# **Protein Unfolding**

# **Experimental Guidelines**

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## How to quantitate protein stability? $T_m$ ? $[D]_m$ ?

# Stability $\triangle G$ Difference in Gibbs energy between the native state and the unfolded state

We might determine equilibrium constants by measuring equilibrium concentrations ( $\Delta G = -RT \ln K_{eq} = -RT \ln[U]/[N]$ ) But, populations of conformational states are not comparable (e.g., what if  $\Delta G = 5$  kcal/mol)

We must "stress" the protein in order to populate other less stable conformational states...

Temperature or denaturant trigger evolution through a series of equilibrium states, and, later on, final extrapolation will provide the sought information





Energy required to destabilize the molecular tridimensional structure



**Experimental techniques:** 

- Spectroscopy (UV, F, CD, NMR, FTIR)
- Calorimetry (DSC)
- Property with different values for different states
- Signal proportional to the unfolding progress
- Local or global information













- Information on inter- and intramolecular interactions involved in (un)folding
- Information on functional groups involved in (un)folding
- Information on physiological and pathological mechanisms

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Setting experiment parameters

- Experimental technique?
- Solution composition
- Sample preparation
- Sample concentration
- Scanning rate



# Experimental conditions

- Physiological conditions?
- Thermodynamic stability
- Kinetic stability
- Aggregation/precipitation
- Physiological/technological relevance



# Control experiments

- Experimental condition scanning pH, ionic strength, co-solutes...
- Water/buffer test (clean, contaminants?)
- Reversibility test (equilibrium?)
- Scanning rate test (equilibrium?)
- Concentration test (association, aggregation?)



### Unfolding reversibility test



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- Standard scanning rate: 1°C/min
- Potential kinetic coupling with additional processes



## Protein concentration



- Equilibrium equations are more complex (unfolding coupled to association/dissociation)
- Caveats: protein availability, solubility and/or aggregation
- Test for self-association or aggregation in the native/unfolded state



#### Presence of co-solutes and co-solvents



DMSO for solubilizing hydrophobic compounds Glycerol, salts for solubilizing proteins



# There are Good's buffers, but no bad buffers



A buffer with pKa  $\simeq$  pH and low ionization enthalpy is preferable



# Additional information + Ockham's Razor



- Consider known information
- Revise thermogram
- Look for unusual features (?)
- Start with simplest compatible model
- Increase model complexity until needed
- Assumptions/constraints of model
- Values of the estimated parameters



# Stability & folding cooperativity



Rodriguez-Cardenas et al. *PLoS ONE* 2016 **11** e0161020 Cremades et al. *Biochemistry* 2008 **47** 627-639 Irun et al. *Journal of Molecular Biology* 2001 **306** 877-888





Stability & folding cooperativity

Structural similarity does not result in stability or folding cooperativity

- Similar proteins may exhibit different structural stability
- Similar proteins may exhibit different conformational landscape and (un)folding cooperativity







Freire. Pure & Applied Chemistry1997 **69** 2253-2261 Robertson & Murphy. Chemical Reviews 1997 **97** 1251-1268 Xie & Freire. Proteins 1994 **19** 291-301 Murphy & Freire. Advances in Protein Chemistry 1992 **43** 313-361





# No kinetic information No irreversible processes





#### Irreversible unfolding & kinetic stability coupling







# Stability: kinetic study



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# Advanced experimental design

- Several thermally-induced transitions
- Oligomer unfolding assay
- Ligand-induced stabilization (TSA)
- Lattice-like molecule assay
- Membrane transition assay
- Kinetically coupled transitions
- Irreversible transitions
- Association/dissociation coupling
- Absolute heat capacity determination

...

# Protein stability depends on:

- Sequence (net charge, charge distribution, hydropathy profile, mutations, natural variability...)
- Post-translational modifications (deletion, phosphorylation, hydroxylation, acetylation, methylation...)
- External conditions (T, P, pH, μ...) and excipients/solutes
- Chemical modifications (deamidation, oxidation, proteolysis and hydrolysis, β-elimination, racemization...)
- Interacting molecules (ligands, osmolytes...)
- Presence of surfaces and interfaces/interphases



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