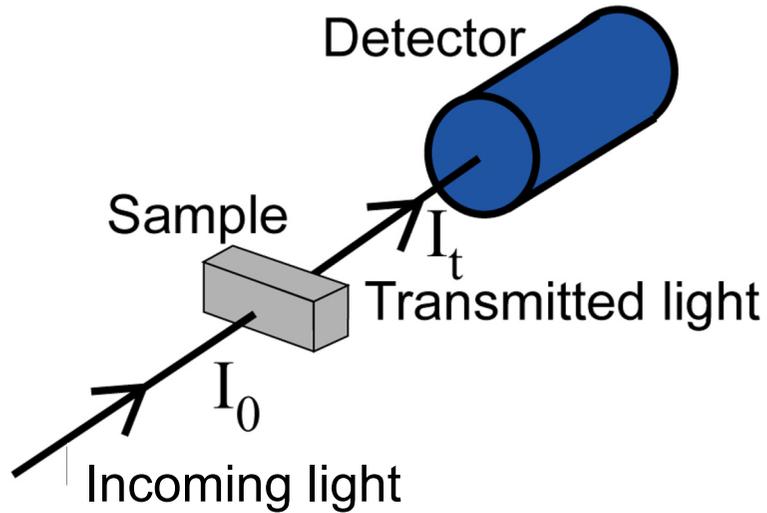


ESC1: Circular Dichroism: best practice and data analysis

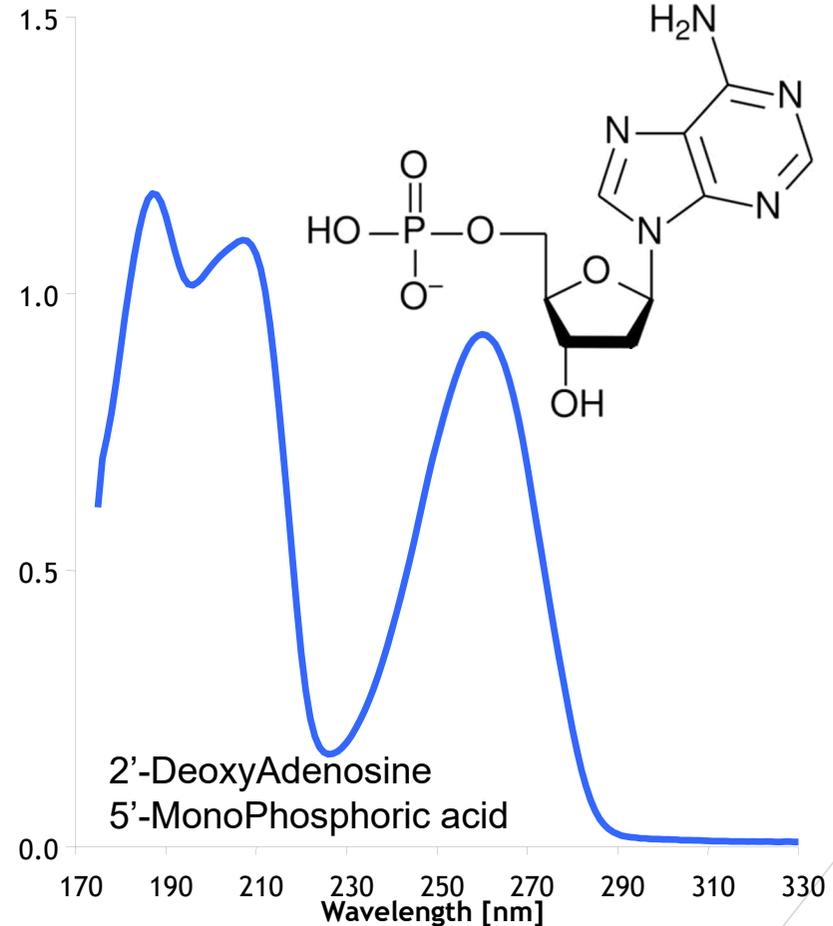
Lecture 1: Introduction to CD spectroscopy: Principles and information content

Absorption Spectroscopy

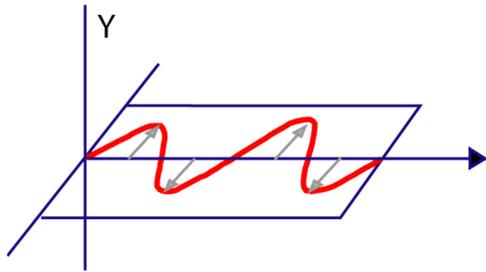


Absorbance:

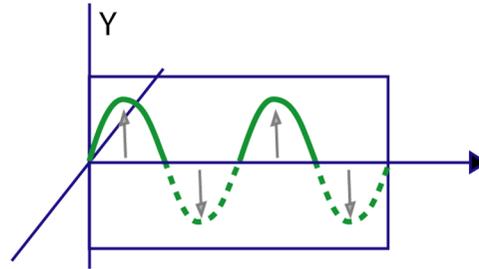
$$A = \log(I_0/I_t)$$



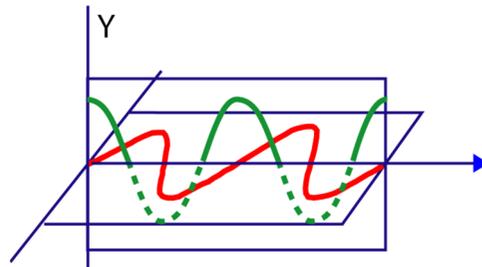
Polarized light



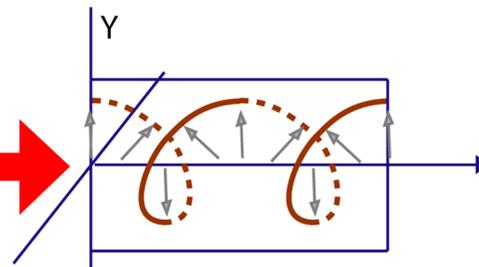
Horizontally pol.



Vertically pol.



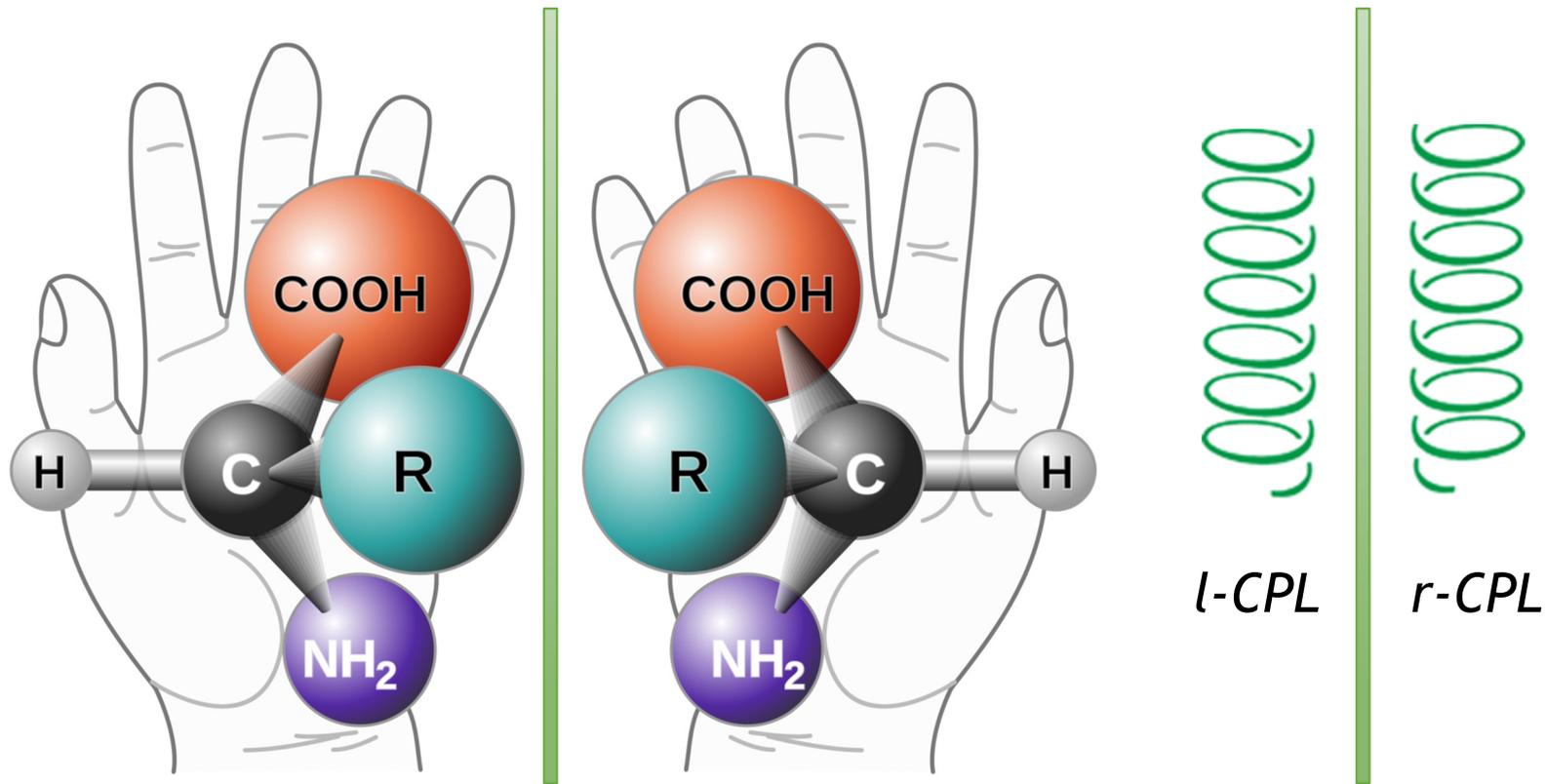
Sum of two
plane pol.



Circularly pol.



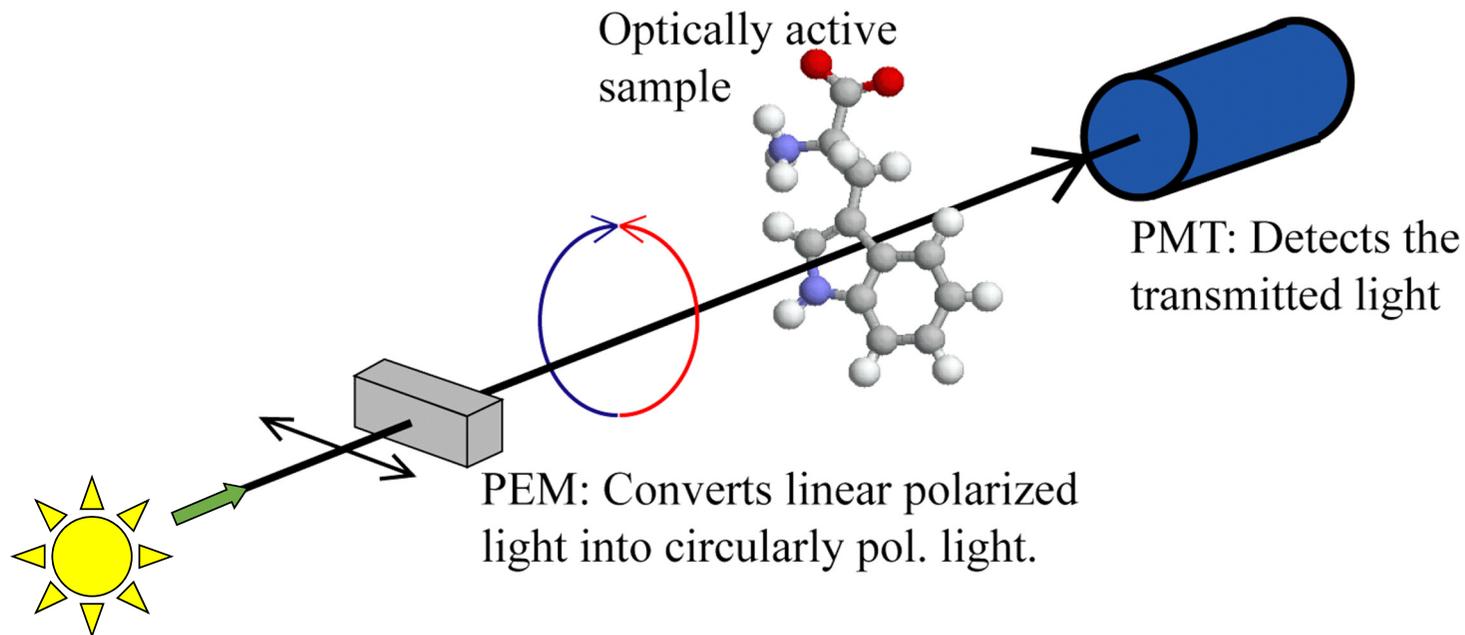
Chirality



Mirror images



Circular Dichroism



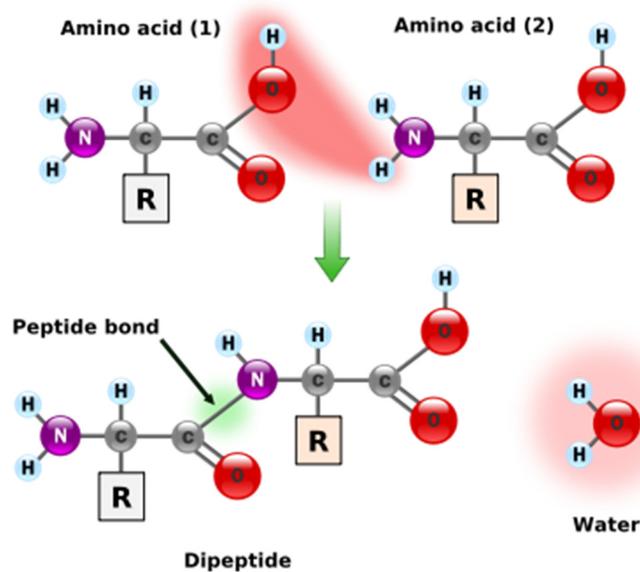
$$\text{The CD signal: } CD = A_L - A_R$$



