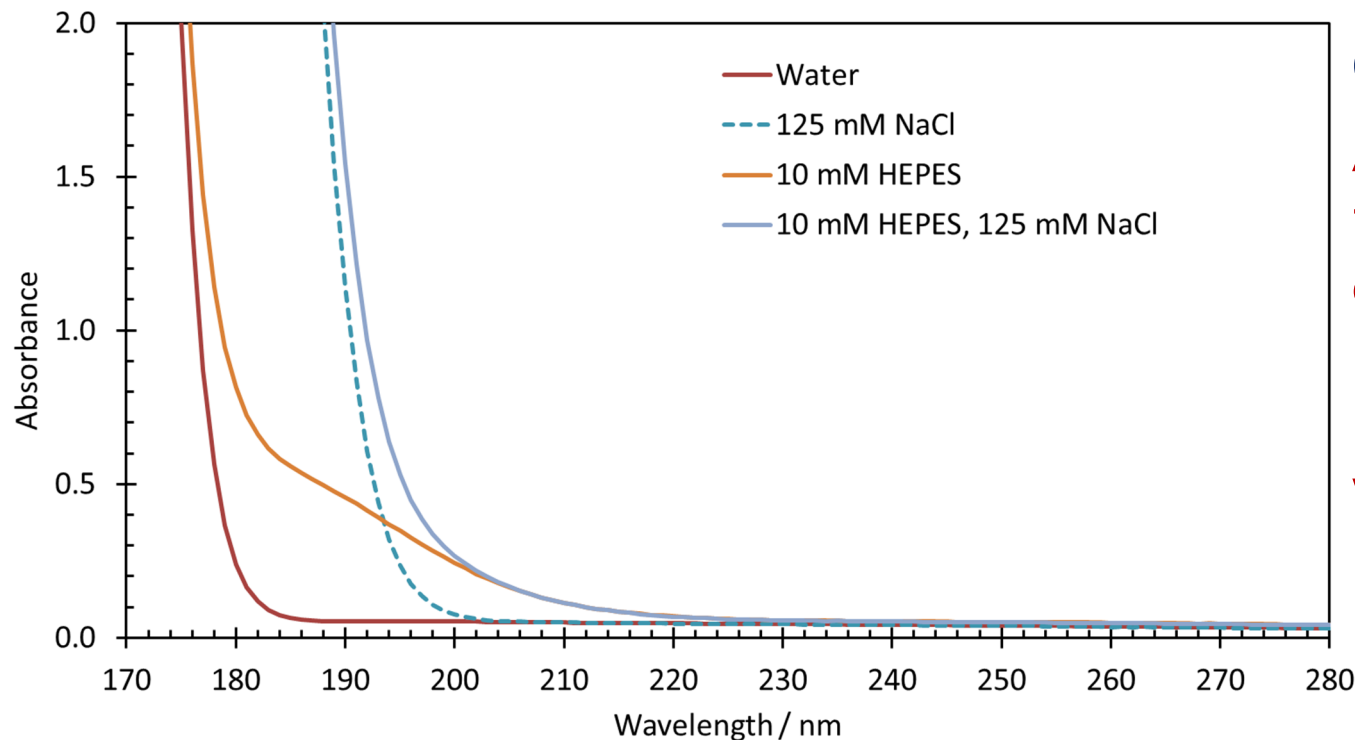
Sodium Phosphate  
only absorbs a little.

➤ *NaCl is the issue*

Try to avoid anything with  $\text{Cl}^-$  ions. E.g. use NaF instead of NaCl

# Sample buffers

What is the best choice for buffer for *far-UV* CD



Other buffers?

As an example, HEPES is not very good either

Especially when used with NaCl

More tomorrow: Hands on “*Buffer limitations study*”