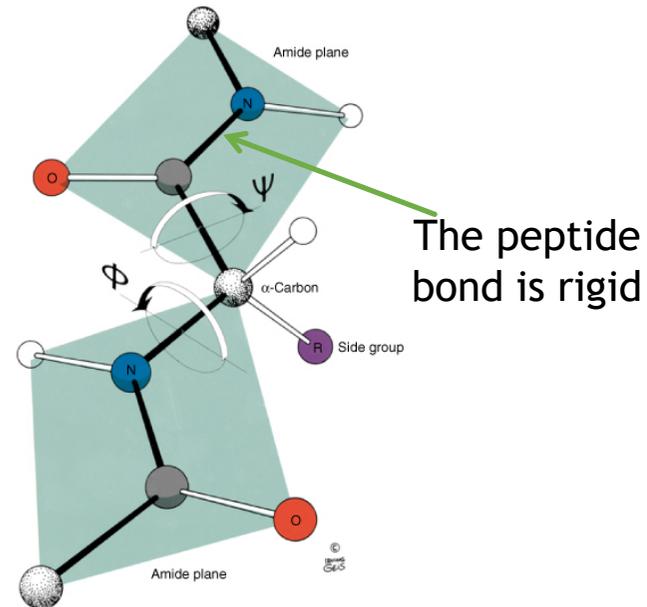
Protein structure

The peptide bond

Amino acids form polymers with a peptide bond between them



The peptide bond is formed between the carboxyl and the amine groups

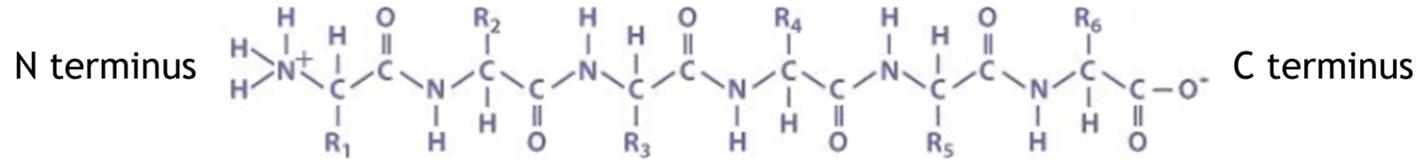


Phi (ϕ) and Psi (ψ) angles define the secondary structure

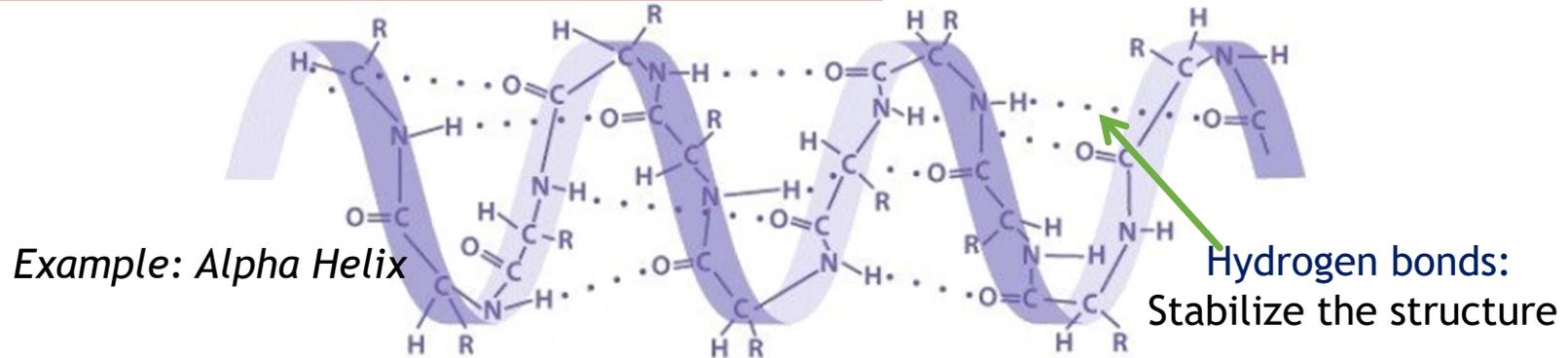


Protein structure: Levels of structures

Primary structure (amino acid sequence):

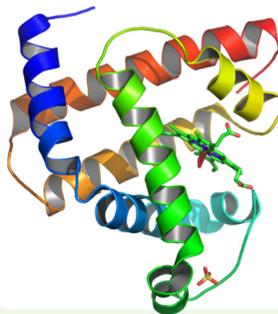


Secondary structure (Phi (φ) and Psi (ψ) angles):



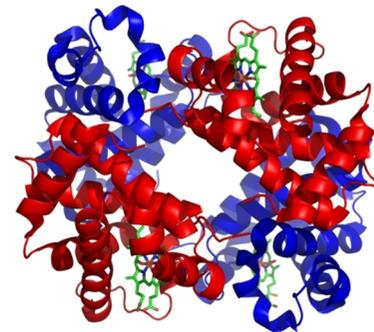
Tertiary structure:

Relative organization
of secondary
structures

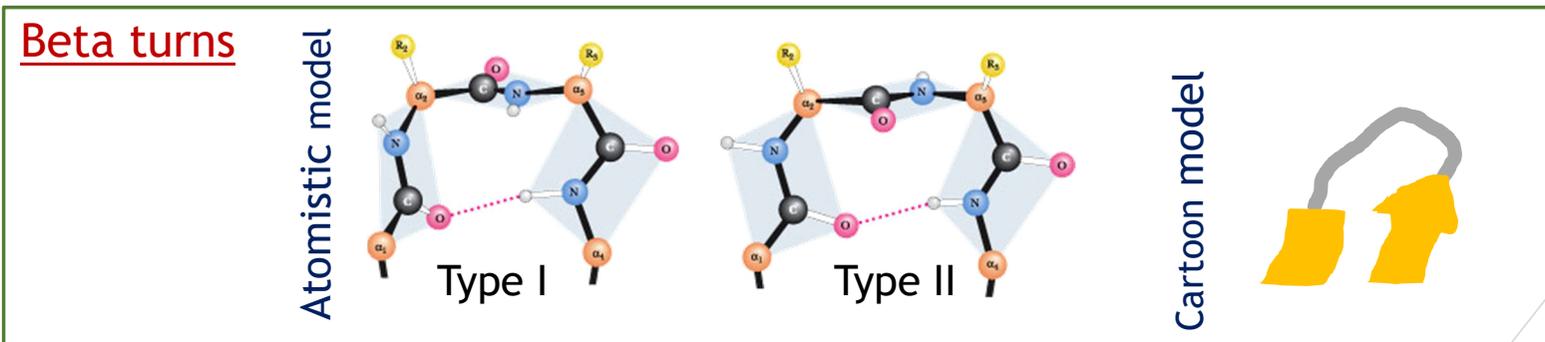
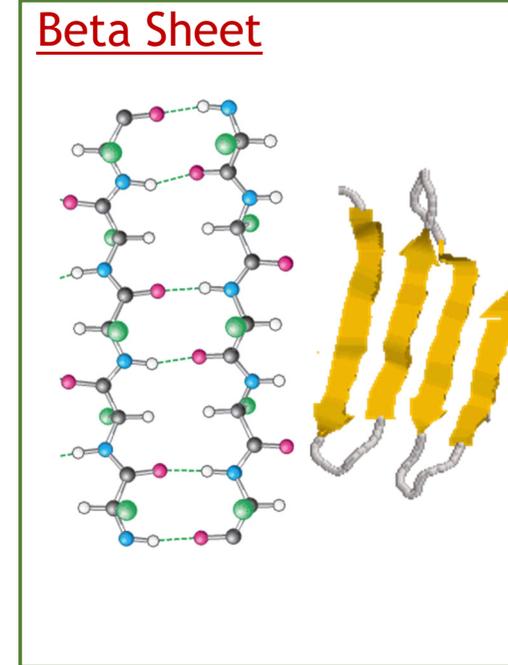
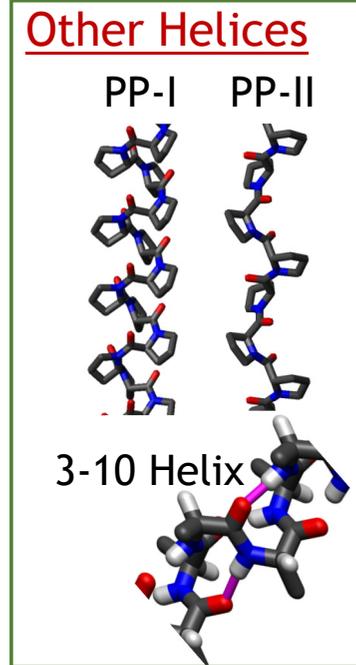
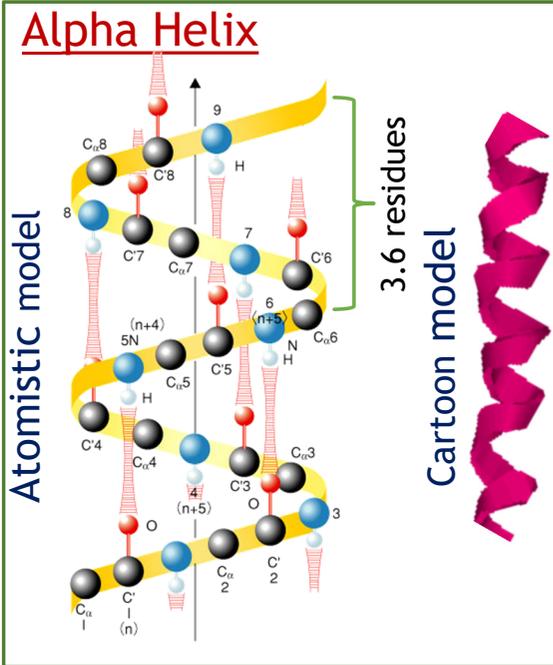


Quaternary structure:

Relative organization
of tertiary structures



Protein structure: Secondary structures



Protein structure: Secondary structures

Circular dichroism can give secondary structure information
from a large range of proteins/peptides

Very simple



Glucagon (Hormone)
 $M_w \sim 3485$ Daltons
29 amino acids



Pepsin (Enzyme)
 $M_w \sim 35.000$ Daltons
 ~ 320 amino acids

Very complicated

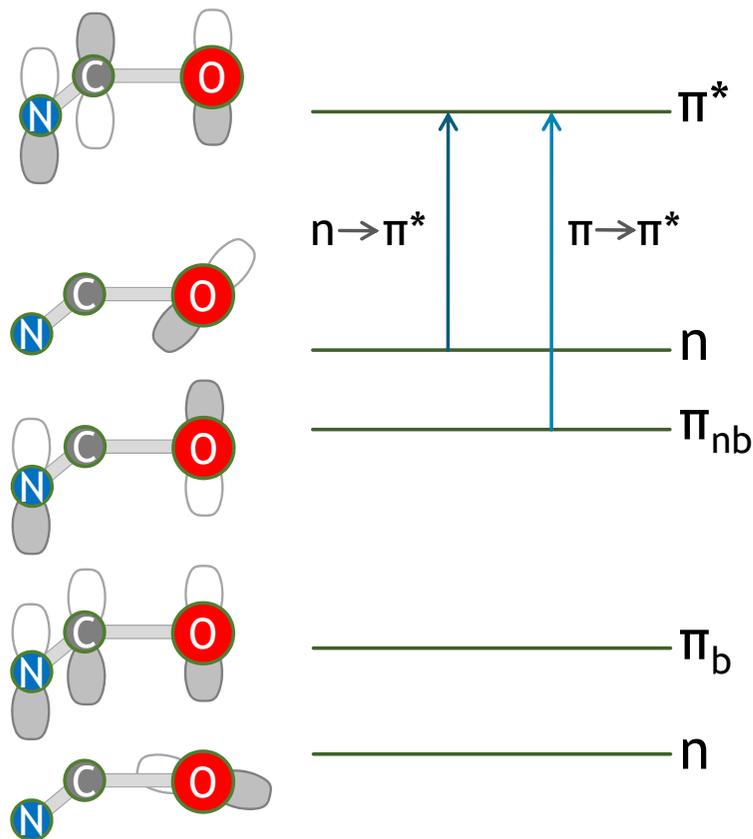


Antibody
 $M_w \sim 150.000$ Daltons
 ~ 1.400 amino acids



Protein *electronic* structure

The backbone of the polypeptide has important chromophores



The $n \rightarrow \pi^*$ and the $\pi \rightarrow \pi^*$ have transitions in the *far UV spectral range*

$n \rightarrow \pi^* \sim 220 \text{ nm}$

$\pi \rightarrow \pi^* \sim 190 \text{ nm}$

These transitions are sensitive to the φ and ψ angles

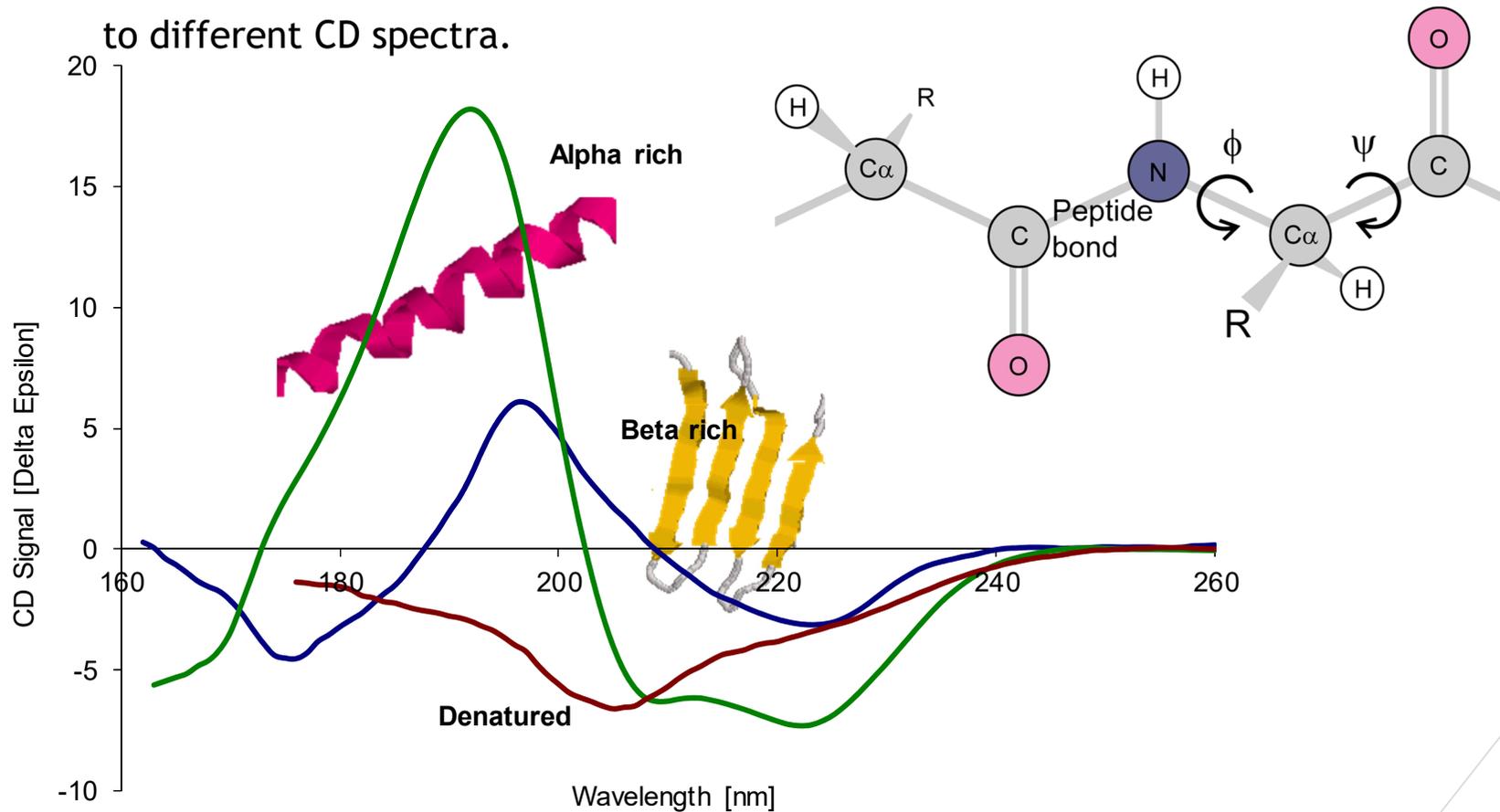
i.e. sensitive to the *Secondary Structure*

Obviously important for Circular Dichroism

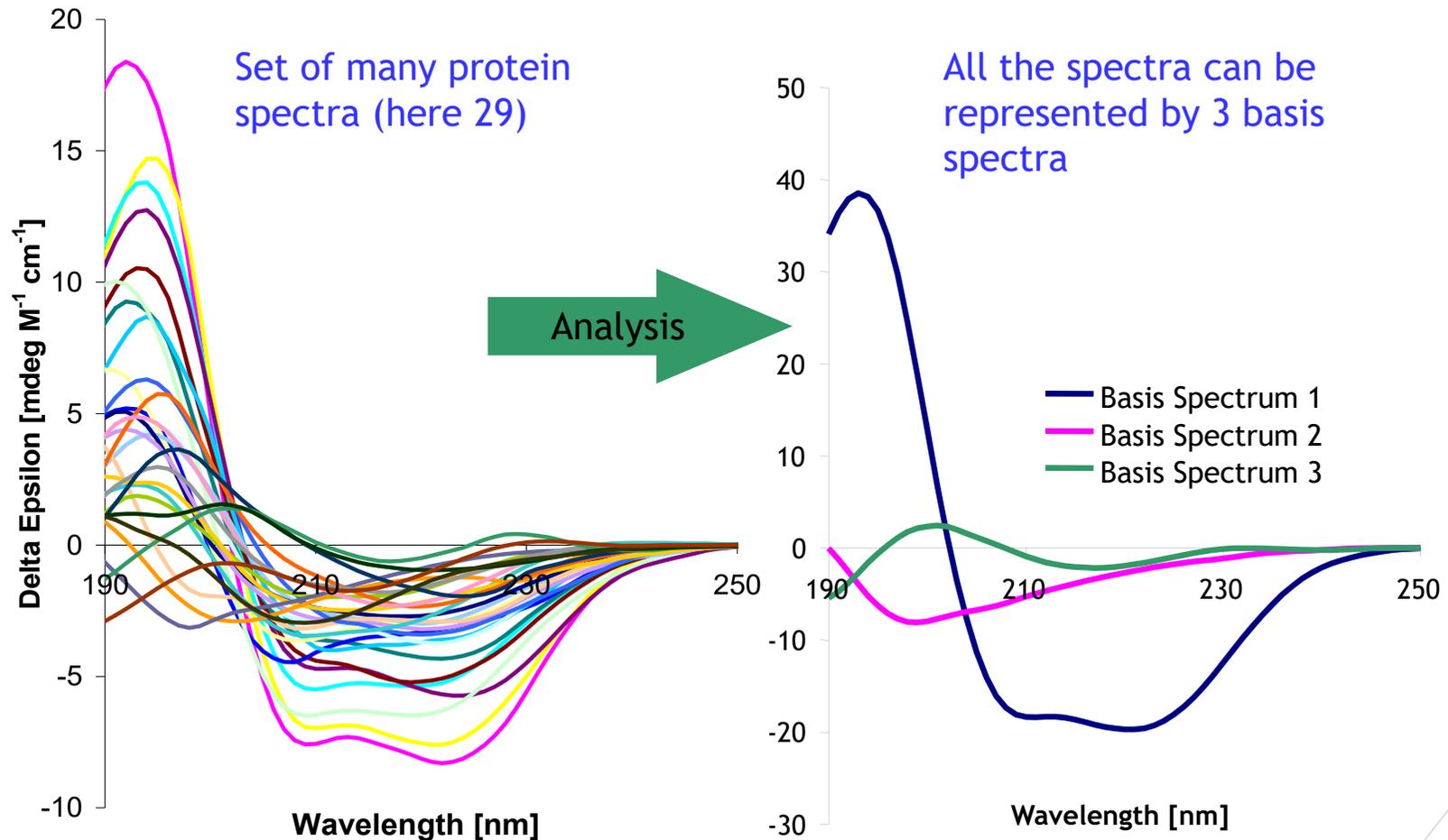


Information in a protein CD spectrum

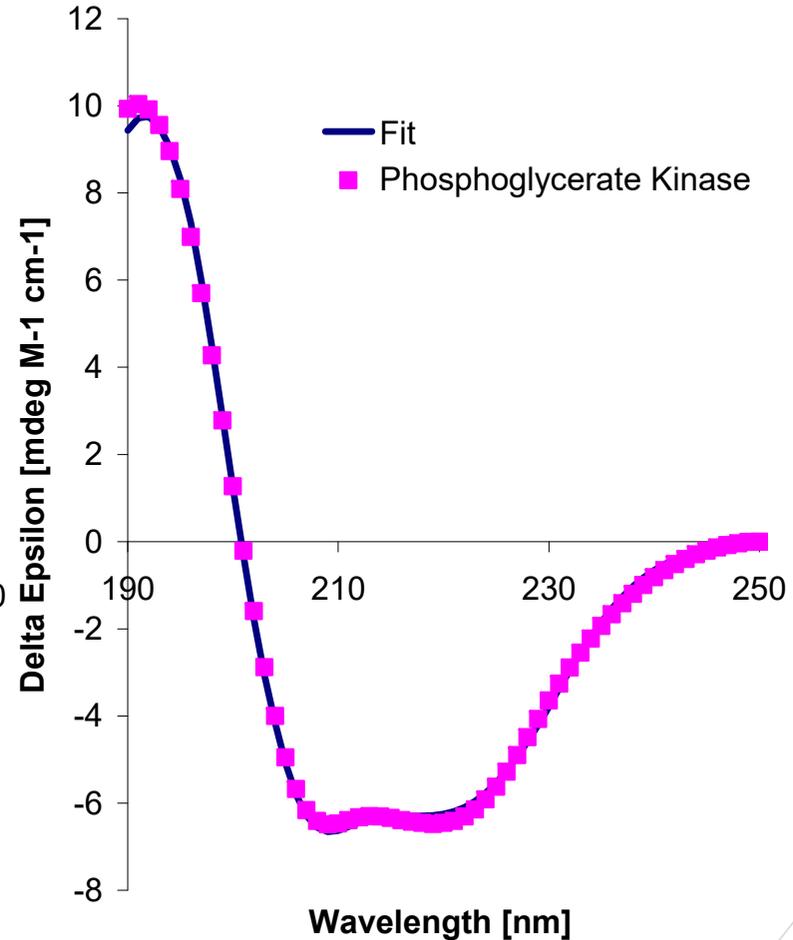
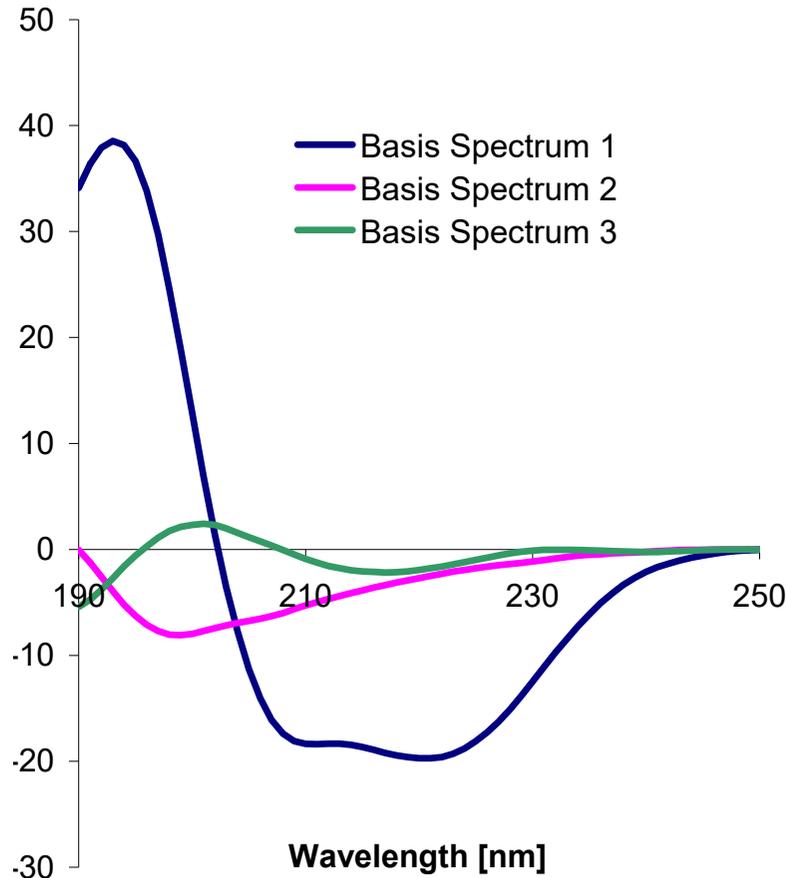
Different secondary structure in proteins gives rise to different CD spectra.



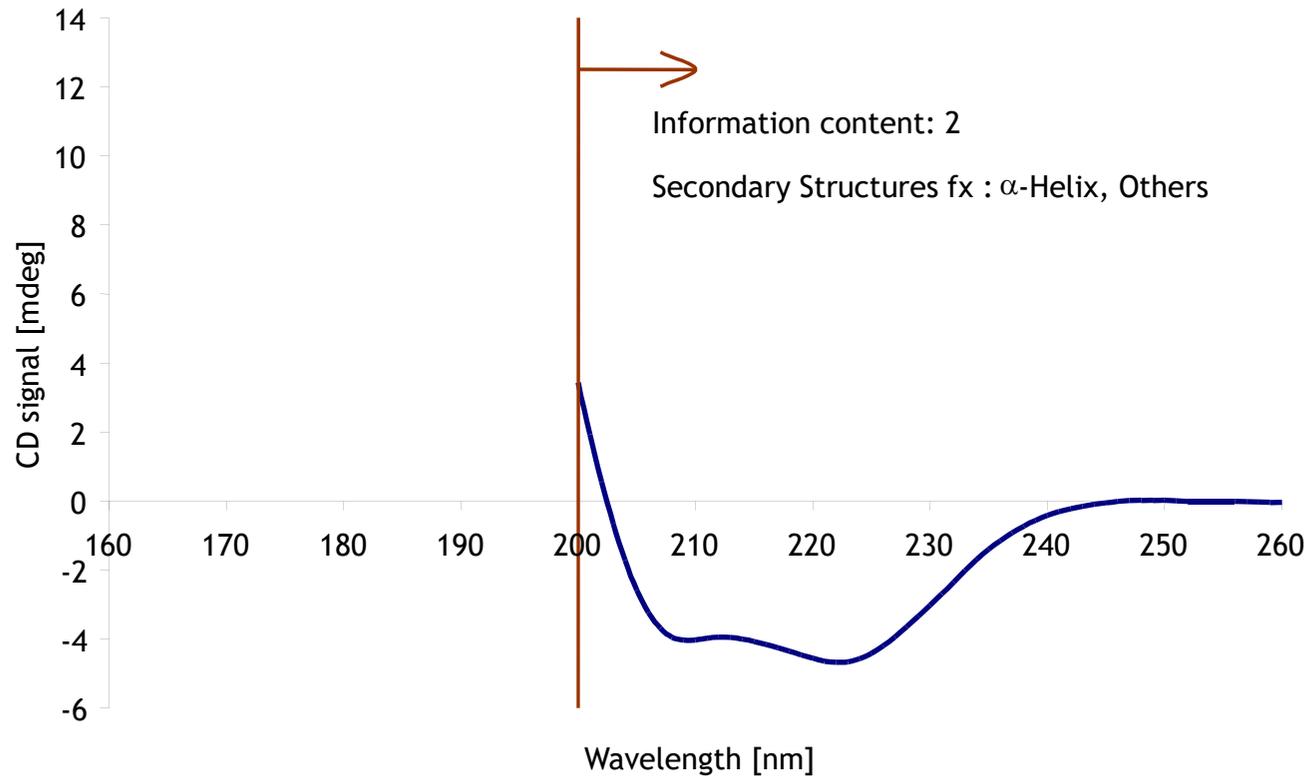
Information in a protein CD spectrum



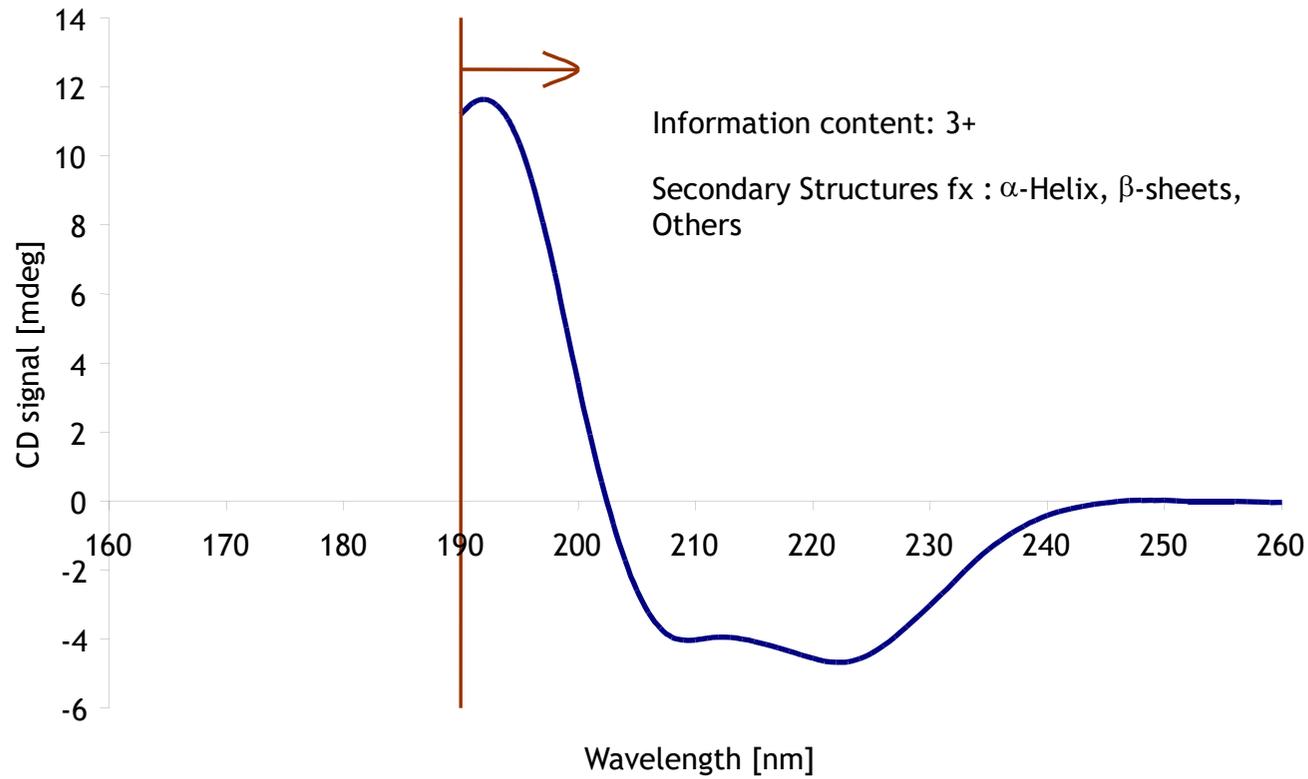
All spectra may be reconstructed from the three basis spectra



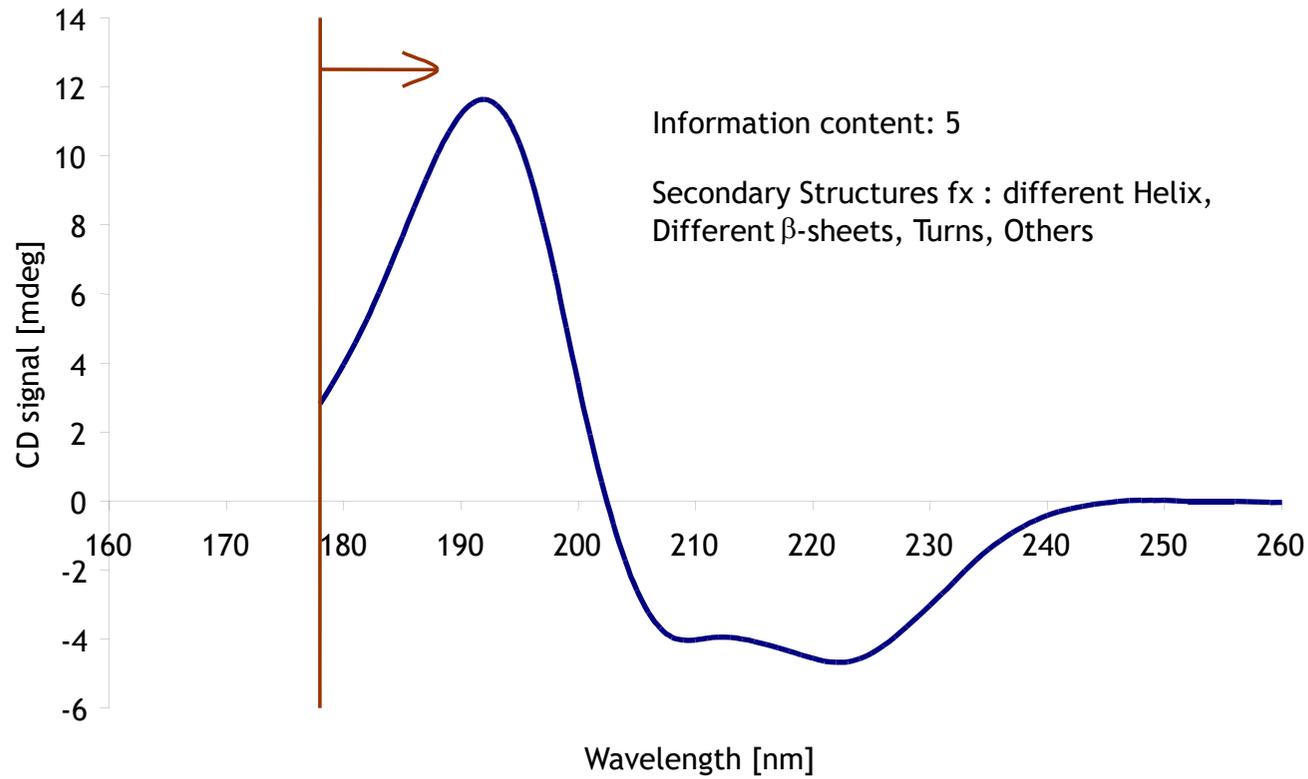
Information in a protein CD spectrum



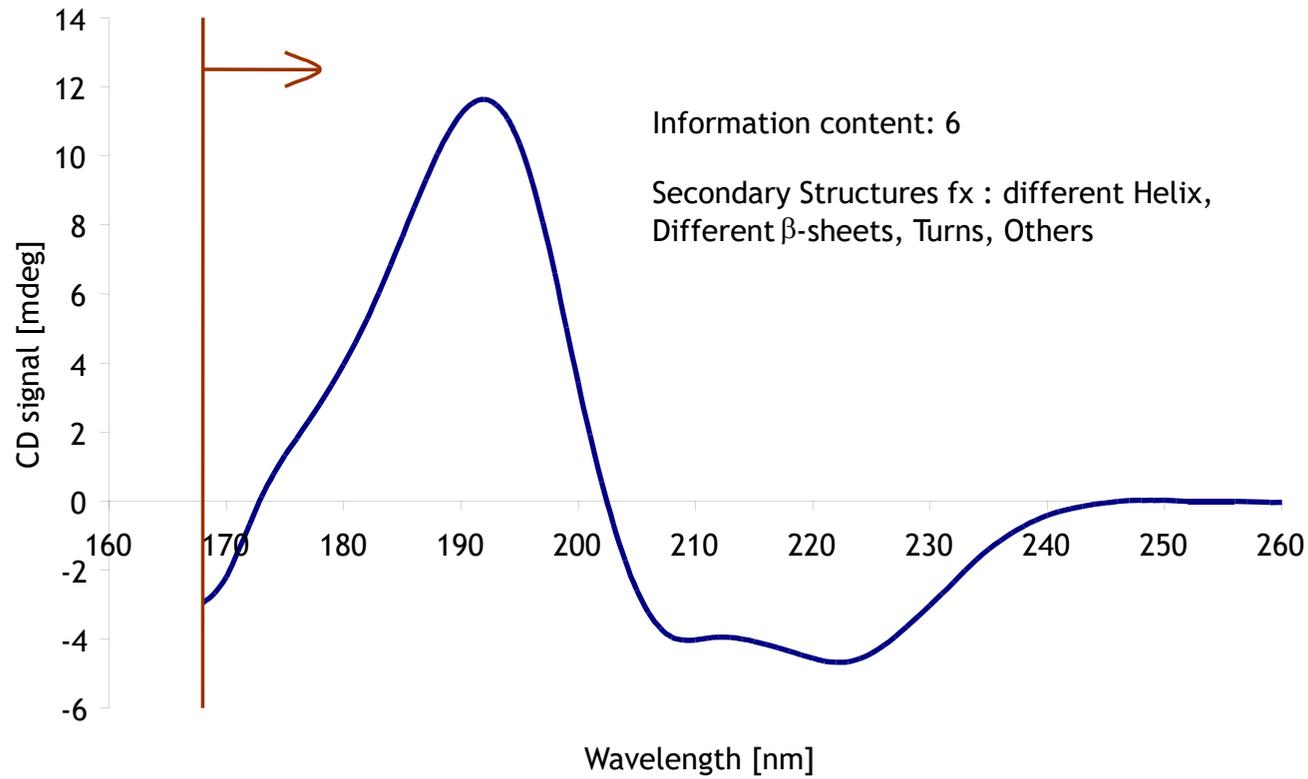
Information in a protein CD spectrum



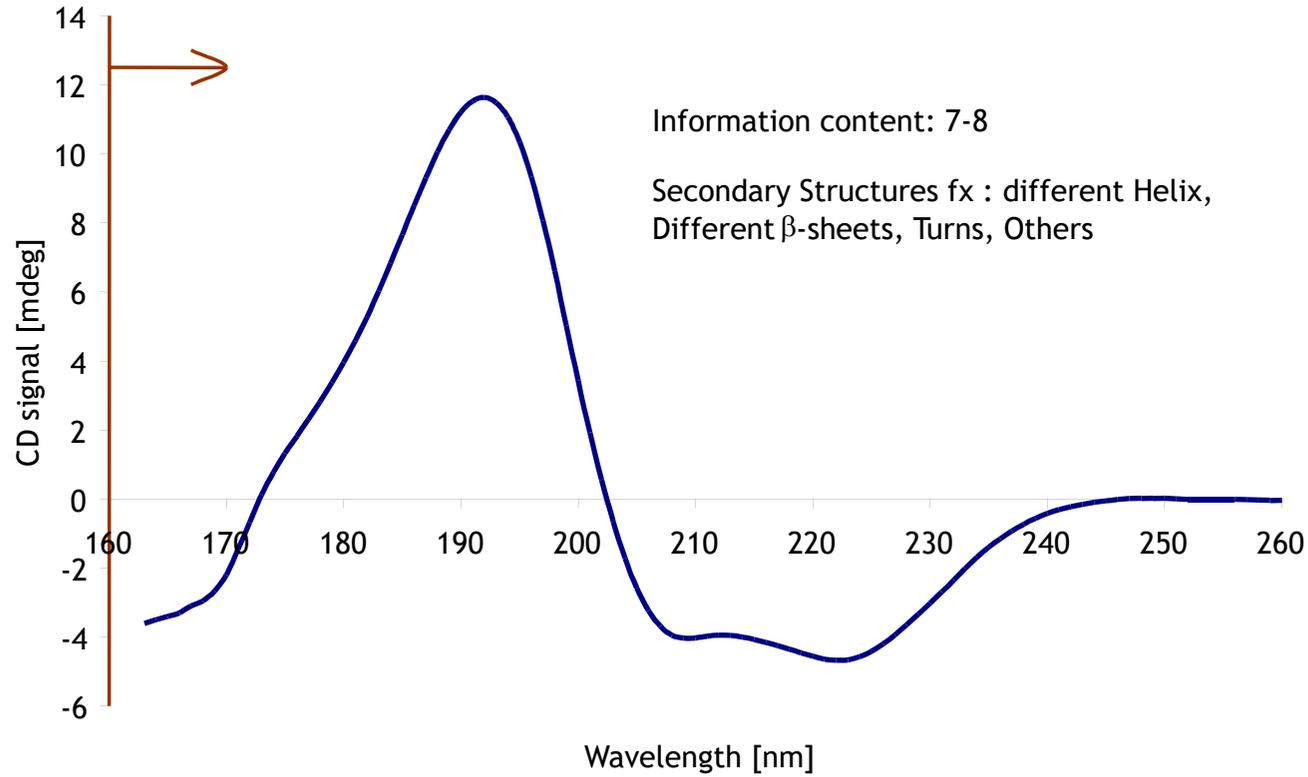
Information in a protein CD spectrum



Information in a protein CD spectrum



Information in a protein CD spectrum



CD units

CD units:

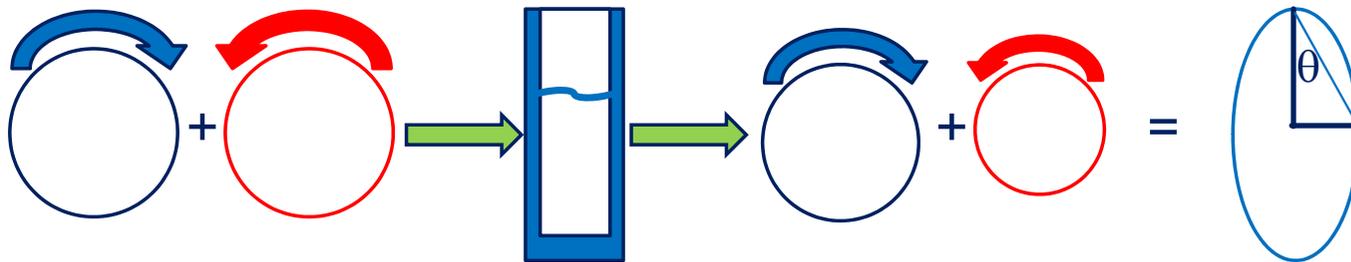
$$A = \log (I_0/I_t)$$

Typically $\sim 0.1 - 1.5$

$$CD = \Delta A = A_L - A_R$$

Typically $\sim 10^{-4} - 10^{-3}$

Historically CD measured as an angle (deg. or mdeg.)



$$\theta[\text{mdeg}] = 32980 \Delta A$$

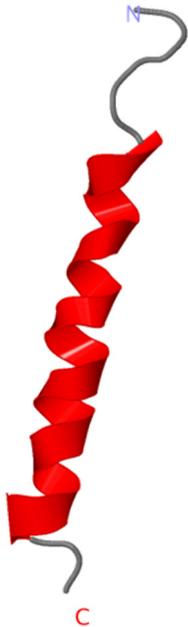
θ is typically 1-100 mdeg



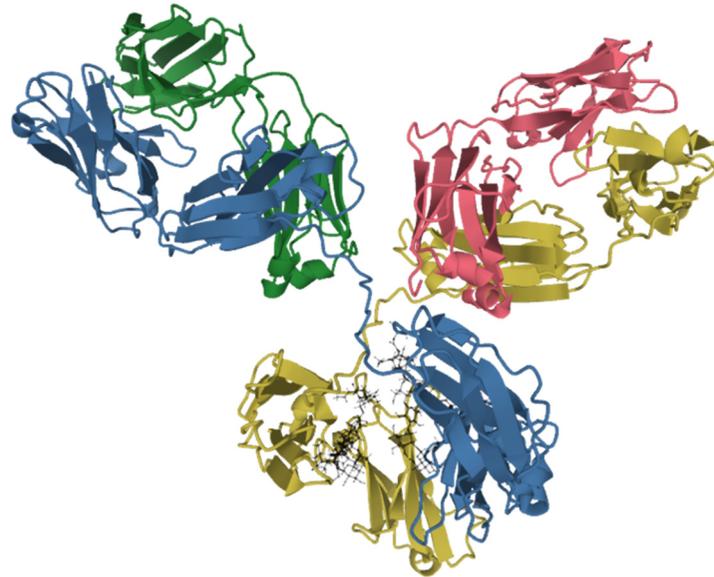
CD spectrum: compare signals

CD units: $CD = \Delta A = A_L - A_R$

Compare CD between proteins:



Short peptide
Vs
Large protein

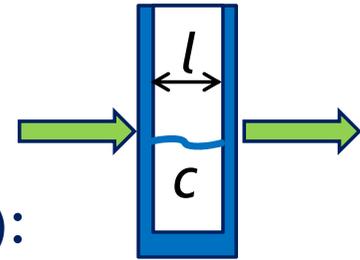


CD spectrum: compare signals

CD units: $CD = \Delta A = A_L - A_R$

Compare CD between proteins:

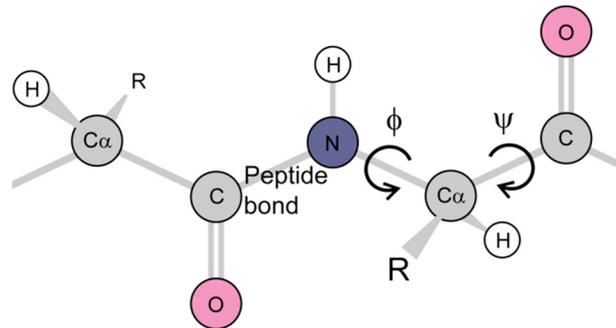
Correct for pathlength (l) and concentration (c):



$$\Delta \varepsilon = \Delta A / (l \cdot c)$$

Unit $M^{-1} \text{ cm}^{-1}$

CD signal
originates from
each amino acid



 Concentration is per residue/amino acid

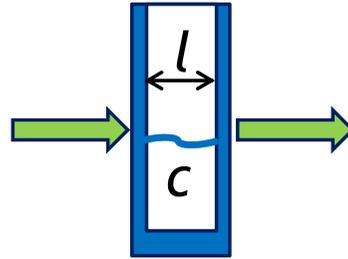


CD units

CD units:

$$\Delta\varepsilon = \Delta A / (l \cdot c)$$

Unit $M^{-1} \text{ cm}^{-1}$



CD signal originates from each amino acid

→ Concentration is per residue/amino acid

MRW: Mean Residual Weight (~110 Da)

Average weight of the amino acids in the protein

c measured in mg/mL (g/L) **→** c/MRW is 'mol residues per L'

$$\Delta\varepsilon [M^{-1} \text{ cm}^{-1}] = \text{CD}[\text{mdeg}] \cdot \text{MRW}[\text{Da}] / (32980 \cdot l[\text{cm}] \cdot C[\text{g/L}])$$



How to get the concentration?

There are many ways used to get the concentration of proteins

- Quantitative Amino Acid Analysis
- Bradford Assay: Coomassie blue G-250 assay reagent, 595 nm
- BCA Protein Assay Kits: reduction of Cu^{2+} to Cu^{1+} by protein in an alkaline medium, 562 nm
- Protein absorbance 280 nm:
 - Concentration (mg/ml) = Absorbance at 280 nm divided by path length (cm)



This just makes me sad



- **Better** use calculated ϵ_{280} from primary sequence

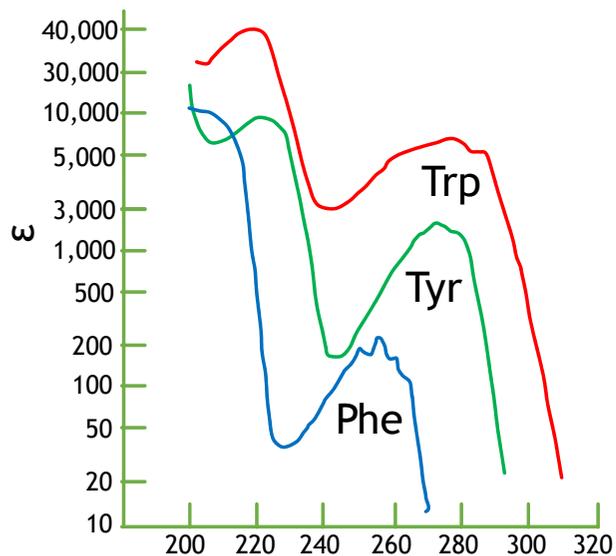
$$c = A / (l \cdot \epsilon_{280}), \text{ Unit for } \epsilon_{280} \text{ M}^{-1} \text{ cm}^{-1}$$



How to get the concentration?

Protein absorbance 280 nm: $c = A / (l \cdot \epsilon_{280})$, Unit for ϵ_{280} $M^{-1} \text{ cm}^{-1}$

The 280 nm protein absorbance in protein originates mostly from Tryptophan and Tyrosine



Known primary sequence

$$\epsilon_{280} = 5690 \cdot n_{\text{Trp}} + 1280 \cdot n_{\text{Tyr}}$$

Absorbance at 280 nm is much lower than in the far-UV range (below 240 nm)

- Often necessary to use different pathlengths (l) for 280 nm and CD measurements



How to get the concentration?

Protein absorbance 205 nm: $c = A / (l \cdot \epsilon_{205})$, Unit for ϵ_{280} $M^{-1} \text{ cm}^{-1}$

This is our method of choice for CD measurements

N.J. Anthis and G.M. Clore. *Protein Science* **22** (2013) 851-858

Side chain/feature	ϵ_{205} ($M^{-1} \cdot \text{cm}^{-1}$)
Tryptophan	20,400
Phenylalanine	8600
Tyrosine	6080
Histidine	5200
Methionine	1830
Arginine	1350
Cysteine	690
Asparagine ^a	400
Glutamine ^a	400
Cystine ^b	2200
Backbone peptide bond ^c	2780 ± 168

^bIf the protein has a disulfide bond, add 820 to its ϵ_{205}

Calculate the ϵ_{205}

$$\epsilon_{205} = \sum_{aa} (n_{aa} \cdot \epsilon_{aa}) + n_{S-S} \cdot 820 + (\text{Seq. length} - 1) \cdot \epsilon_{backbone}$$

Sum over amino acids in table

Disulfide bonds

n_{aa} : number of type *aa* amino acid

Seq. length : number of all amino acids

The absorbance is measured simultaneously with CD on the same exact sample !!!



How to get the concentration?

Protein absorbance 205 nm: $c = A / (l \cdot \epsilon_{205})$, Unit for ϵ_{280} $M^{-1} \text{ cm}^{-1}$

Primary sequence of Myoglobin:

MGLSDGEWQQVLNVWGKVEADIAGHGQEVLRIRLFTGHPETLEKFDK
FKHLKTEAEMKASEDLKKHGTVVLTALGGILKKKGHHEAELKPLAQSH
ATKHKIPIKYLEFISDAIIHVLHSHKHPGDFGADAQGAMTKALELFRNDIA
AKYKELGFQG

Amino Acids

Code	n_{aa}	ϵ_i
F	7	8600
M	3	1830
C	0	690
Y	2	6080
W	2	20400
Q	6	400
N	2	400
H	11	5200
R	2	1350
Disulfide bonds	0	820
Sequence length	154	
Seq. length - 1	153	2780

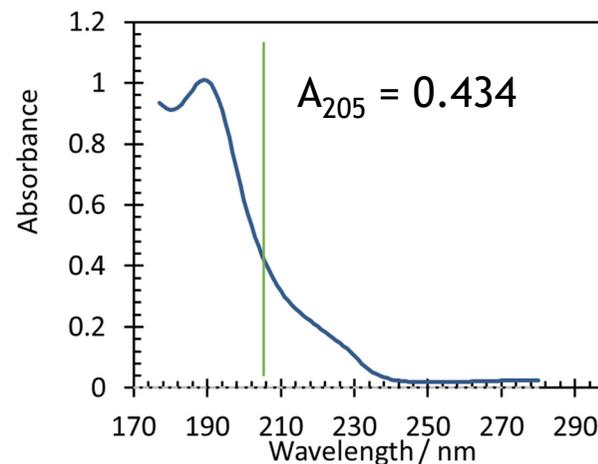
$$\epsilon_{205} = \sum_{aa} (n_{aa} \cdot \epsilon_{aa}) + n_{S-S} \cdot 820 + (\text{Seq. length} - 1) \cdot \epsilon_{backbone}$$

For the entire protein:

$$\epsilon_{205} = 607\,090 \text{ M}^{-1} \text{ cm}^{-1}$$

Per residue:

$$\epsilon_{205} = 3942 \text{ M}^{-1} \text{ cm}^{-1}$$



$$c \cdot l = A / \epsilon_{205} = 0.434 / 3942 = 0.11 \text{ mM} \cdot \text{cm}$$



How to get the concentration?

What if I want the concentration in mg/ml ?

Primary sequence of Myoglobin:

MGLSDGEWQQVLNVWGKVEADIAGHGQEVLRIRLFTGHPETLEKFDKFKHLKTEAEMKASEDLKKHGTVVLTALGGILKKKGHHEALKPLAQSHATKHKIPIKYLEFISDAIIHVLHSHKHPGDFGADAQGA
MTKALELFRNDIAAKYKELGFQG

$$MW_{protein} = \sum_{aa} (n_{aa} \cdot MW_{aa}) - (Seq. length - 1) \cdot MW_{water}$$

Amino Acids			
Code	n_{aa}	MW_{aa}	
I	9	113.2	Y
L	17	113.2	W
V	7	99.1	Q
F	7	147.2	N
M	3	131.2	H
C	0	103.1	E
A	15	71.1	D
G	15	57.1	K
P	4	97.1	R
T	7	101.1	Sequence length
S	5	87.1	Water

Molecular weight

$$MW = 17082.6 \text{ Da}$$

Mean Residual Weight

$$MRW = 110.93 \text{ Da}$$

Using:

$$c \cdot l = 0.11 \text{ mM} \cdot \text{cm}, l = 0.01 \text{ cm}$$

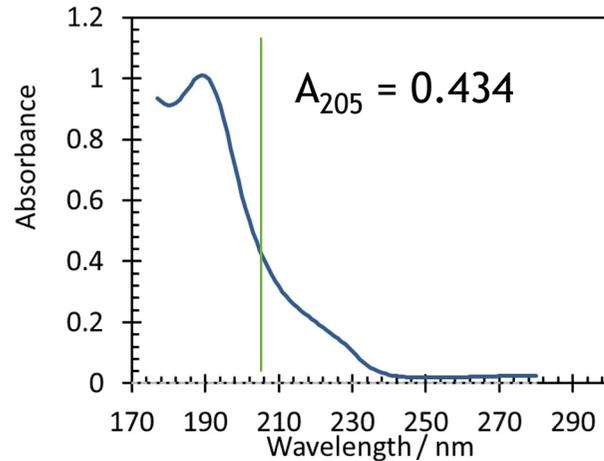
We get:

$$\begin{aligned} c[\text{mg/ml}] &= 110.93 \text{ g/mol} * 0.11 \text{ mM} \cdot \text{cm} / 0.01 \text{ cm} \\ &= \underline{1.2 \text{ mg/ml}} \end{aligned}$$



What concentration to use?

Protein absorbance 205 nm and below:



$$c \cdot l = A / \epsilon_{205} = 0.434 / 3942 = 0.11 \text{ mM} \cdot \text{cm}$$

$$\begin{aligned} c[\text{mg/ml}] &= 110.93 \text{ g/mol} * 0.11 \text{ mM} \cdot \text{cm} / 0.01 \text{ cm} \\ &= \underline{1.2 \text{ mg/ml}} \end{aligned}$$

Measured in a 0.01 cm = 0.1 mm pathlength cell

Notice that for this protein the absorbance peaks at ~1 near 190 nm

➤ This is a good absorbance to optimize signal to noise, and thus a good concentration

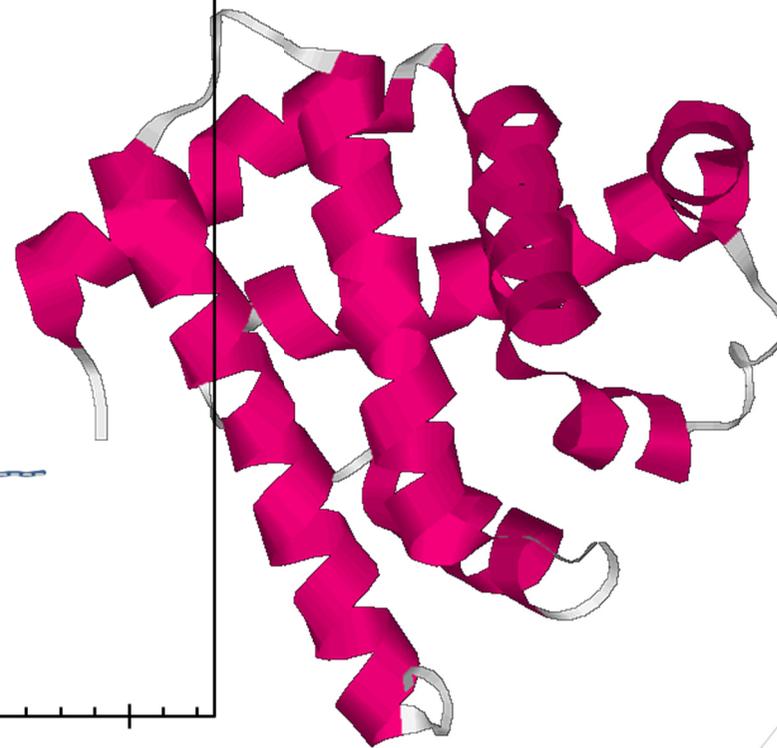
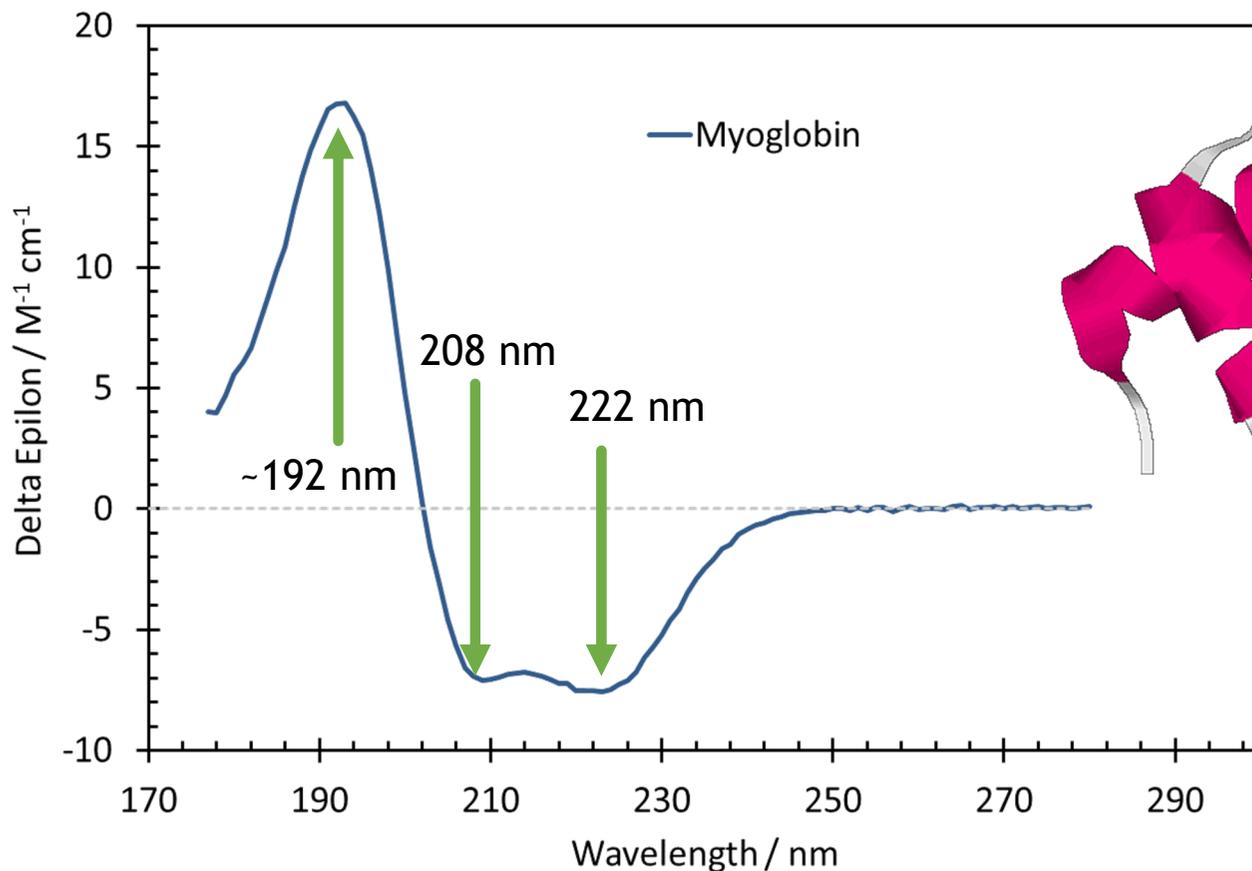
Roughly: 1 mg/ml protein in a 0.1 mm cell

- Lower for highly alpha helical proteins
- Higher for highly beta sheet or unordered proteins
- Scale concentration with pathlength of the cell you are using



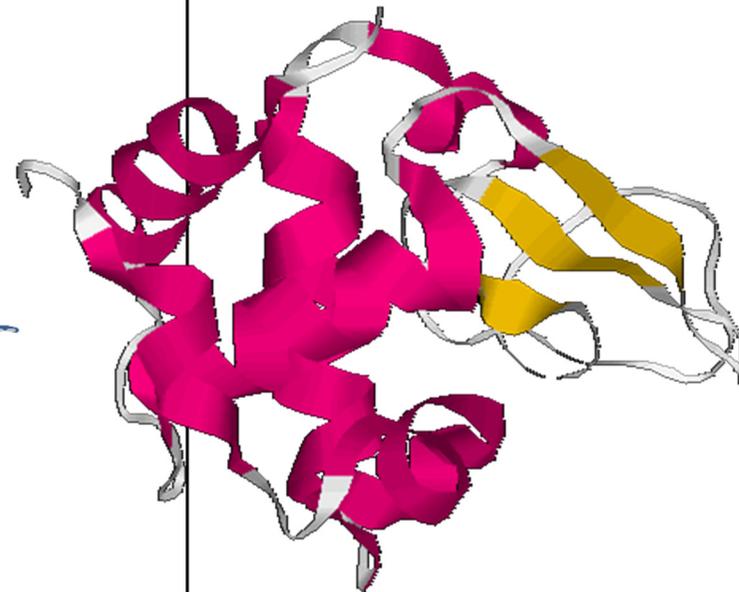
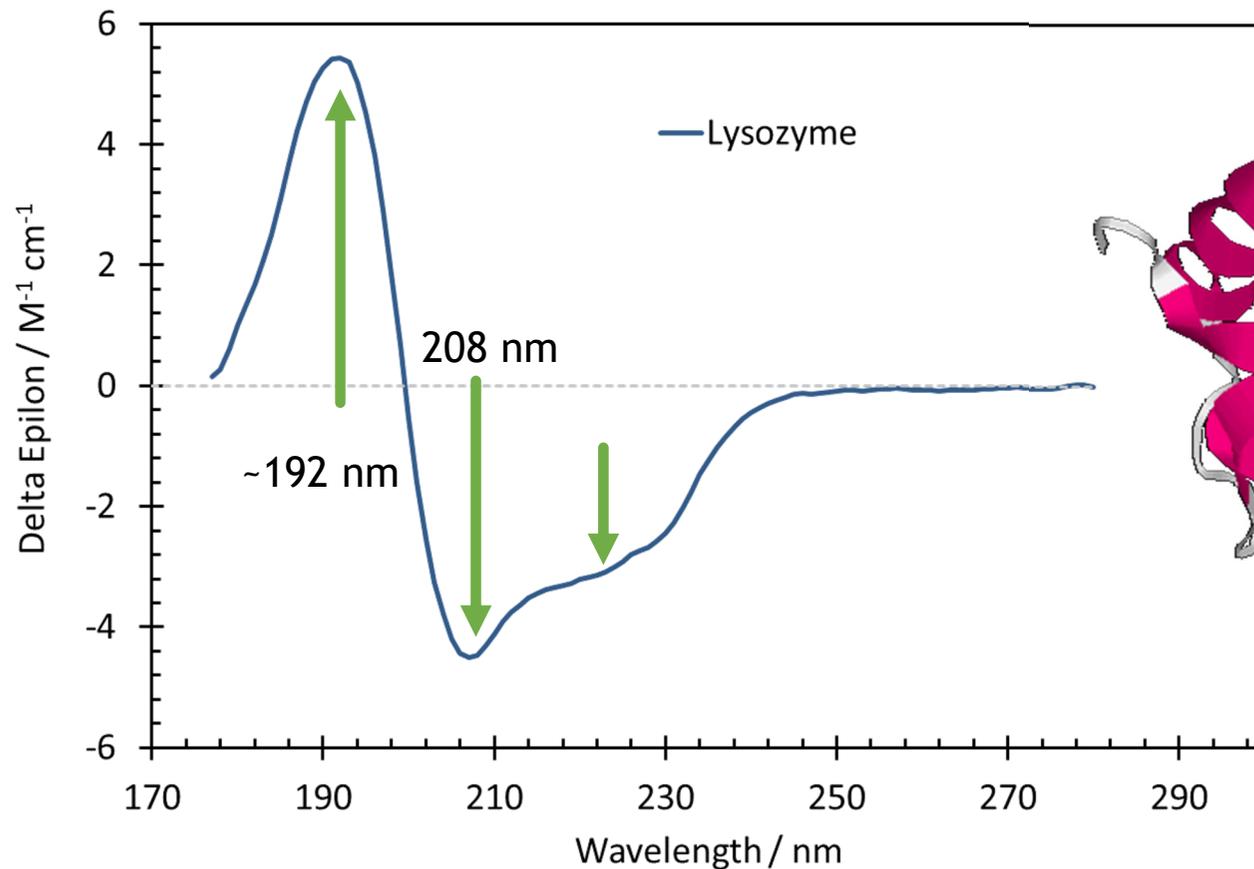
Getting acquainted with protein CD spectra

Myoglobin *A typical Alpha Helical protein*



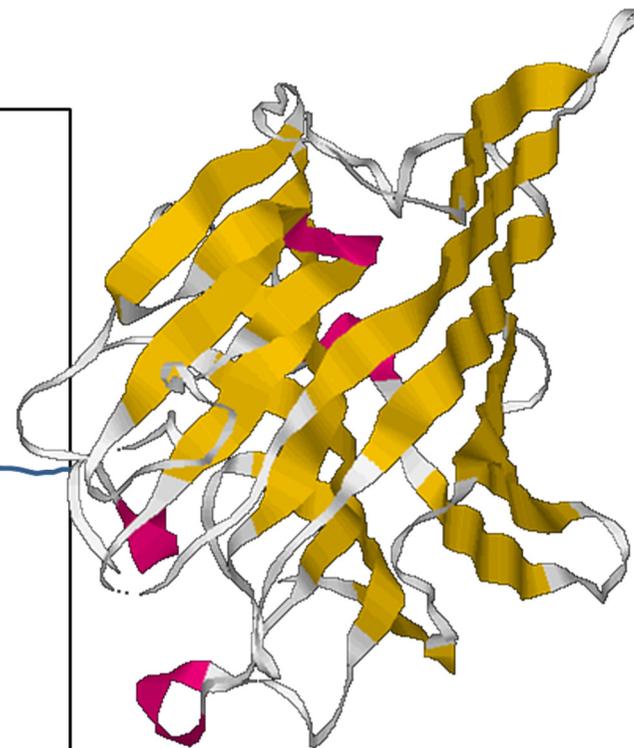
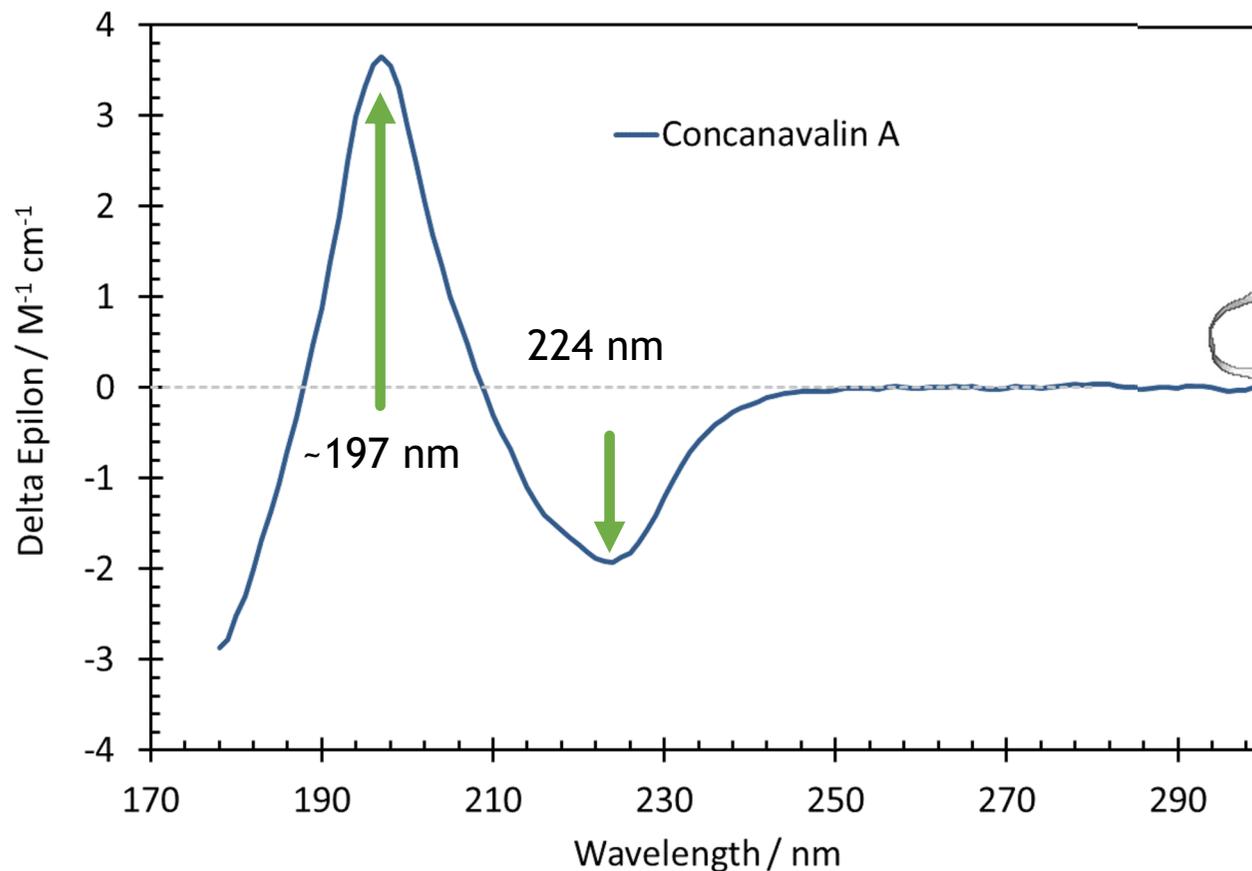
Getting acquainted with protein CD spectra

Lysozyme *A Alpha Helical and Beta sheet mix protein*



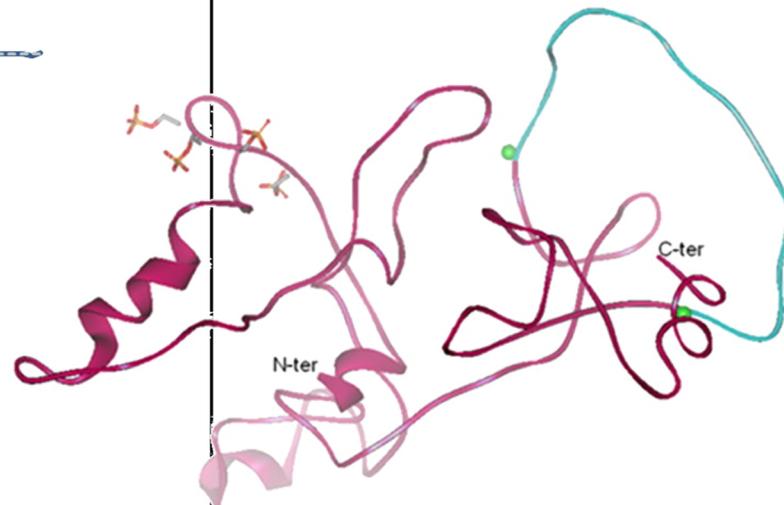
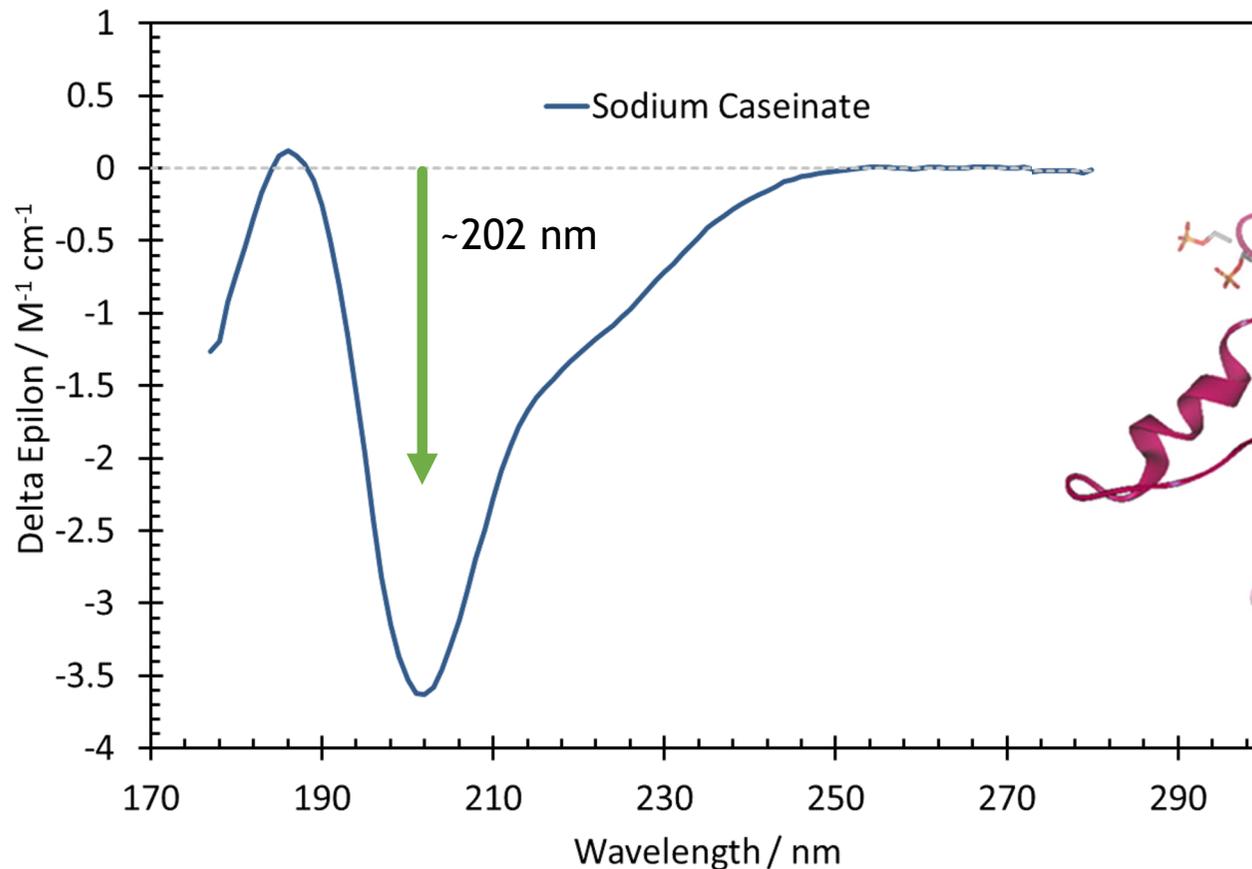
Getting acquainted with protein CD spectra

Concanavalin A *A Beta sheet protein*



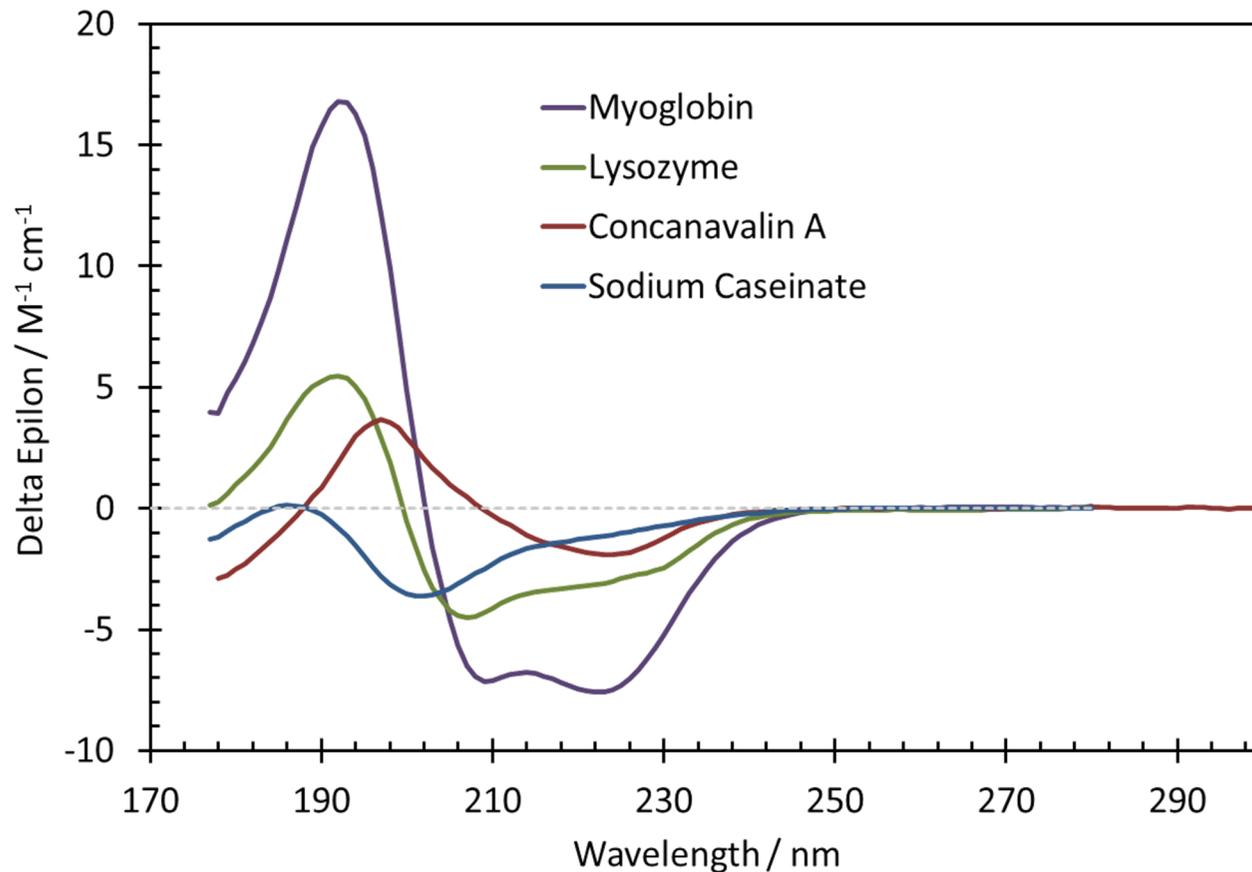
Getting acquainted with protein CD spectra

Sodium Caseinate *Intrinsically disordered protein (IDP)*



Getting acquainted with protein CD spectra

All spectra



Note the difference in

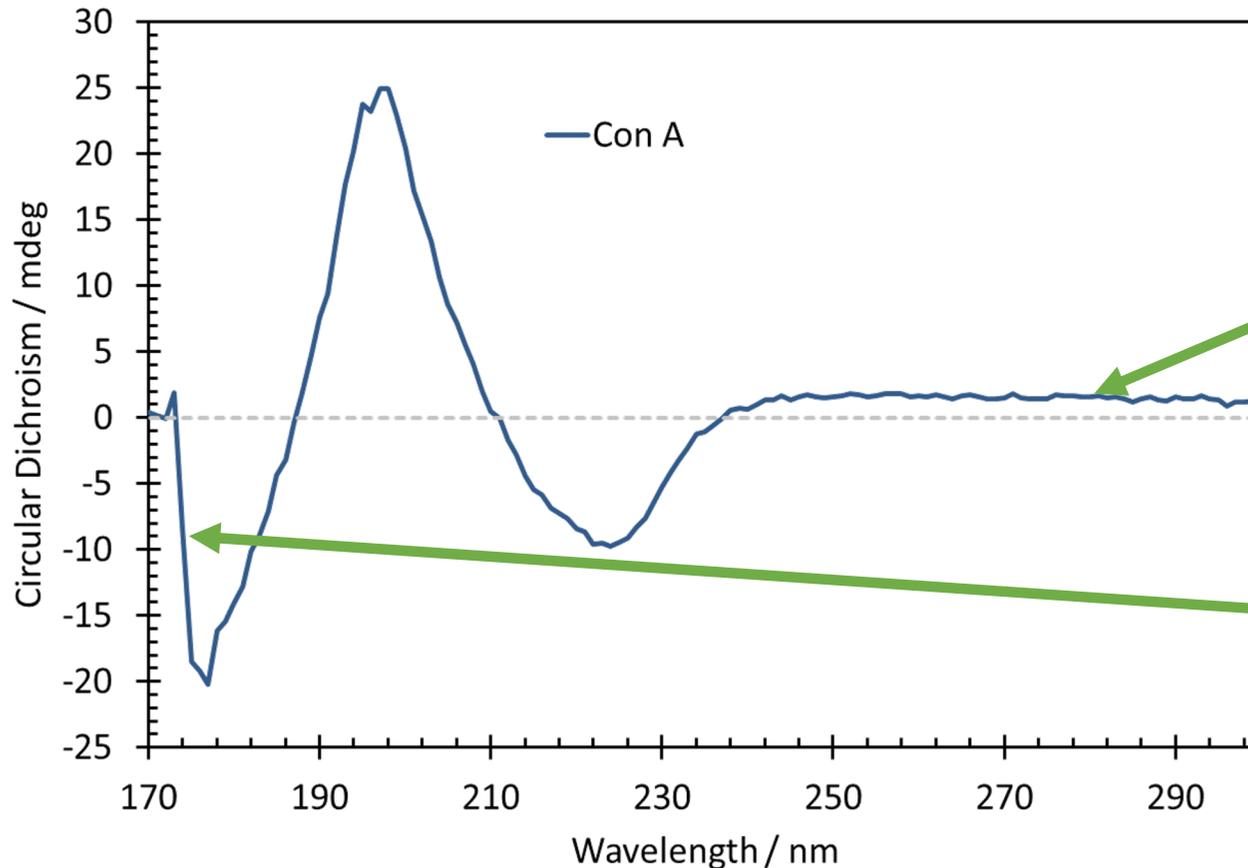
- Signal magnitude
- Peak position
- Zero signal crossings

All this is why we can differentiate between different secondary structure components



Actually acquiring data

Sample scan



Acquiring a single scan of your sample

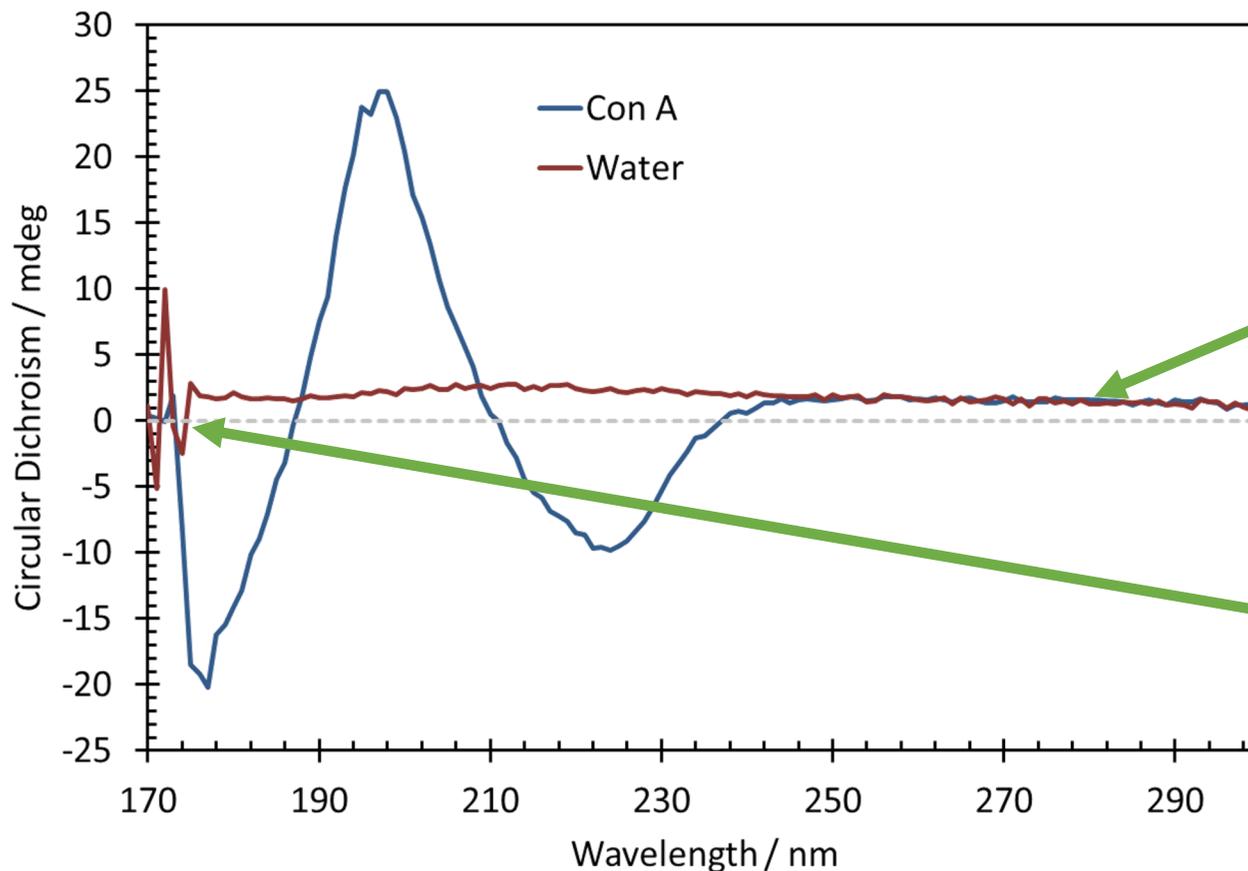
Non Zero signal at long wavelength !

Abrupt change at low wavelength, tending to zero !



Actually acquiring data

Sample and buffer scan



Acquiring a single scan of your sample and buffer

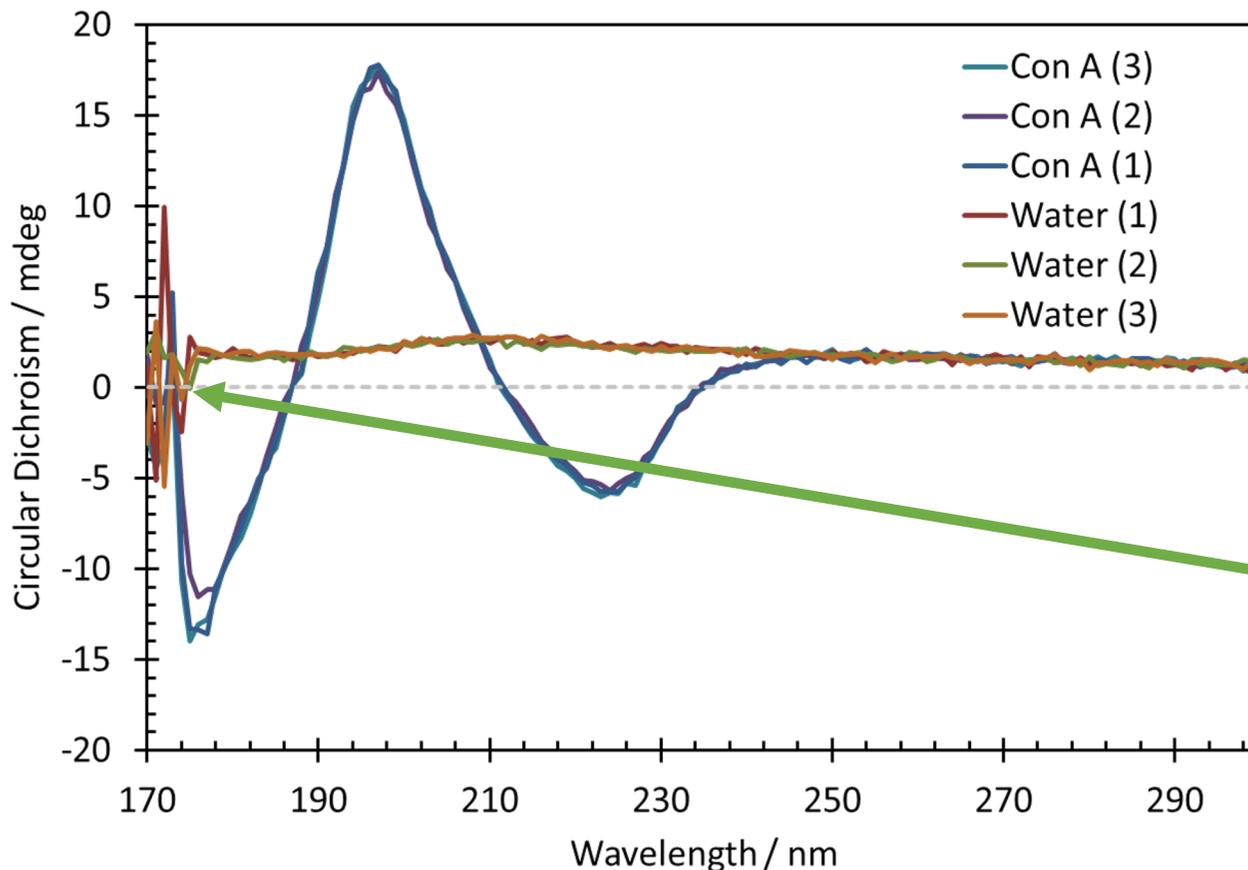
Buffer matches at long wavelength

Abrupt change at low wavelength, tending to zero and noisy !



Actually acquiring data

Multiple sample and buffer scans



Acquiring multiple scans of your sample and buffer

All spectra match within the noise

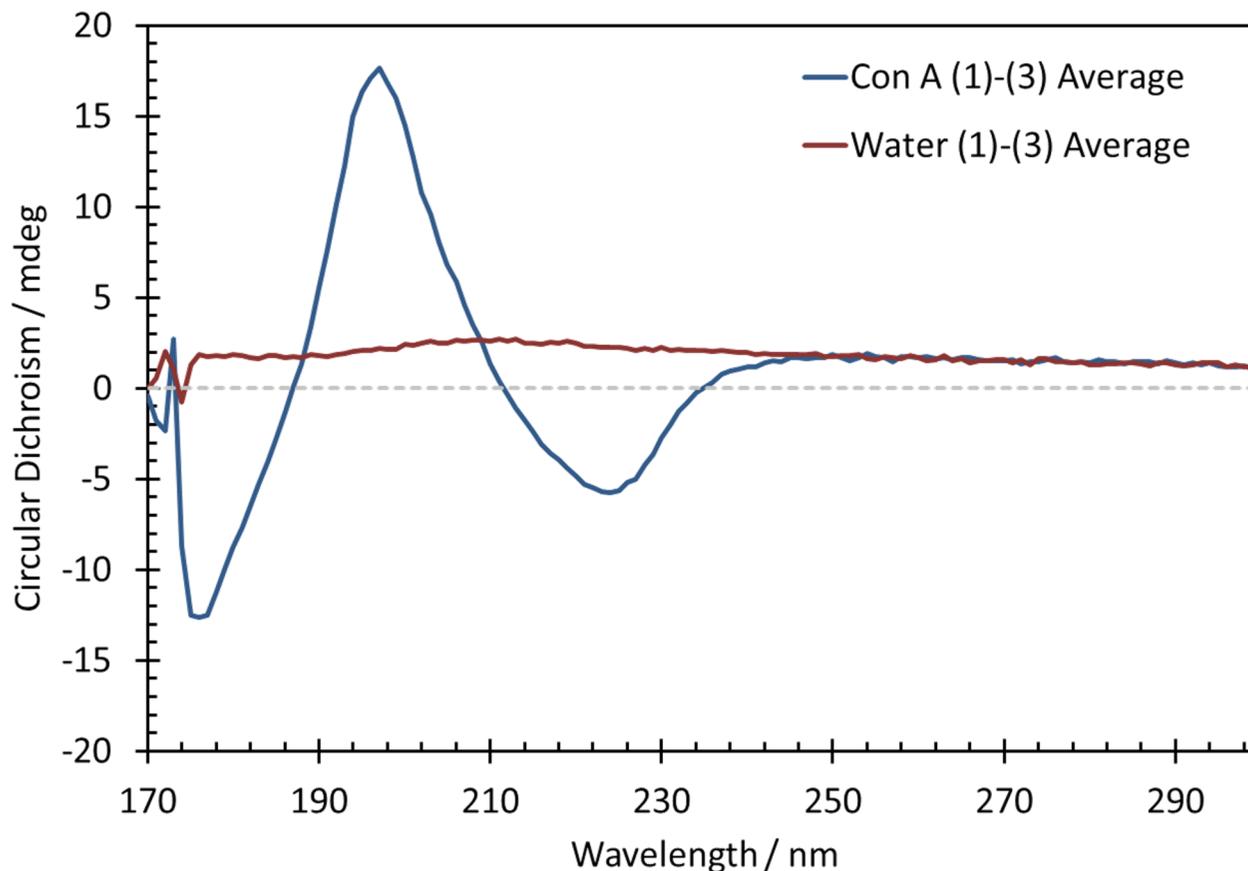
Why the noise at low wavelength ?

Buffer and sample absorption high



Actually acquiring data

Multiple sample and buffer scans - Average scans



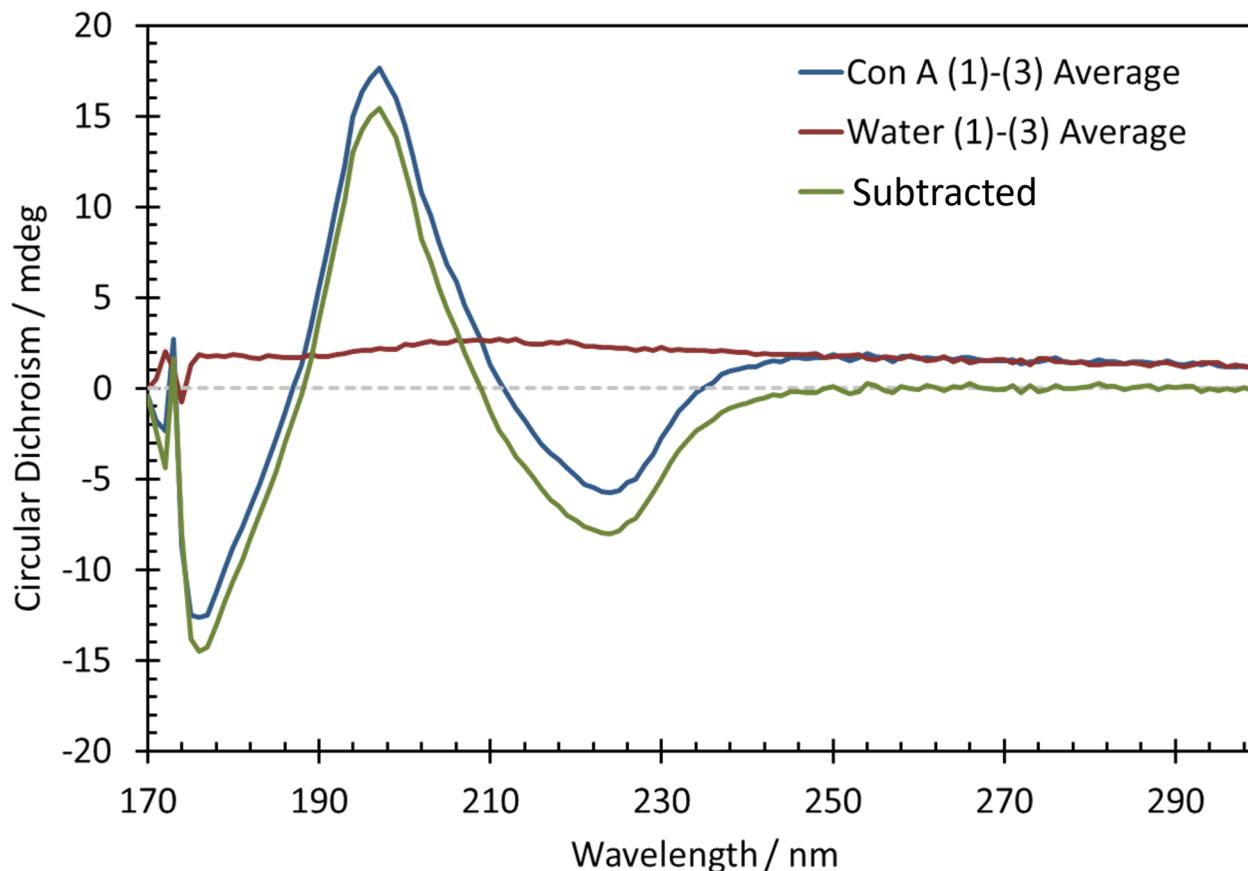
Acquiring multiple scans of your sample and buffer

Average sample and buffer scans



Actually acquiring data

Multiple sample and buffer scans - Average scans - Subtract



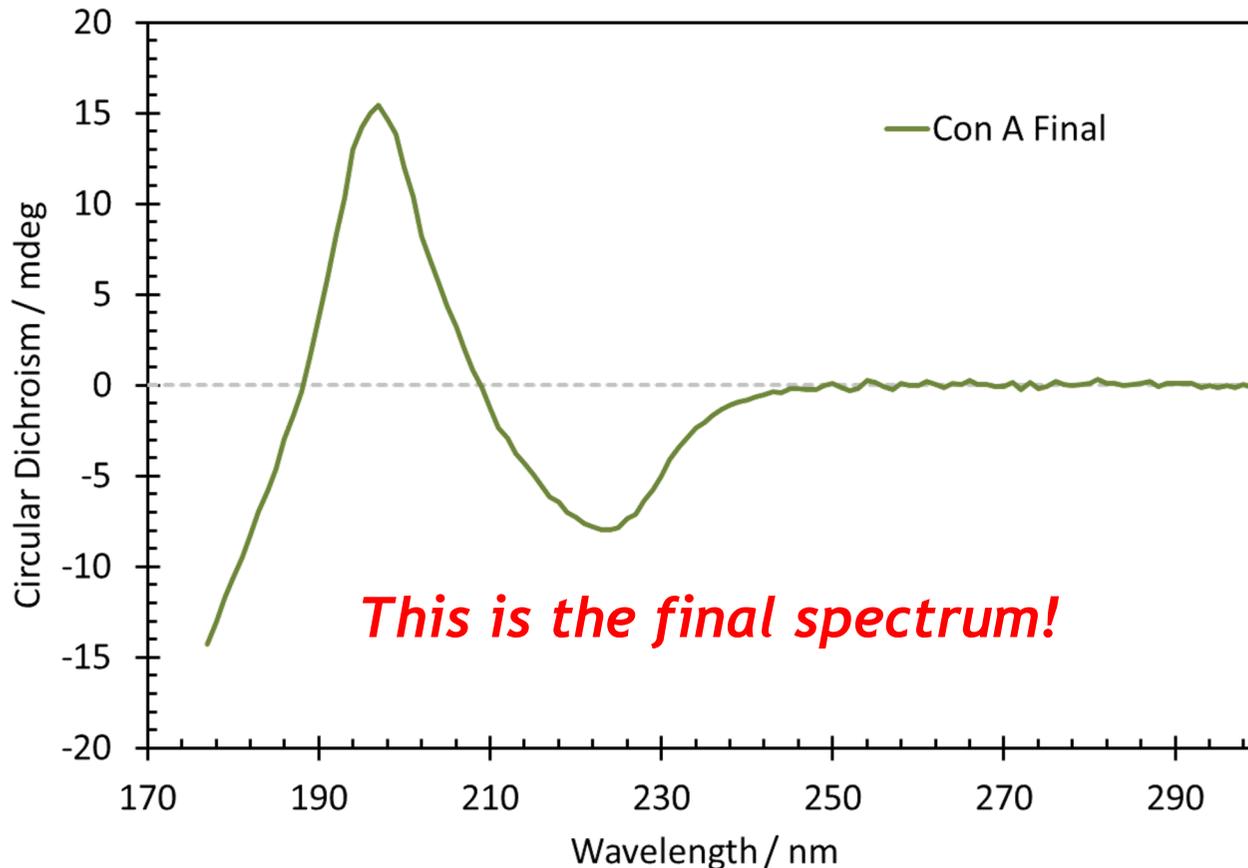
Acquiring multiple scans of your sample and buffer

Subtract the average buffer scan from the average sample scan



Actually acquiring data

Multiple sample and buffer scans - Average scans - Subtract



Acquiring multiple scans of your sample and buffer

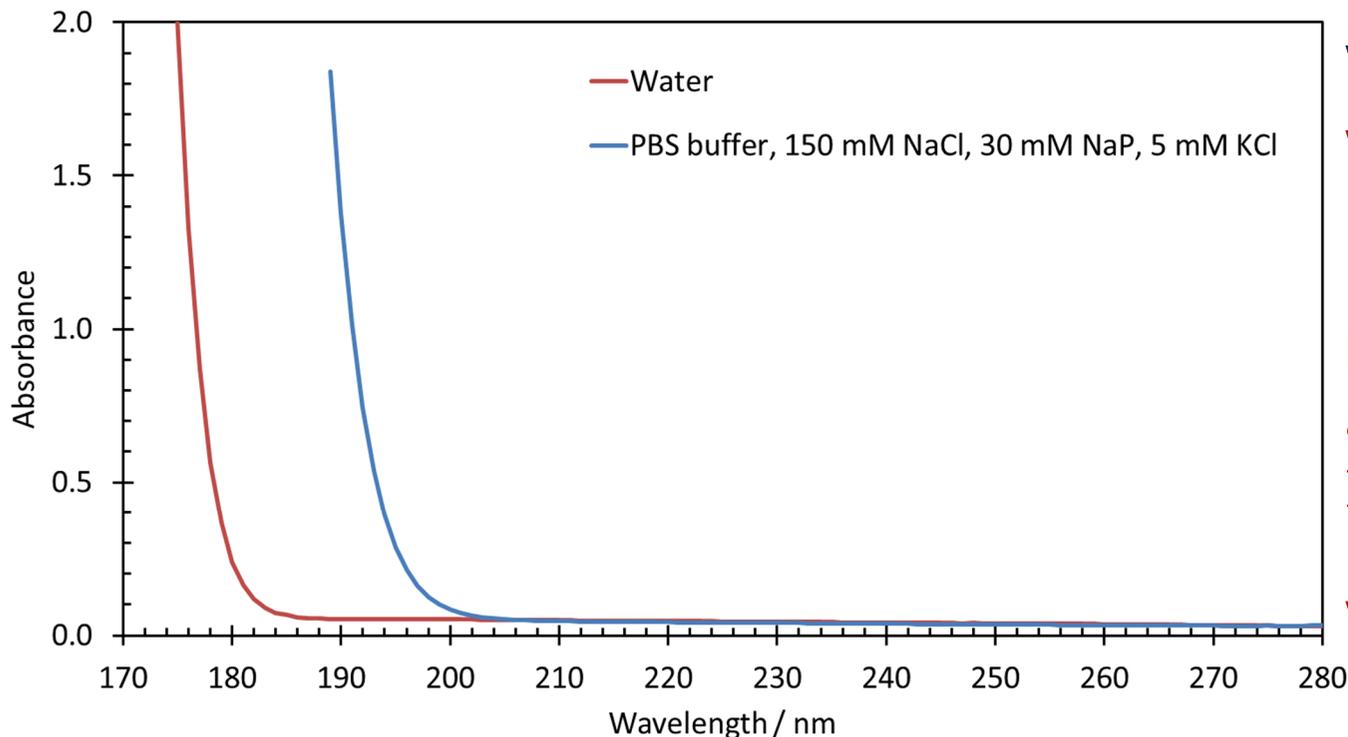
Remove the untrustworthy low wavelength data points

... and you are done



Sample buffers

What is the best choice for buffer for *far-UV* CD



Water is the reference

We like to use phosphate buffers

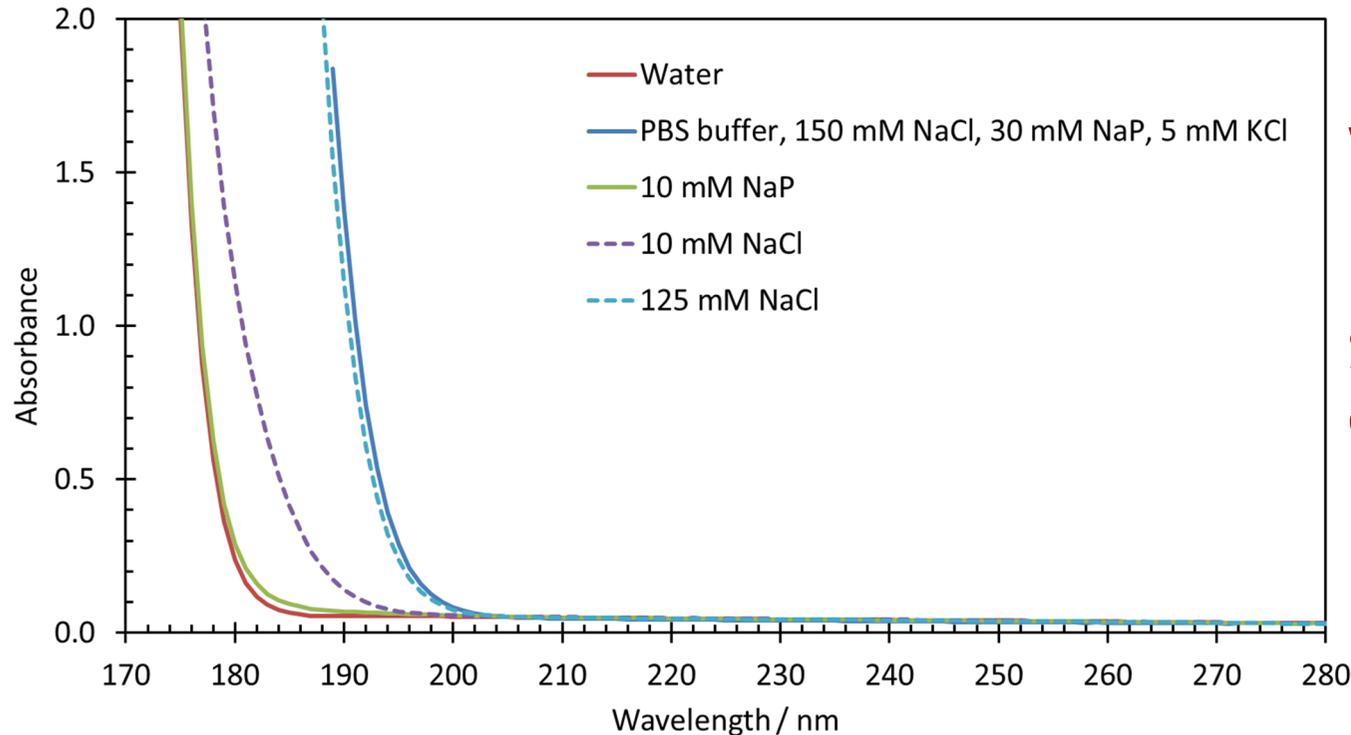
But PBS has a high absorbance below 200 nm

Why...?



Sample buffers

What is the best choice for buffer for *far-UV* CD



We like to use phosphate buffers

Sodium Phosphate only absorbs a little.

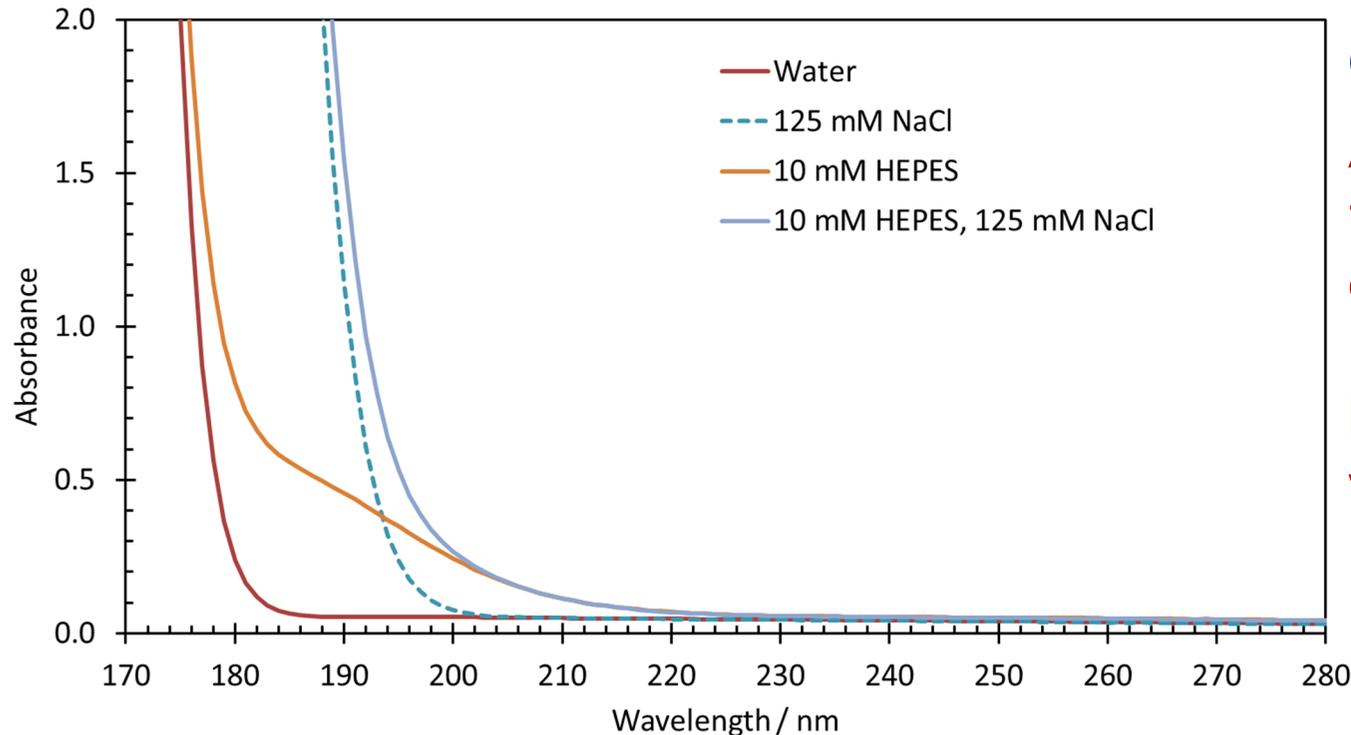
➤ *NaCl is the issue*

Try to avoid anything with Cl^- ions. E.g. use NaF instead of NaCl



Sample buffers

What is the best choice for buffer for *far-UV* CD



Other buffers?

As an example, HEPES is not very good either

Especially when used with NaCl

More tomorrow: Hands on “*Buffer limitations study*”

